

Simulations of Solution Spectra using SimFonia

SimFonia Features



- Fast, easy-to-use Simulation Program. Runs under Microsoft Windows.
- The simulation algorithm is based upon the perturbation theory
- Choice of solution or powder spectra
- Allows for elementary data processing
- Easy transfer of simulated spectra to WIN-EPR for further post-processing
- Efficient (FT) Fourier Transform algorithm for multiline spectra
- Simulation of the m_i dependent linewidth by a polynomial approximation
- Up to 20 inequivalent nuclei with a large number of equivalent nuclei
- The Powder simulation program simulate spectra for electron spin 1/2 to spin 7/2. For spins greater than 1/2, D and E zero-field splitting terms are implemented.

Starting a simulation, Instrument parameters

WINEPR SimFonia - [Sim4]

File Parameter Operation Processing View Options Window Info

Solution [I] [H] [S] [RUN] [STOP]

10³] 2.50_

Instrument Parameter

Operator: [] Date: 07/22/2022 Time: 18:07

Comment: []

Automatic field detection

Center Field [3480.00] [G]

Sweep Width [50.00] [G]

Resolution in X [1024]

Signal Channel

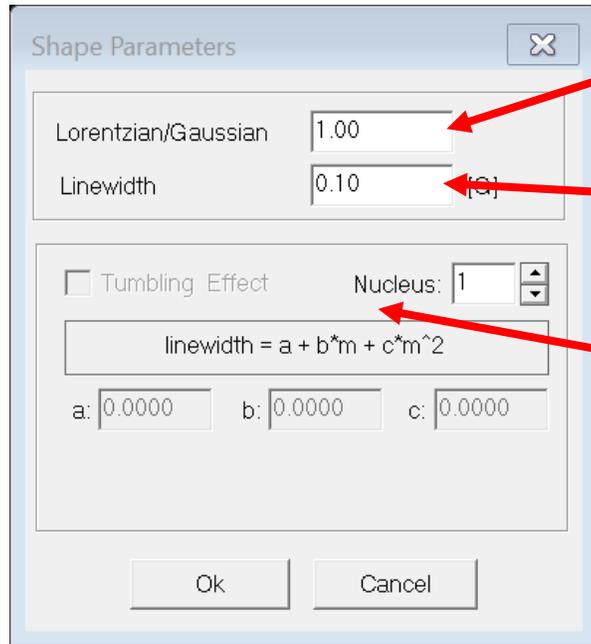
Modulation Amplitude [0.10] [G]

Time Constant [1.25] [msec]

Conversion Time [5.12] [msec]

MW Frequency [9.7500] [GHz] Harmonic [1]

