



BRUKER NANO ANALYTICS' CULTURAL HERITAGE WEBINAR SERIES 2022

Handheld XRF in Cultural Heritage Studies

I. Back to basics – Taking control of Your Path to Meaningful Compositional Information



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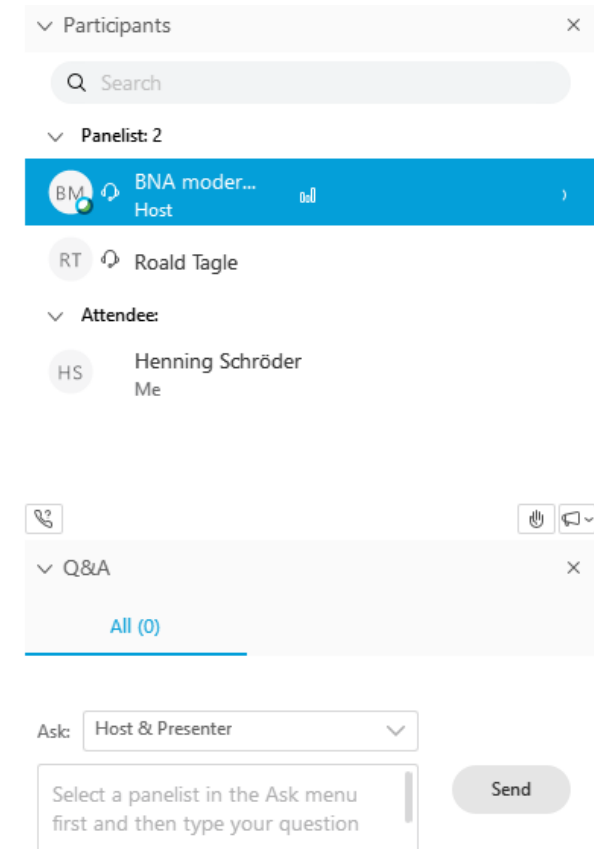


Art & Conservation Webinar Series

Handheld XRF in Cultural Heritage Studies

If you have questions during this webinar, please **type your questions**, thoughts, or comments in the **Q&A box** and **press Send**.

We ask for your understanding if we do not have time to discuss all comments and questions within the session. Any unanswered questions or comments will be answered and discussed by e-mail or in another WebEx session.



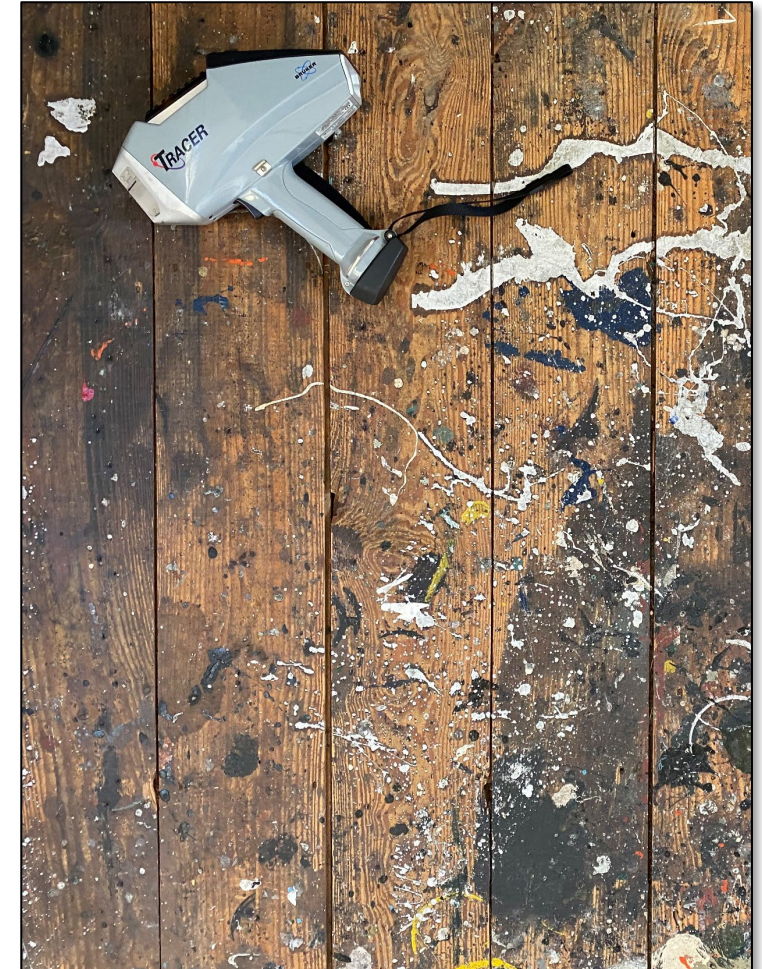
Handheld XRF in Cultural Heritage Studies Webinar Series