Starting a simulation, Shape parameters

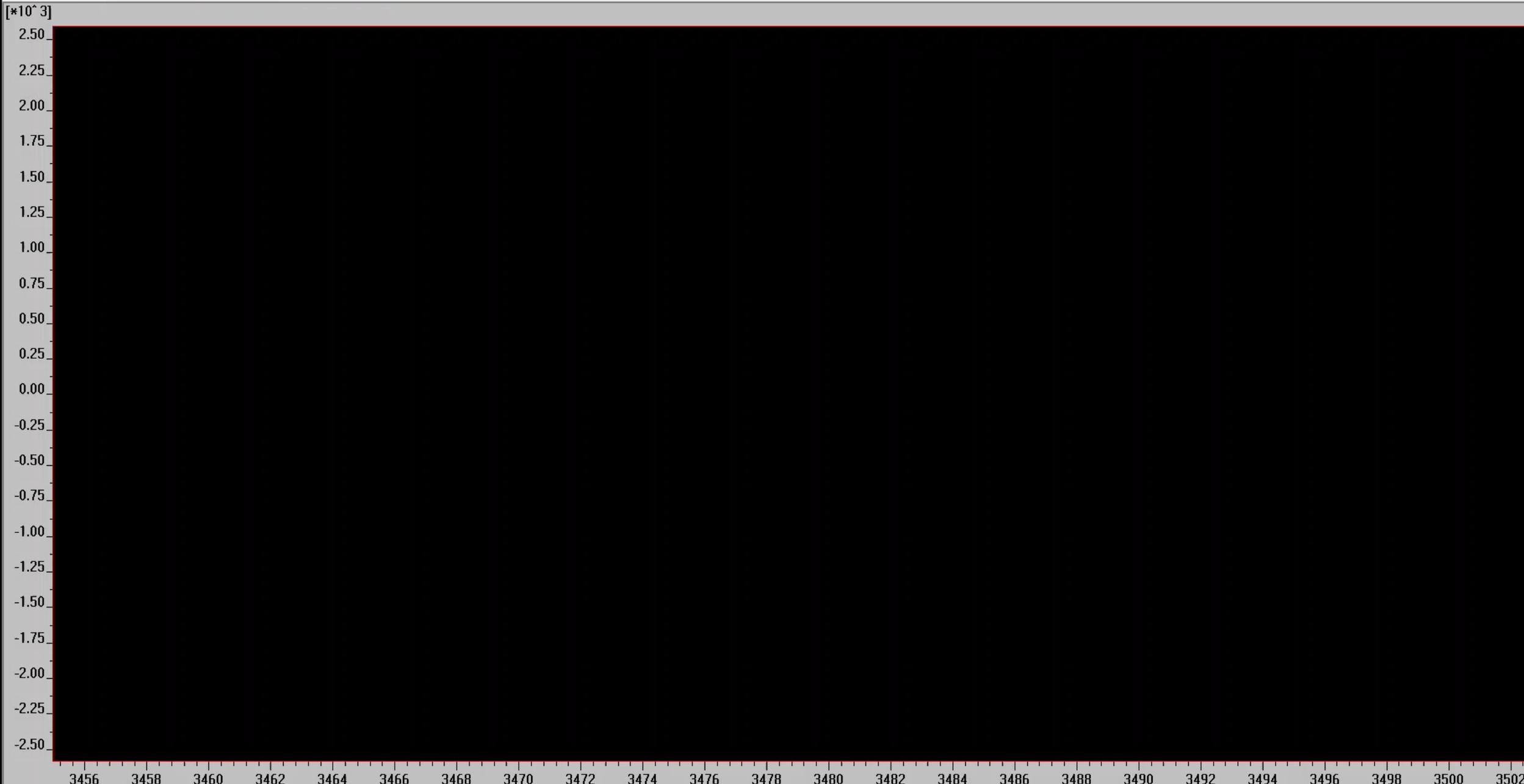


Line Shape Parameter: 0 =Lorentzian, 1 – Gaussian

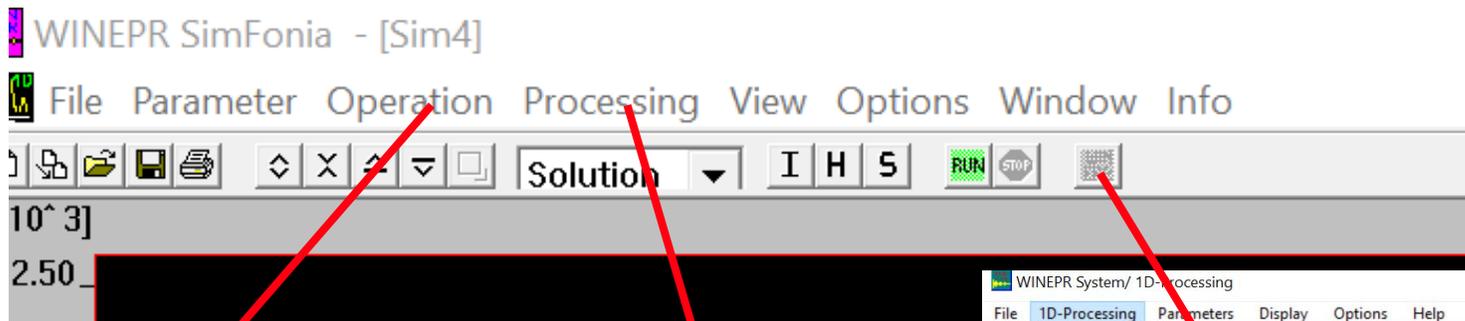
Any value α in between is a Voigt line shape with a fraction of Gaussian equal α and fraction of Lorentzian equal $(1 - \alpha)$

Line width

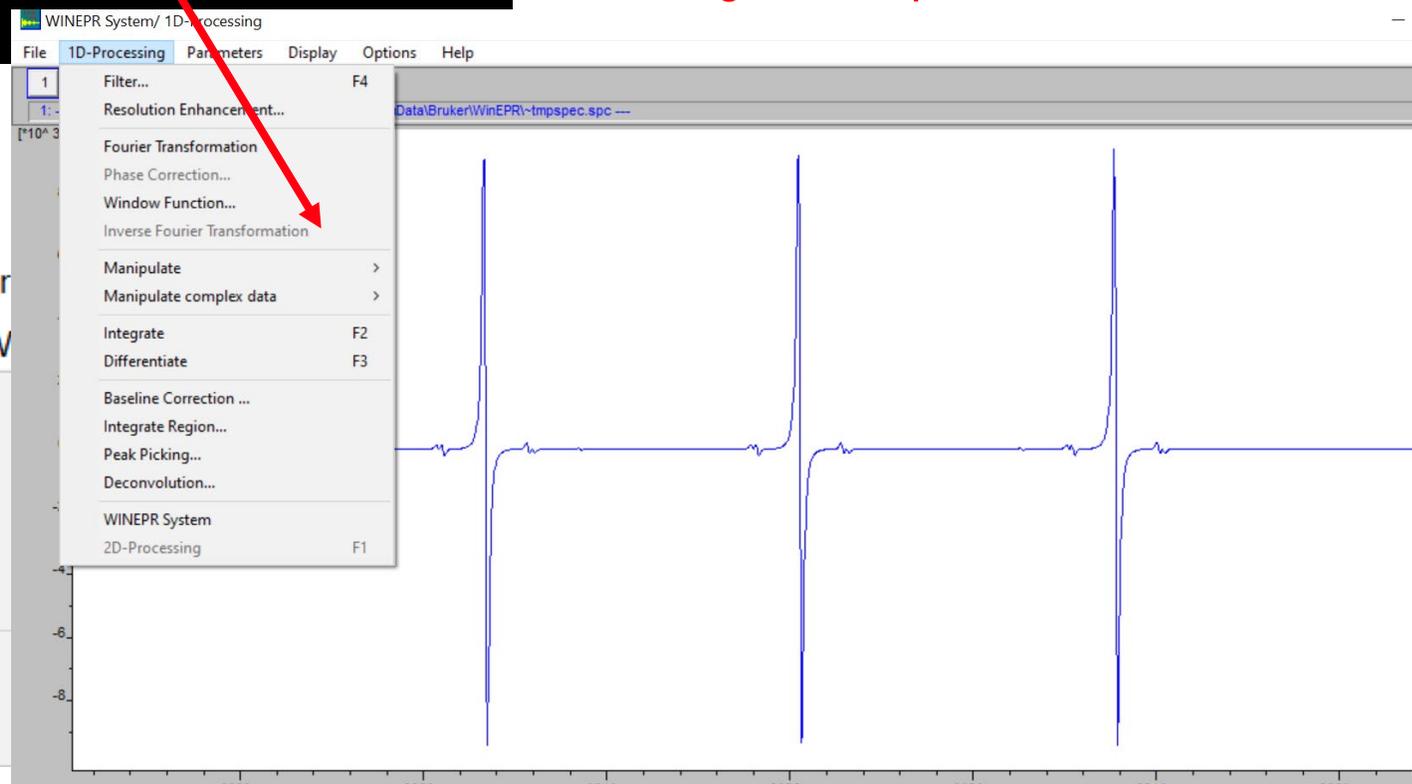
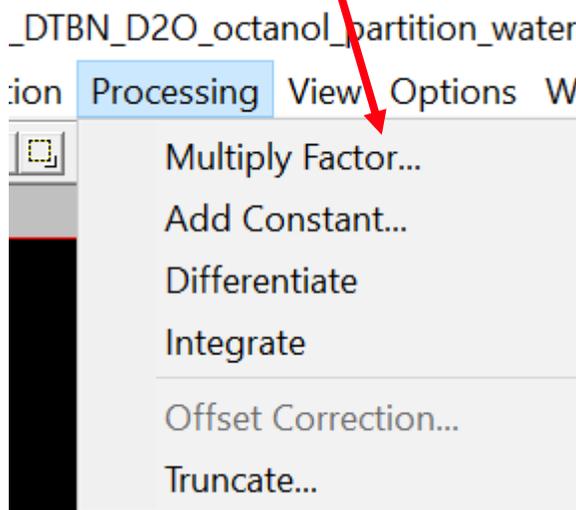
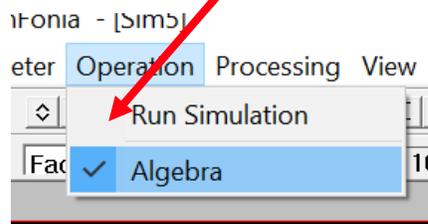
This part, the line width variation according to the Kivelson theory works only if (1) a single isotope is listed in Hamiltonian parameters and (2) the **third order** box is checked