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- Three webinars specifically addressing use of handheld-XRF, presented as part of the ongoing series on elemental analysis in Cultural Heritage Studies
 - I. Back to basics - taking control of your path to meaningful information
 - II. Approaches to challenging measurements - Paintings, pigments and objects
 - III. Quantitative data - what do the numbers mean?
- Other webinars in the series: collaboration with our colleagues at Bruker Optics, where we will demonstrate the benefits of integrating elemental and spectroscopic techniques



Back to basics with Handheld XRF

Our speakers



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Maggi Loubser

**Programme Manager – Tangible Heritage Conservation
University of Pretoria**



Dr. Nigel Kelly

**Senior Market Applications Scientist
Bruker Nano Analytics**

Back to basics with Handheld XRF Presentation Outline



- 01 Anatomy of a spectrometer
- 02 Generation of X-rays
- 03 Modulating the primary X-ray source -
Voltage, current, filters
- 04 The spectrum
Detection of X-rays, understanding artifacts
- 05 Other contributions to the result

Back to basics with Handheld XRF

Why?



“Just because you can aim and shoot and get numbers, it does not mean the numbers mean anything!”

Back to basics with Handheld XRF

XRF is a mature technique



XRF on the Mars Rover
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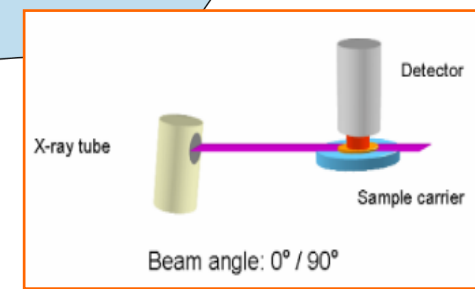
Handheld XRF



~1990
Laboratory based WDXRF with PC control



Bench top EDXRF and WDXRF



Total Reflection XRF (TXRF)