Further processing

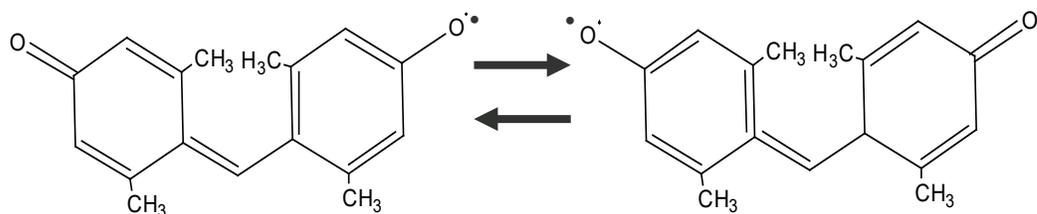


Sending to WinEpr



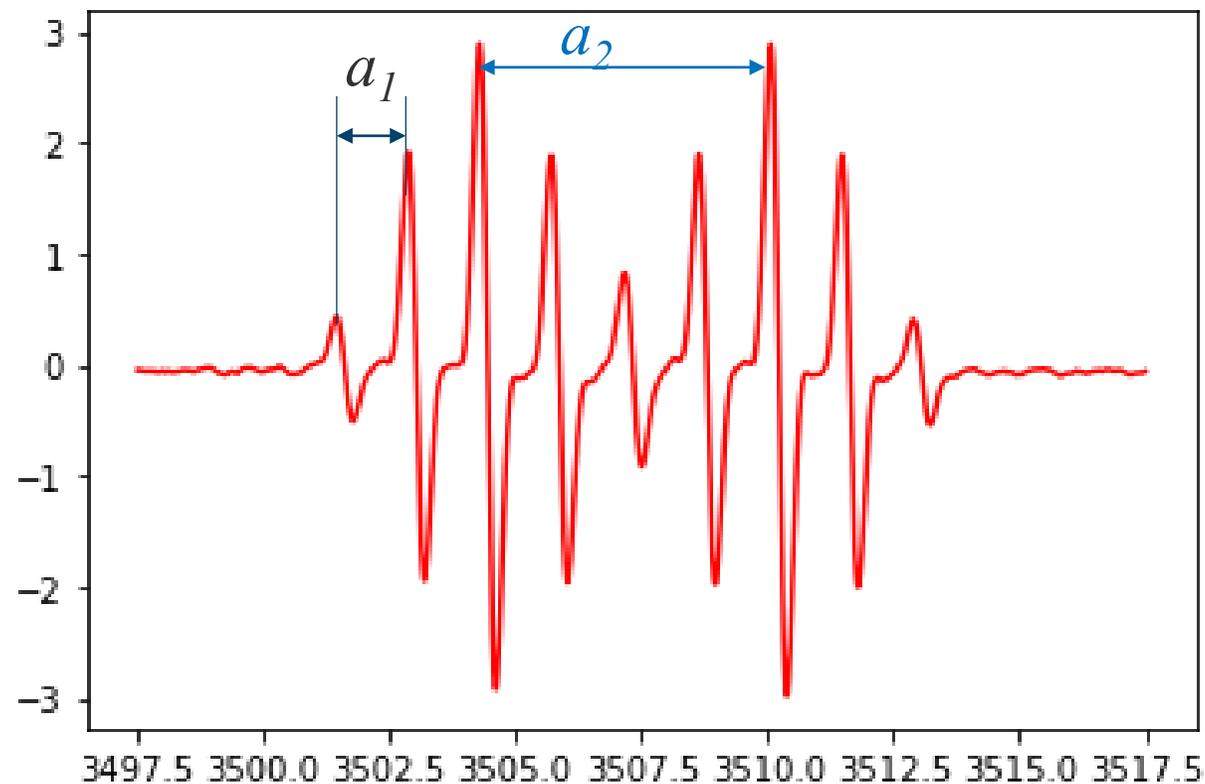
Simulating an experimental spectrum. Galvinoxyl

A galvinoxyl radical



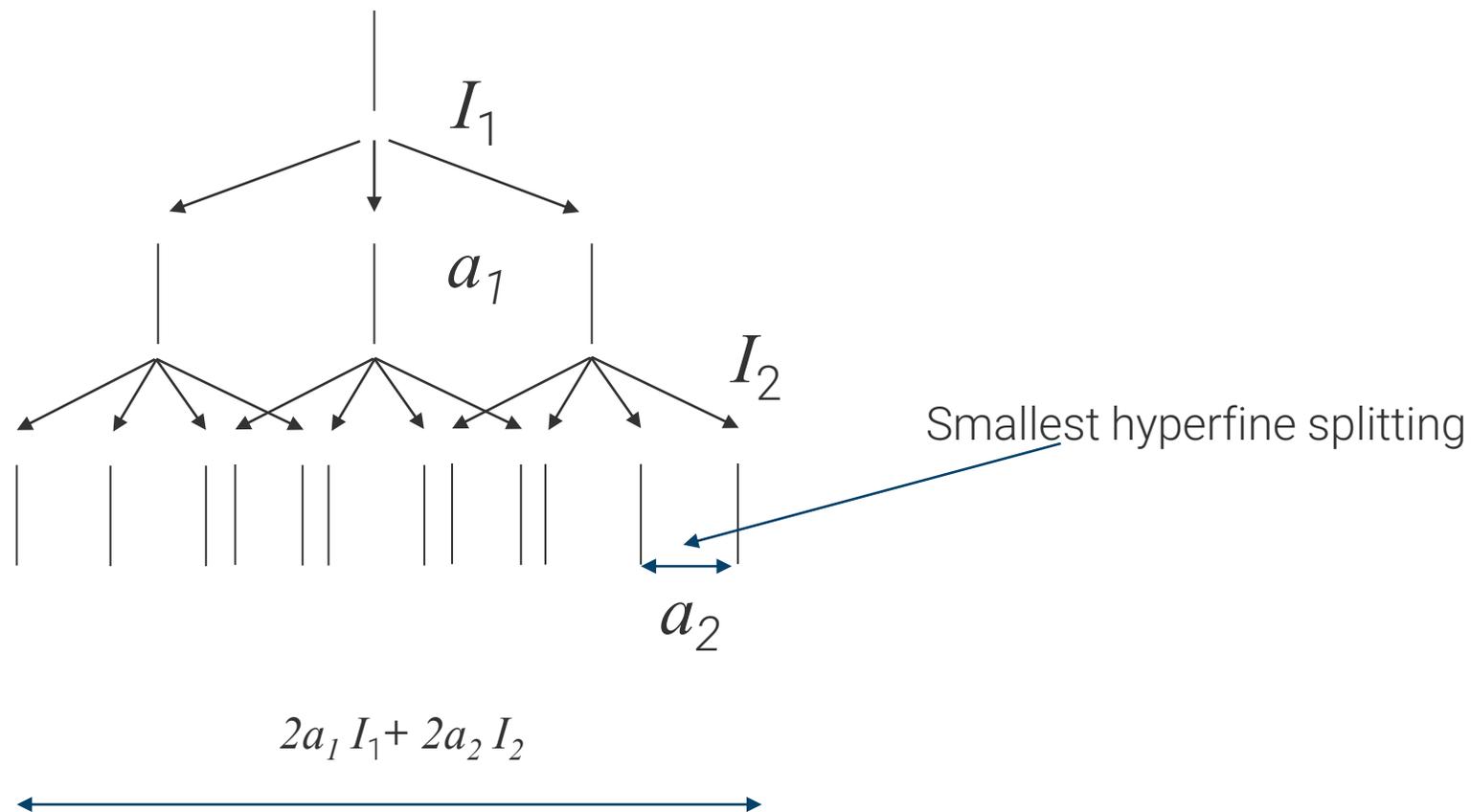
4 protons in the quinone/phenol rings
1 methine proton