WDXRF ~1964 vintage



~1960 first WDXRF



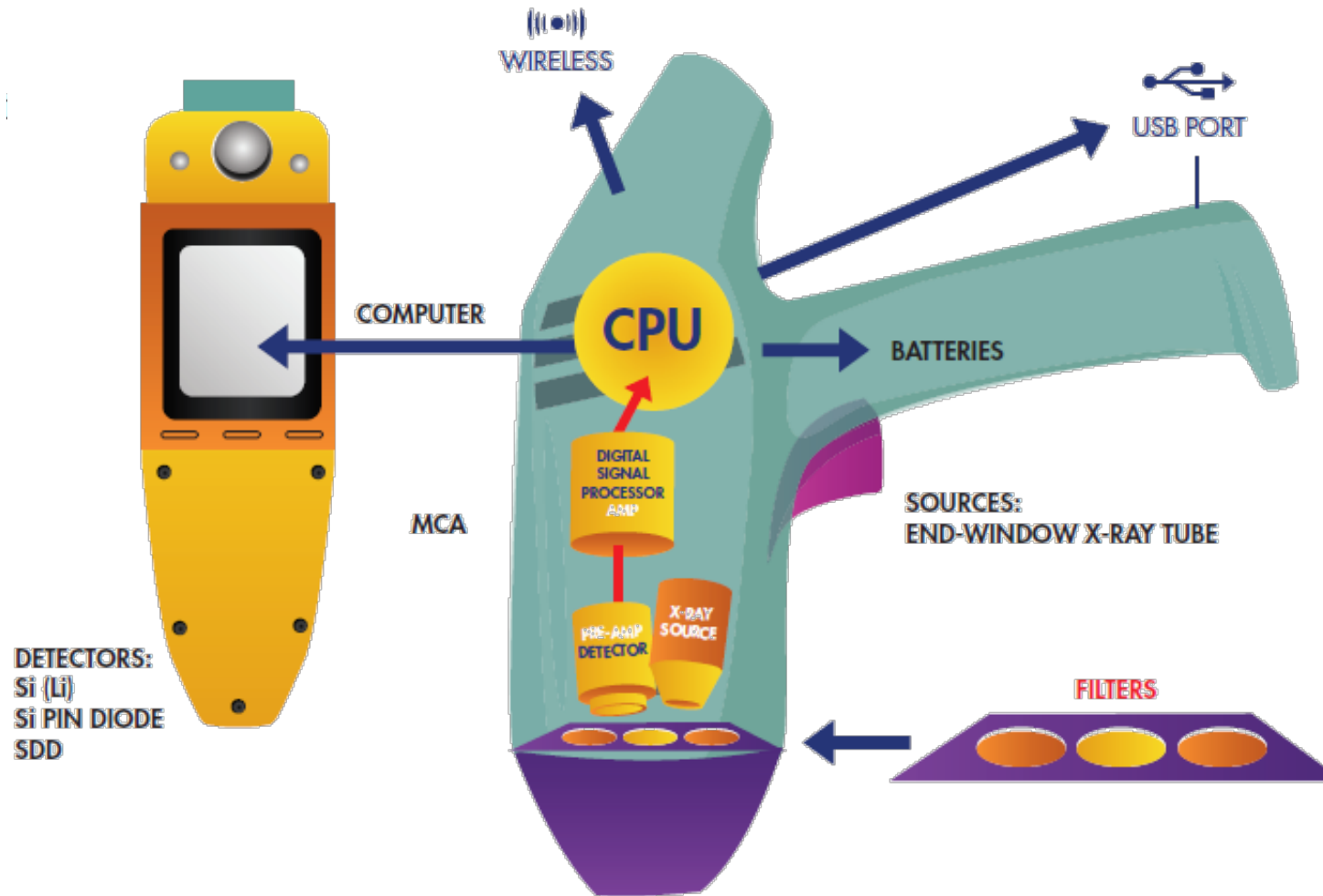
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HANDHELD XRF IN CULTURAL HERITAGE STUDIES

Back to basics in Handheld XRF Anatomy of a Spectrometer

Knowing your handheld XRF Anatomy of a spectrometer



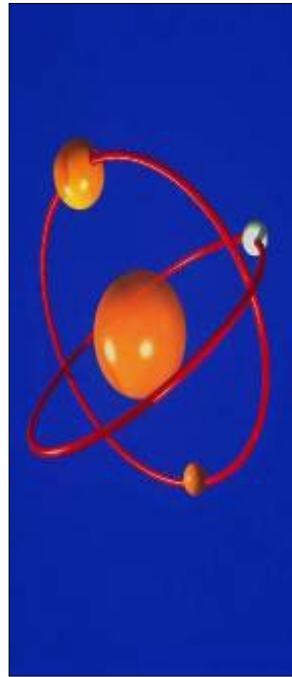
Knowing your handheld XRF Anatomy of a spectrometer



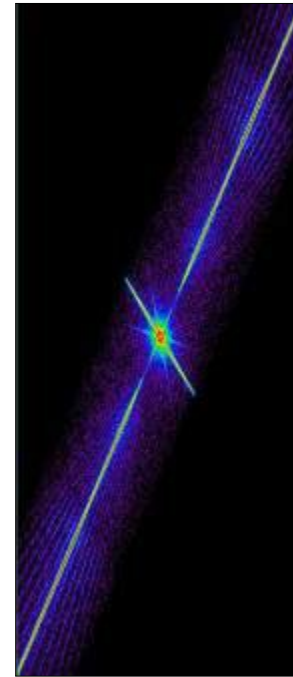
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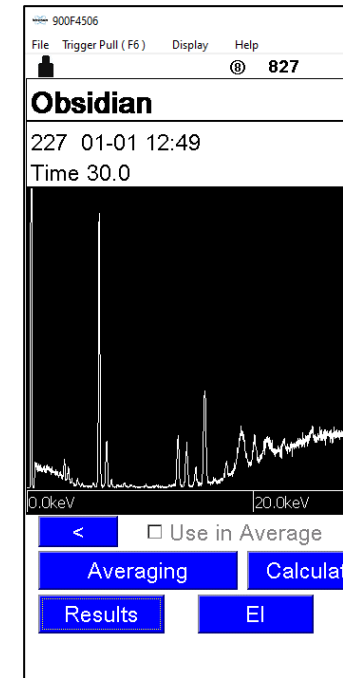
Excitation
Source



Ionizes
Atoms



Energy
Emission



Spectral
Capture

El	PPM	+/- [*3]
Mn	181	59
Fe	12K	372
Zn	41	14
Ga	9	5
Rb	135	11
Sr	99	7
Y	23	5
Zr	197	9
Nb	10	5

Below the table are buttons for '<', 'Use in Average', 'Averaging', 'Calculate', 'Spectrum', and 'Edit Info'.