12 methyl protons do not show resolved splitting, but cause line broadening



$$a_2 \approx 4 \times a_1$$

Assignment of hyperfine splitting values



Simulating an experimental spectrum. Getting instrumental parameters from experiment



Instrument Parameter

Operator: Boris Date: 05/19 Time: 17:42

Comment: 'PDT_CS2_Rt'

Automatic field detection

Center Field: 3315.00 [G]

Sweep Width: 39.99 [G]

Resolution in X: 2048

Signal Channel

Modulation Amplitude: 0.10 [G]

Time Constant: 1.25 [msec]

Conversion Time: 40.96 [msec]

MW Frequency: 9.3123 [GHz]

Harmonic: 1

Simulating an experimental spectrum. Getting numbers from experiment



Hamiltonian Parameters

Nucleus

Element	#Nuclei	Isotope	Spin	Iso.Abd.	g-Factor	A [G]	
N	1	14	1	99.64	0.403761		Exp
		15	1/2	0.36	-0.566378		Exp
		0					Exp
		0					Exp

Clear

Get Element Insert Delete

Electron

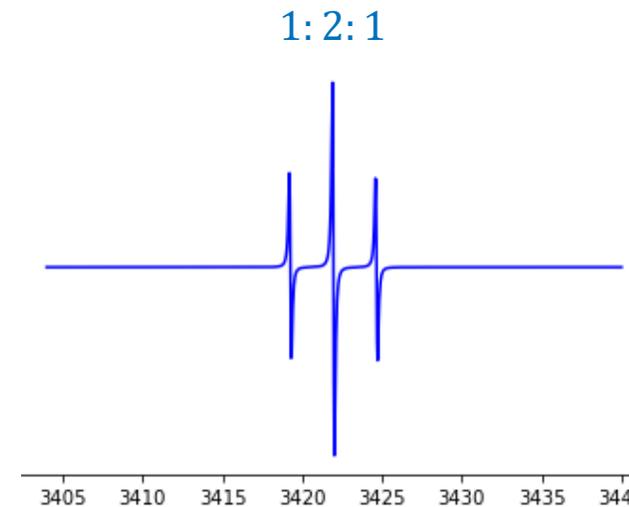
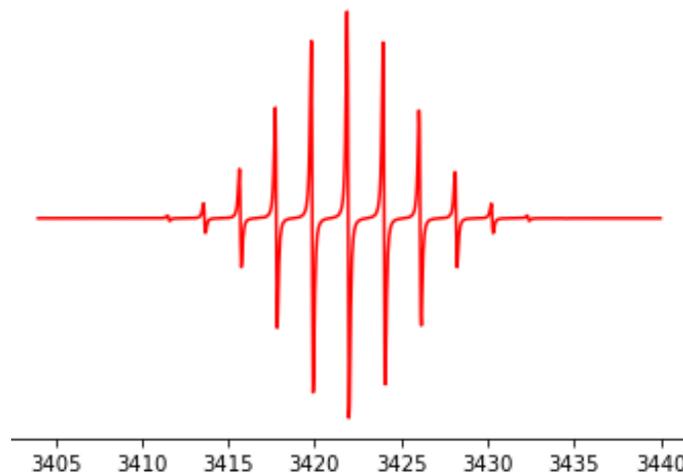
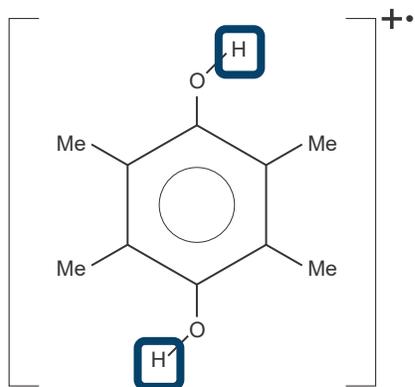
g-Factor Exp Third Order Ok Cancel



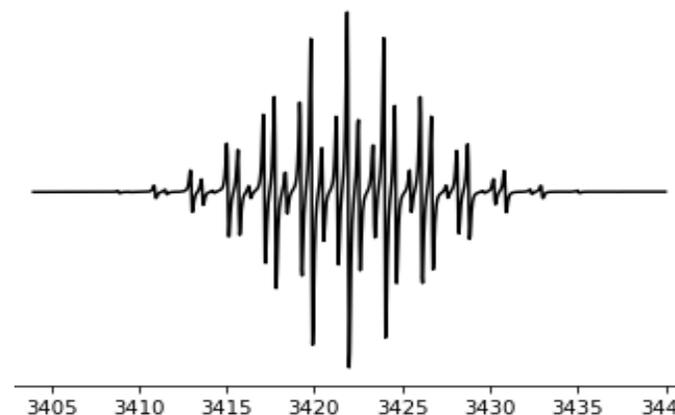


Simulating an experimental spectrum. Duro-semiquinone anion

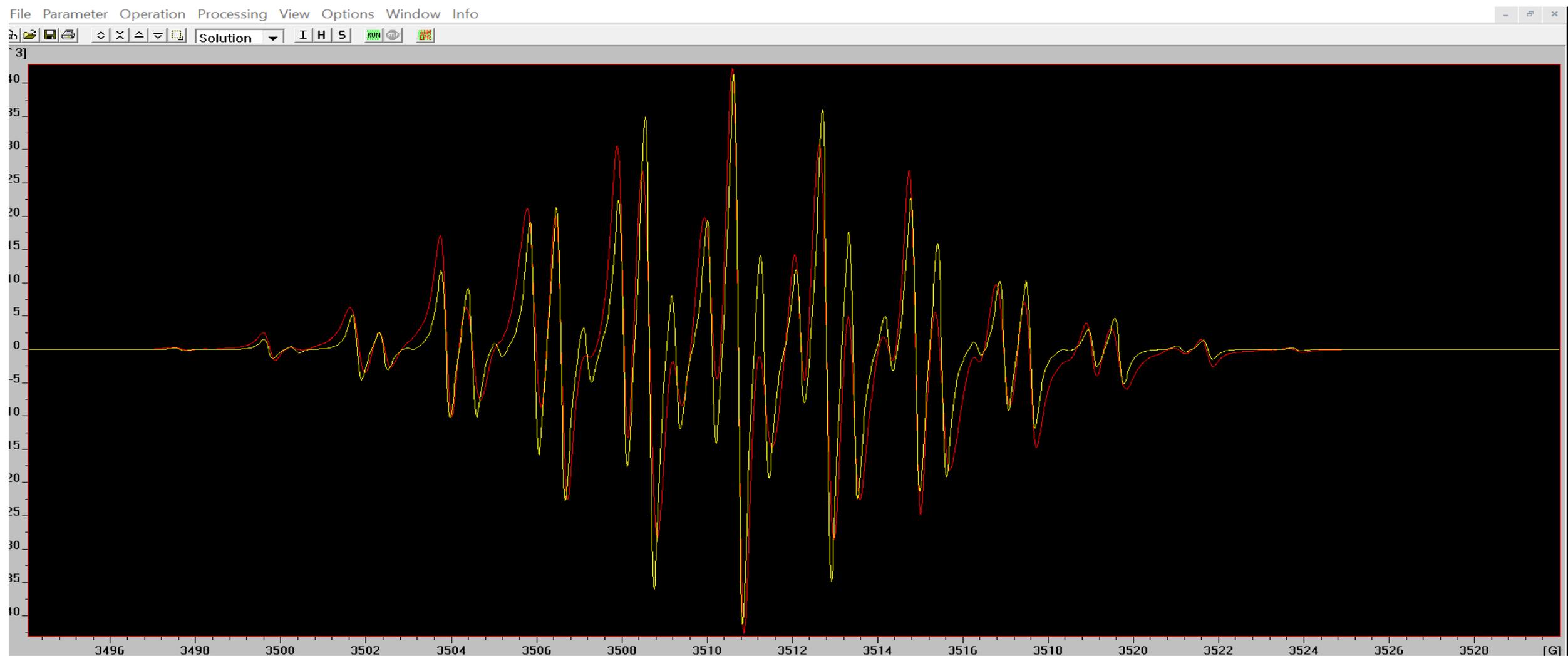
1: 12: 66: 220: 495: 792: 924: 792: 495: 220: 66: 12: 1



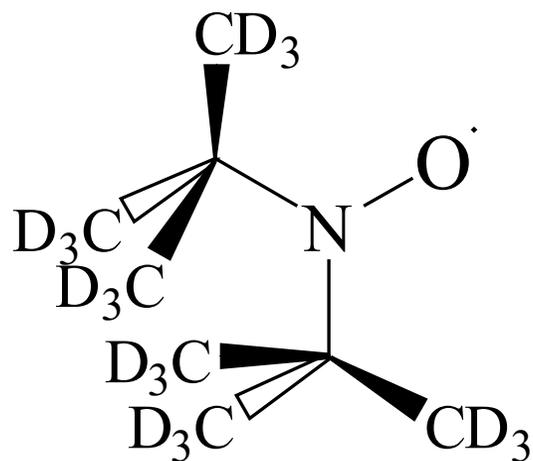
$$\left(2 \cdot 12 \cdot \frac{1}{2} + 1\right) \cdot \left(2 \cdot 2 \cdot \frac{1}{2} + 1\right) = 39 \text{ lines}$$



Duro-semiquinone anion – alternating linewidth



Simulating an experimental spectrum. Di-*tert*-butyl nitroxide



Three main lines from HF splitting on ^{14}N . $I=1$.

Satellite lines:

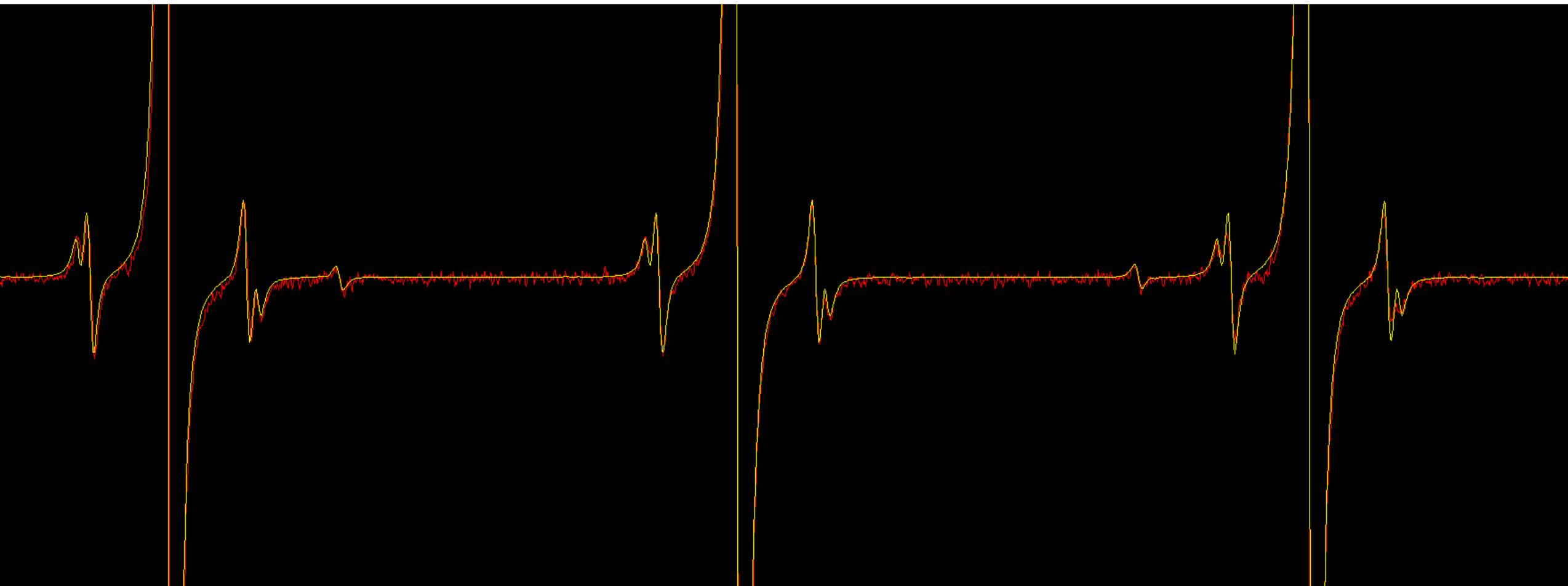
^{15}N – natural abundance 0.37%. $I=1/2$.

^{13}C – natural abundance 1.07%. $I=1/2$. Eight atoms.

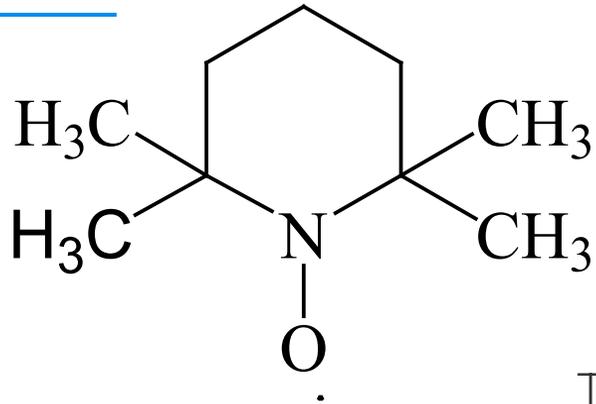
^{17}O - natural abundance 0.038%. $I=5/2$. Can be neglected.