Test
Results

The process – in 5 steps

Knowing your handheld XRF Anatomy of a spectrometer



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Excitation
Source

Ionizes
Atoms

Energy
Emission

Spectral
Capture

Test
Results

Atoms in a sample
are excited by a
beam of primary
X-rays



Knowing your handheld XRF Anatomy of a spectrometer



Excitation
Source

Ionizes
Atoms

Energy
Emission

Spectral
Capture

Test
Results

Excited atoms
relax by emitting
secondary
X-rays



Knowing your handheld XRF Anatomy of a spectrometer



Excitation Source

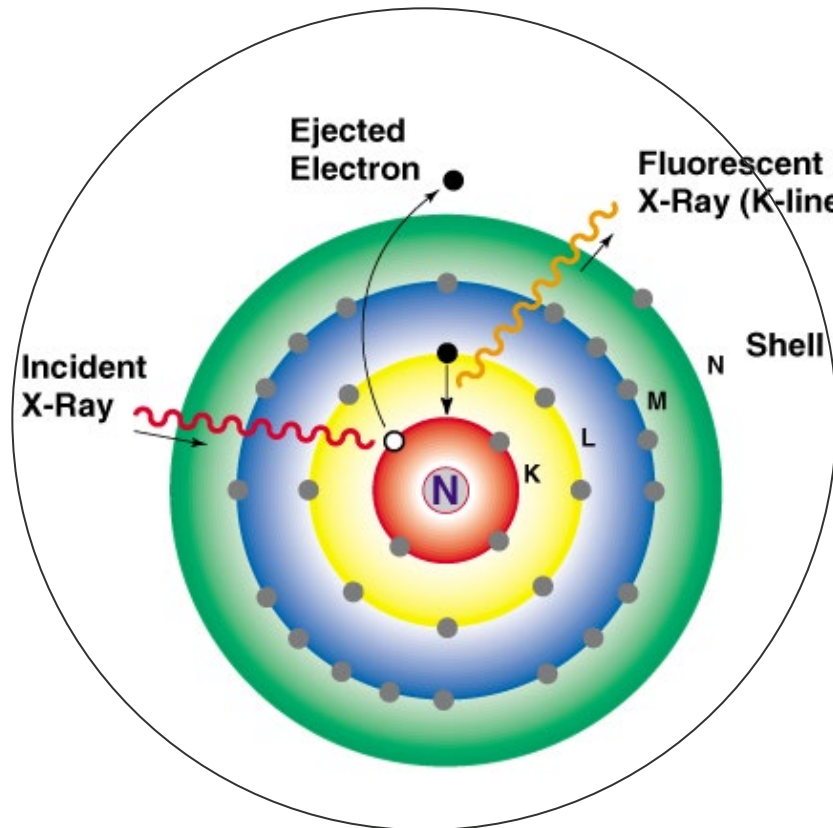
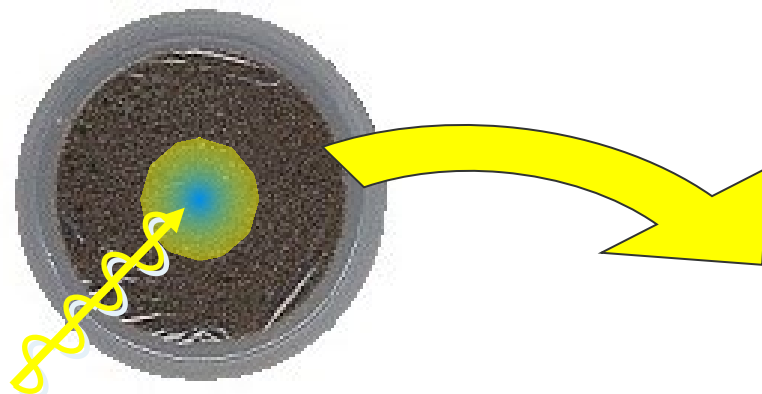
Ionizes Atoms