[*10⁻³]

Simulating an experimental spectrum. Di-*tert*-butyl nitroxide



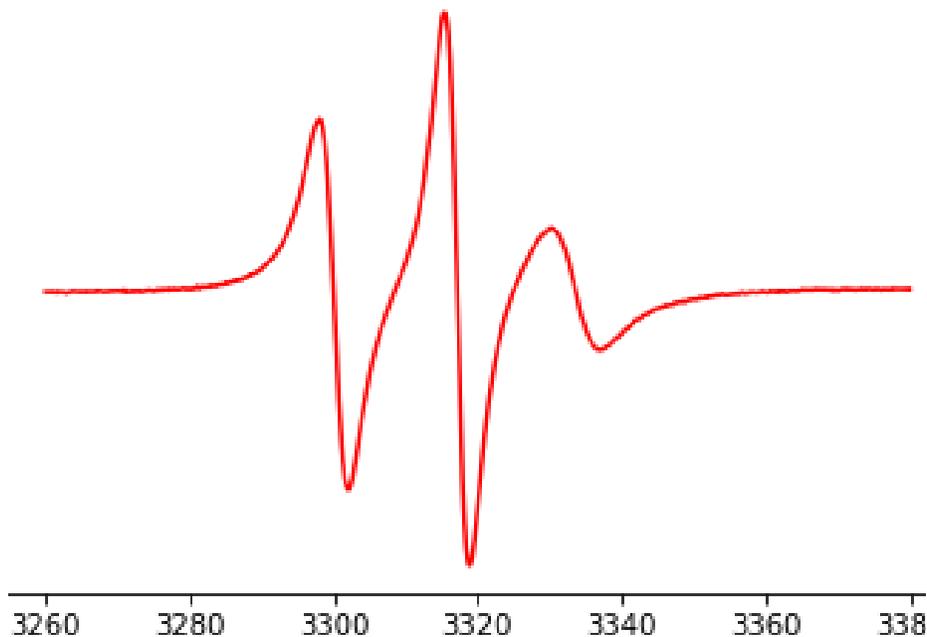
Simulating an experimental spectrum. Tempo in water/glycerol at 230K



$$\tau = \frac{\eta V}{kT} \quad \text{Stocks-Einstein formula}$$

τ is the *rotation correlation time*, the time to rotate one radian
 η is the microviscosity

The lineshape is due to an incomplete motional averaging



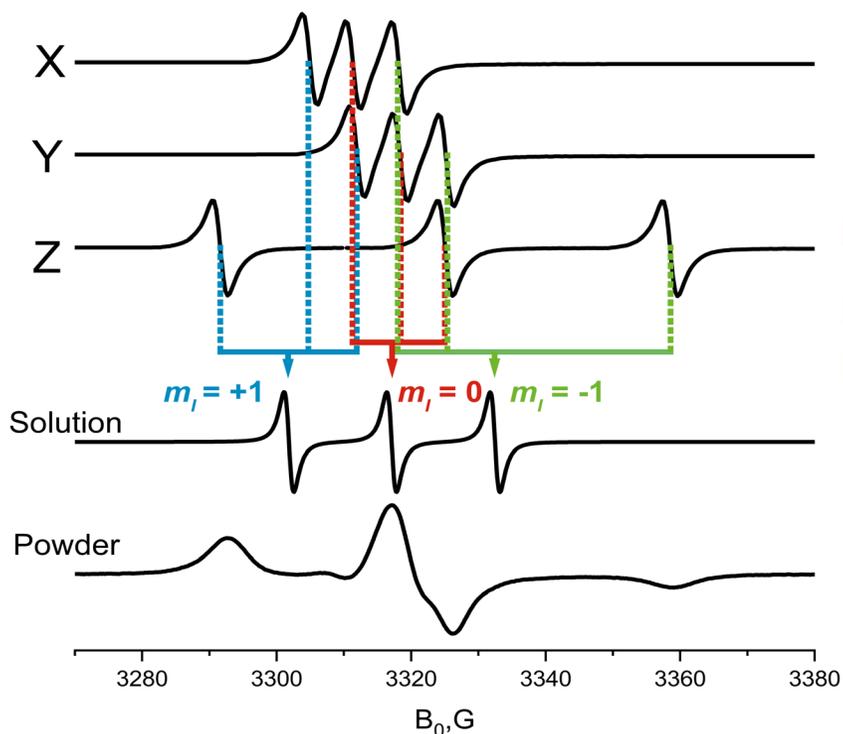
$$\text{Linewidth} = A + Bm_I + Cm_I^2$$

According to the *Kivelson theory*

Tempo in water/glycerol at 230K. Incomplete motional averaging

In the powder spectrum a position of each line for individual θ, φ orientation is given:

$$B_{res}(\theta, \varphi, m) = \frac{h\nu}{g(\theta, \varphi)\beta} - mA(\theta, \varphi)$$