Energy Emission

Spectral Capture

Test Results

Excited atoms relax by emitting secondary X-rays



Knowing your handheld XRF Anatomy of a spectrometer



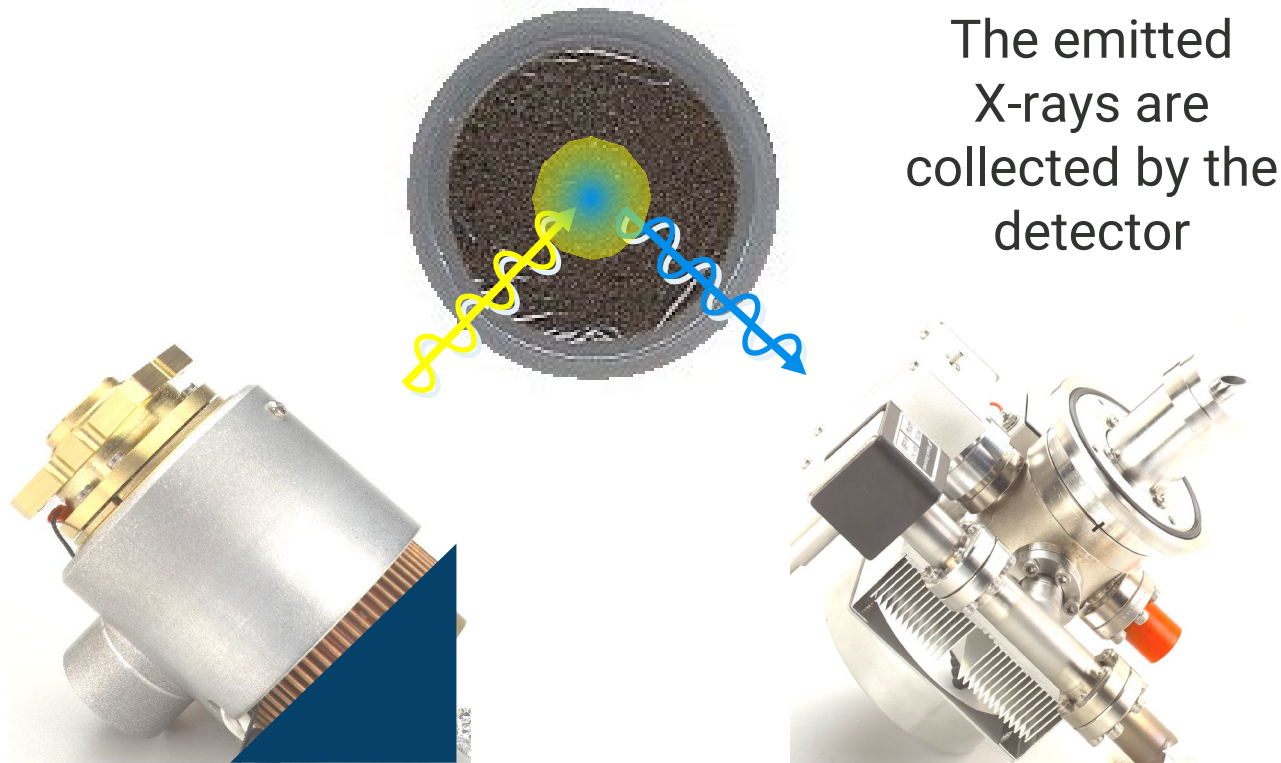
Excitation
Source

Ionizes
Atoms

Energy
Emission

Spectral
Capture

Test
Results



Knowing your handheld XRF Anatomy of a spectrometer



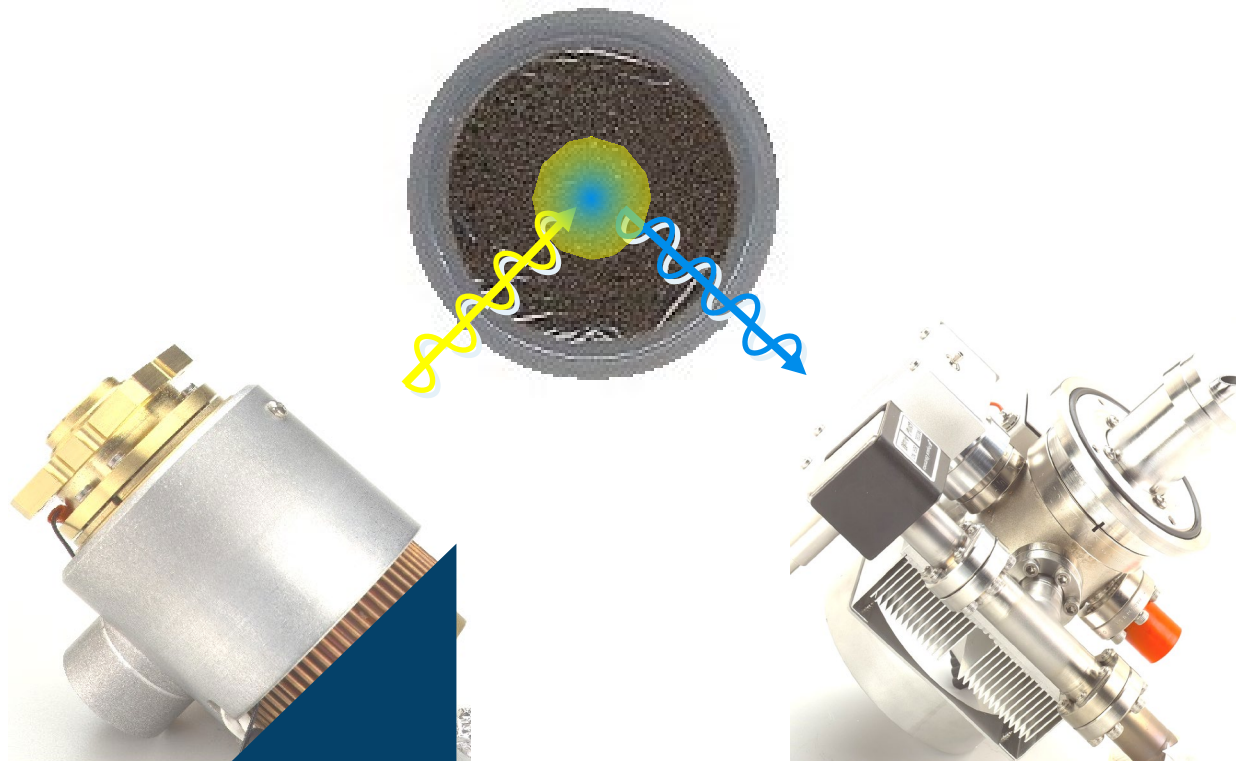
Excitation
Source

Ionizes
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Test