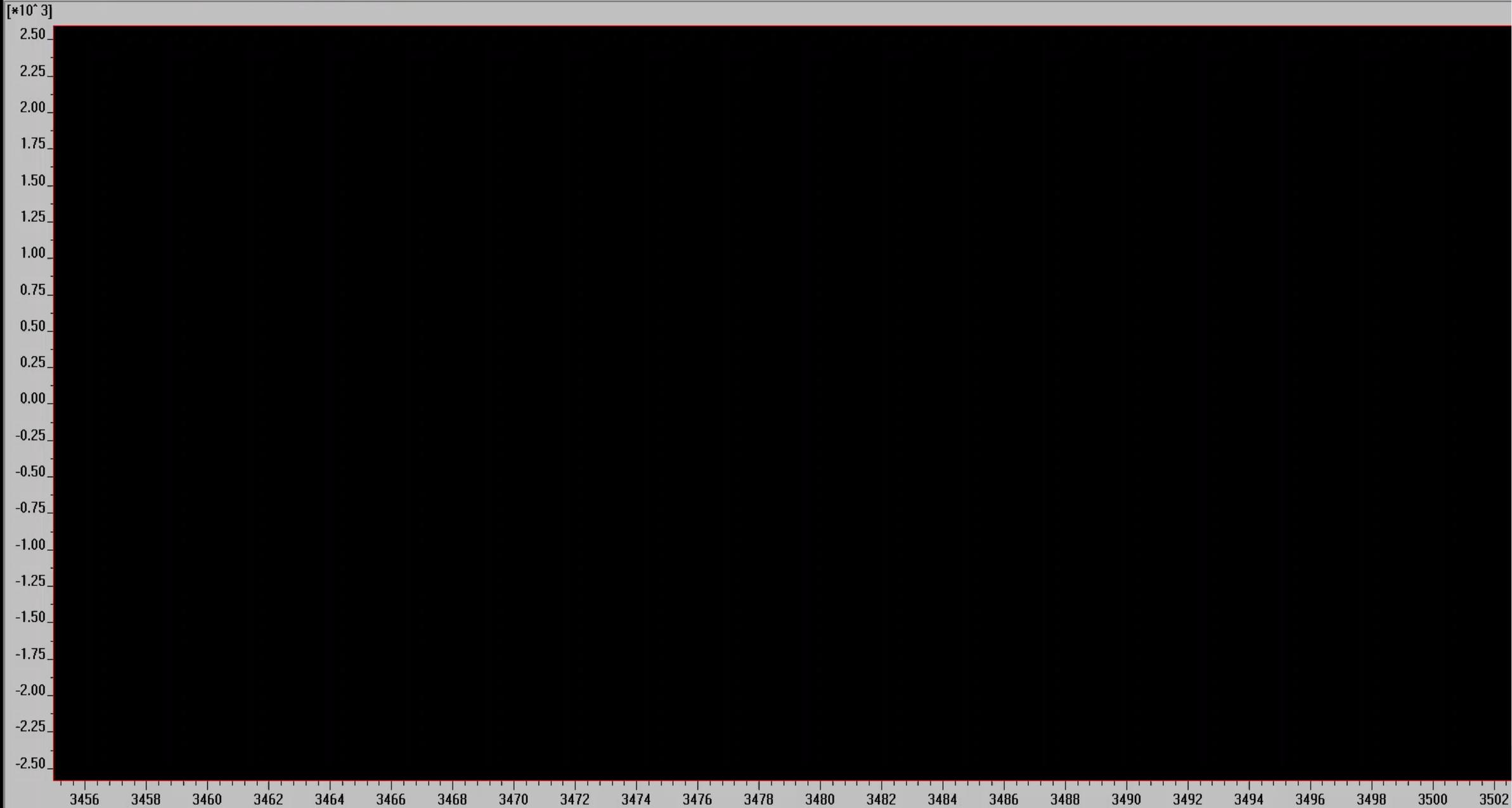
m=0 the line is spread over the effect of Δg anisotropy: $\frac{h\nu}{g_z\beta} - \frac{h\nu}{g_x\beta} \cong \frac{h\nu\Delta g}{4\beta}$
m=+1 the line is spread over the effect of $\Delta A - \Delta g$ anisotropy
m=-1 the line is spread over the effect of $\Delta A + \Delta g$ anisotropy

$g_x=2.0091, g_y=2.0061, g_z=2.0023,$
 $A_x= 6.2, A_y = 6.3, A_z=33.6$

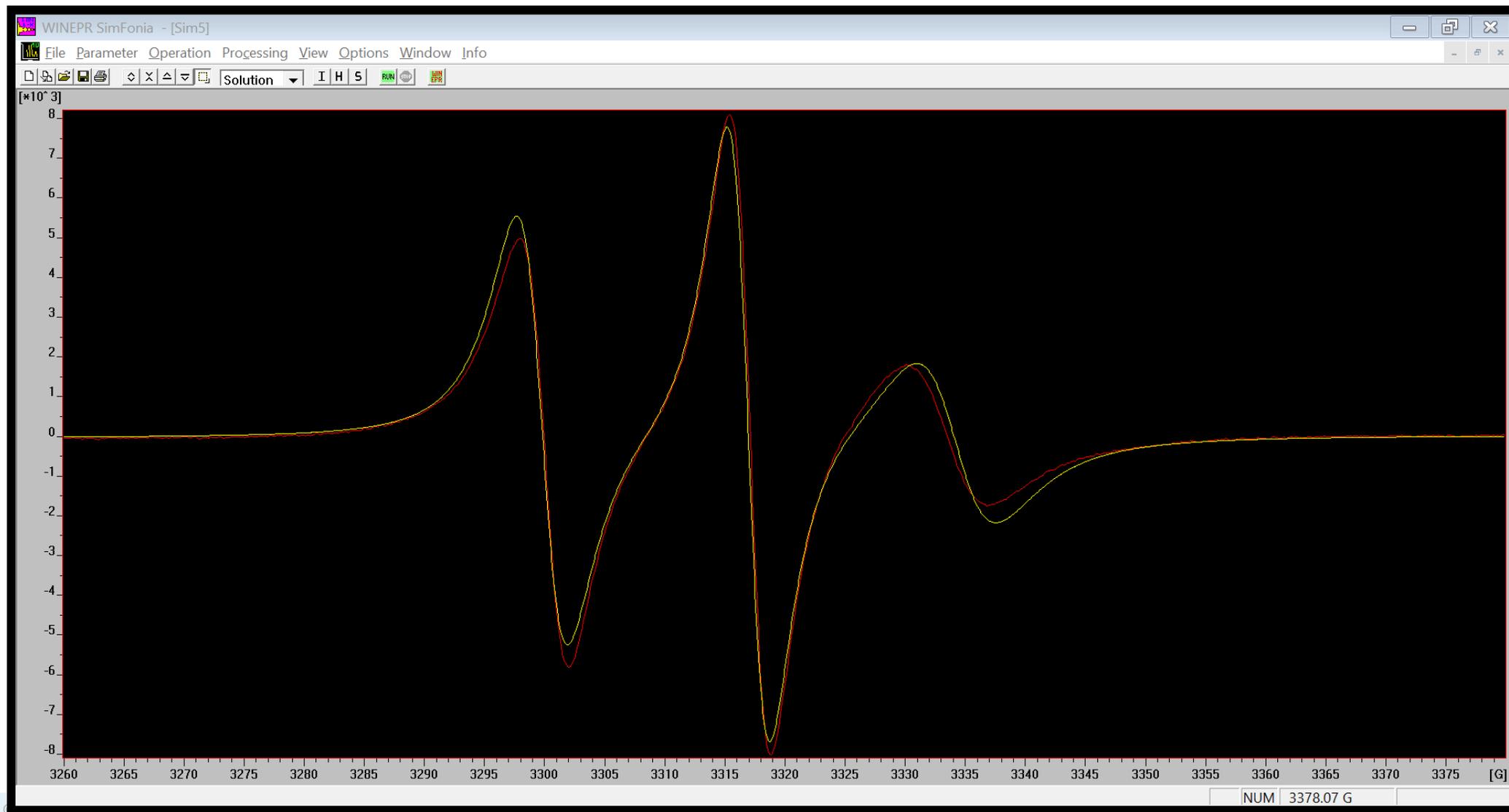
Δg alone gives 11.5G
 $\Delta A - \Delta g$ gives 15.9G
 $\Delta A + \Delta g$ gives 38.9 G

The more the spread the faster motion we need to motionally-narrow the line.

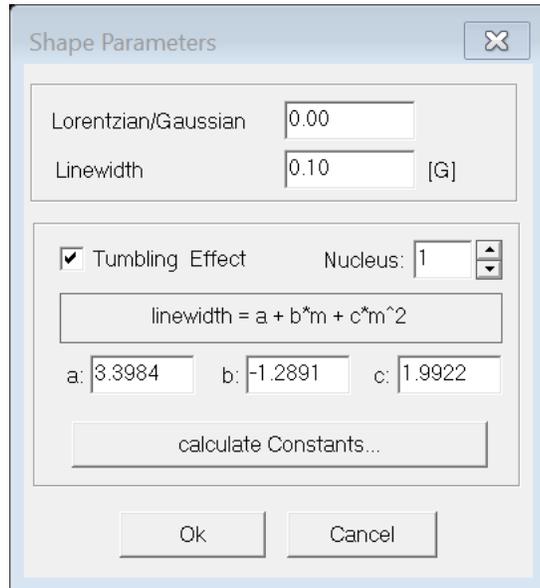
That is why **-1 line** is always broader and less intense for nitroxide spectra with motion effects!



Simulating an experimental spectrum. Tempo in water/glycerol at 230K



Simulating an experimental spectrum. Testing the model



$$B = \frac{1}{2} \left[\sqrt{\frac{I(0)}{I(+1)}} - \sqrt{\frac{I(0)}{I(-1)}} \right] = -1.51$$

$$C = \frac{1}{2} \left[\sqrt{\frac{I(0)}{I(+1)}} + \sqrt{\frac{I(0)}{I(-1)}} - 2 \right] = 2.01$$

τ from C:

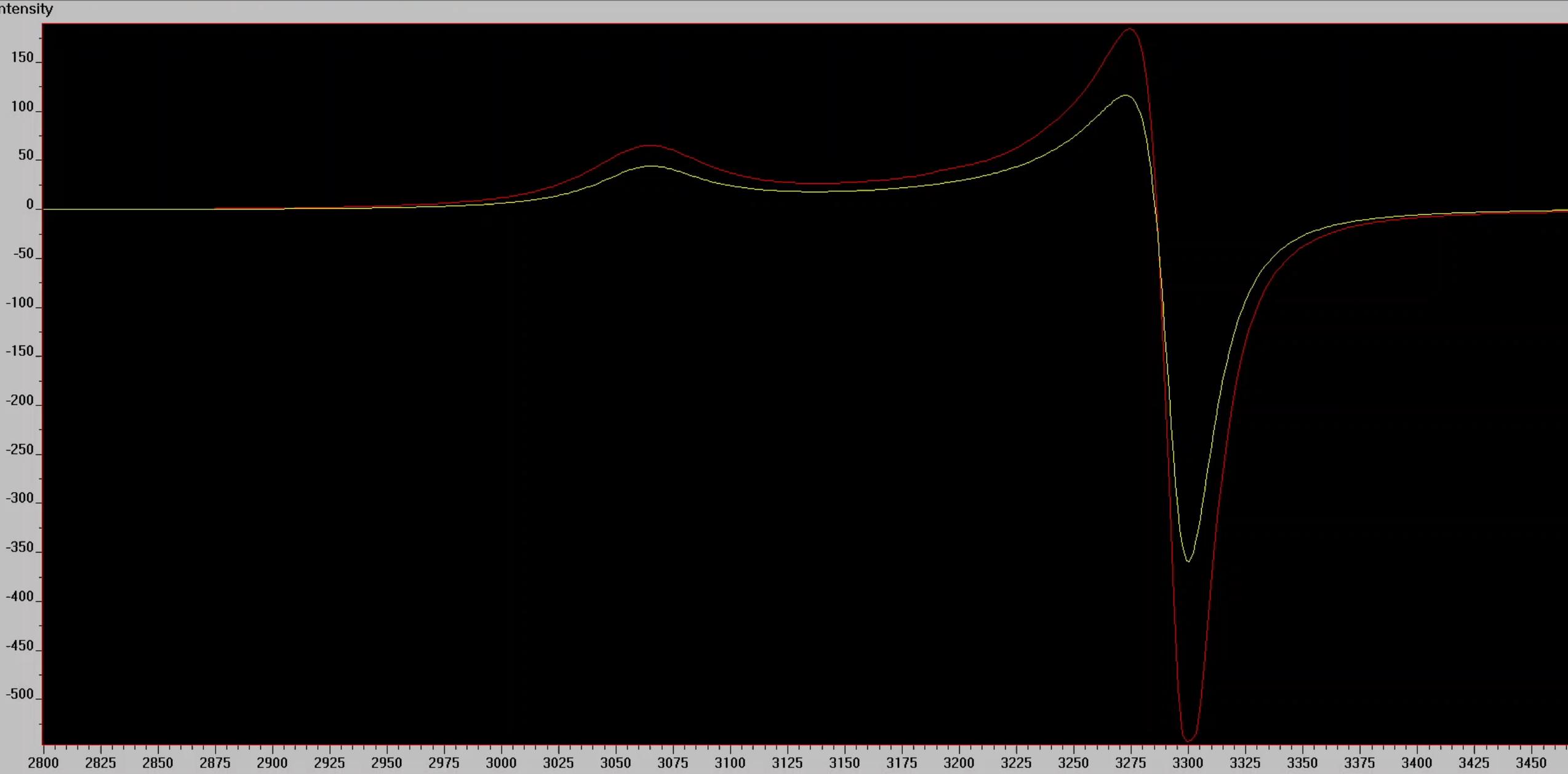
τ from the $I(+1)/I(-1)$ formula applied to the experimental spectrum

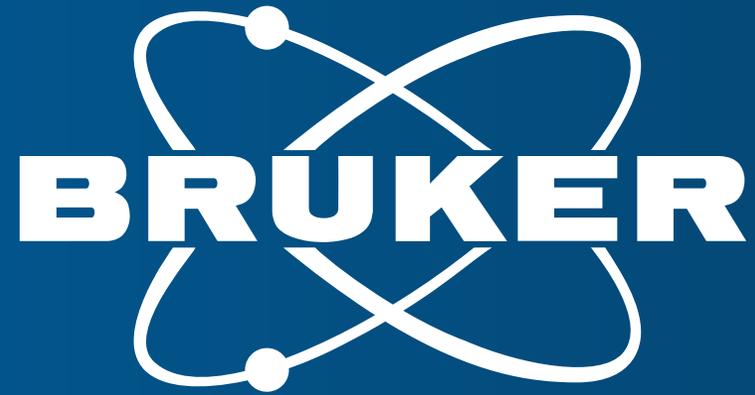
τ from the simulations based on stochastic Liouville equation

2.1 ns

2.4 ns

1.7 ns





Innovation with Integrity

Any questions?

Thank you!