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Each X-ray is
converted into a
voltage pulse \propto to
its energy



Knowing your handheld XRF Anatomy of a spectrometer



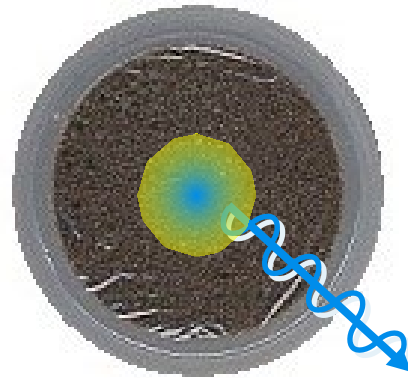
Excitation
Source

Ionizes
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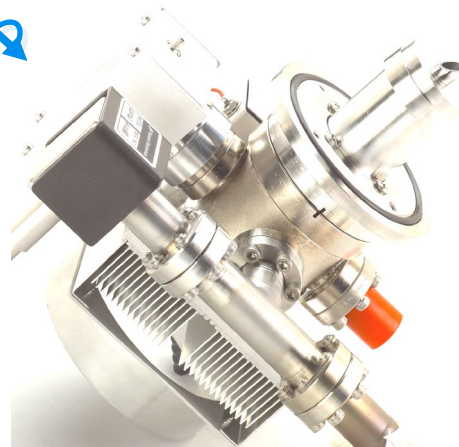
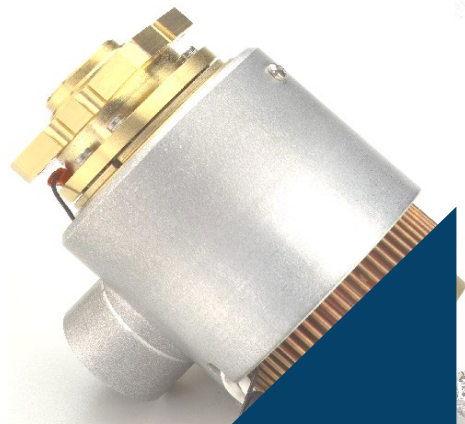
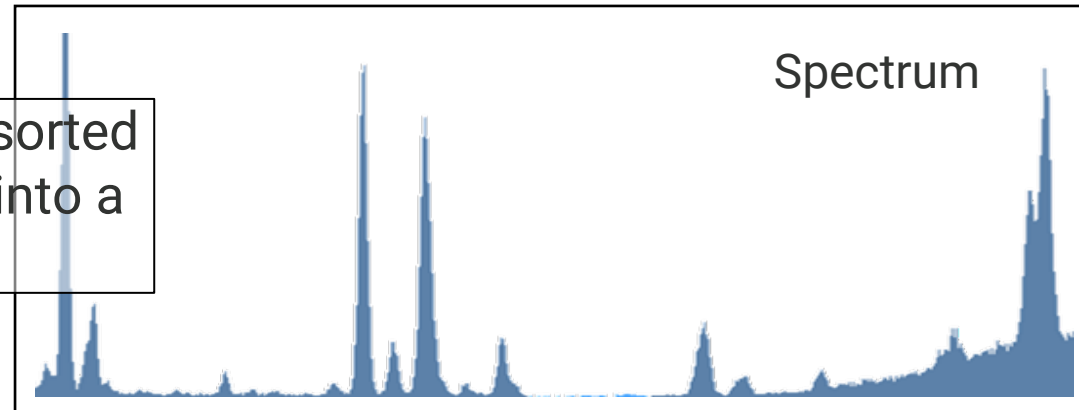
Energy
Emission

Spectral
Capture

Test
Results



The pulses are sorted
and combined into a
spectrum



Knowing your handheld XRF Anatomy of a spectrometer



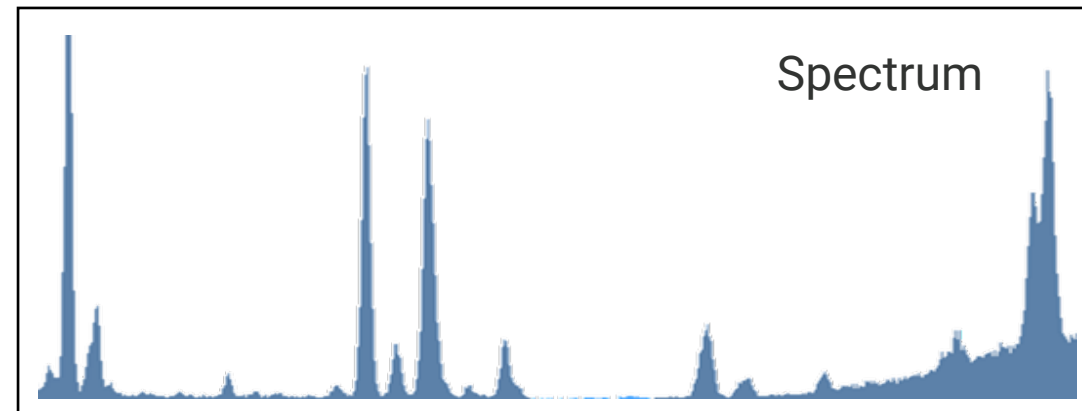
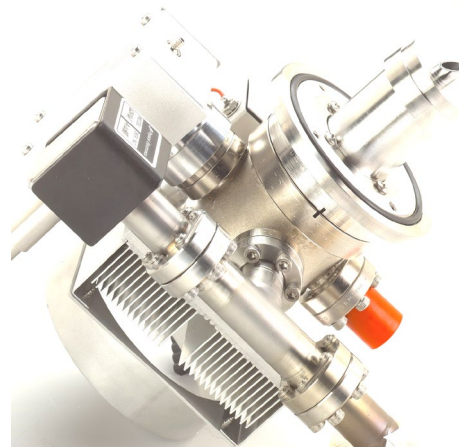
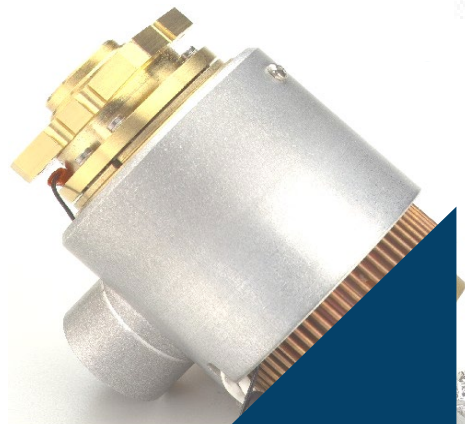
Excitation
Source

Ionizes
Atoms

Energy
Emission

Spectral
Capture

Test
Results



Results

Element	Concentration	Std. Dev.	Peak (cps)
Unknown			
SiO ₂	34.8 %	0.1	3830
Fe	6.82 %	0.05	670
Zn	23.83 %	0.05	6680
Ba	2.41 %	0.01	1246
Pb	5.78 %	0.03	1196

Peak intensities may be related to concentration, thickness or other factors



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HANDHELD XRF IN CULTURAL HERITAGE STUDIES

Back to basics in Handheld XRF

Generation of X-rays

Generation of X-rays & Excitation Potentials

What are X-rays?



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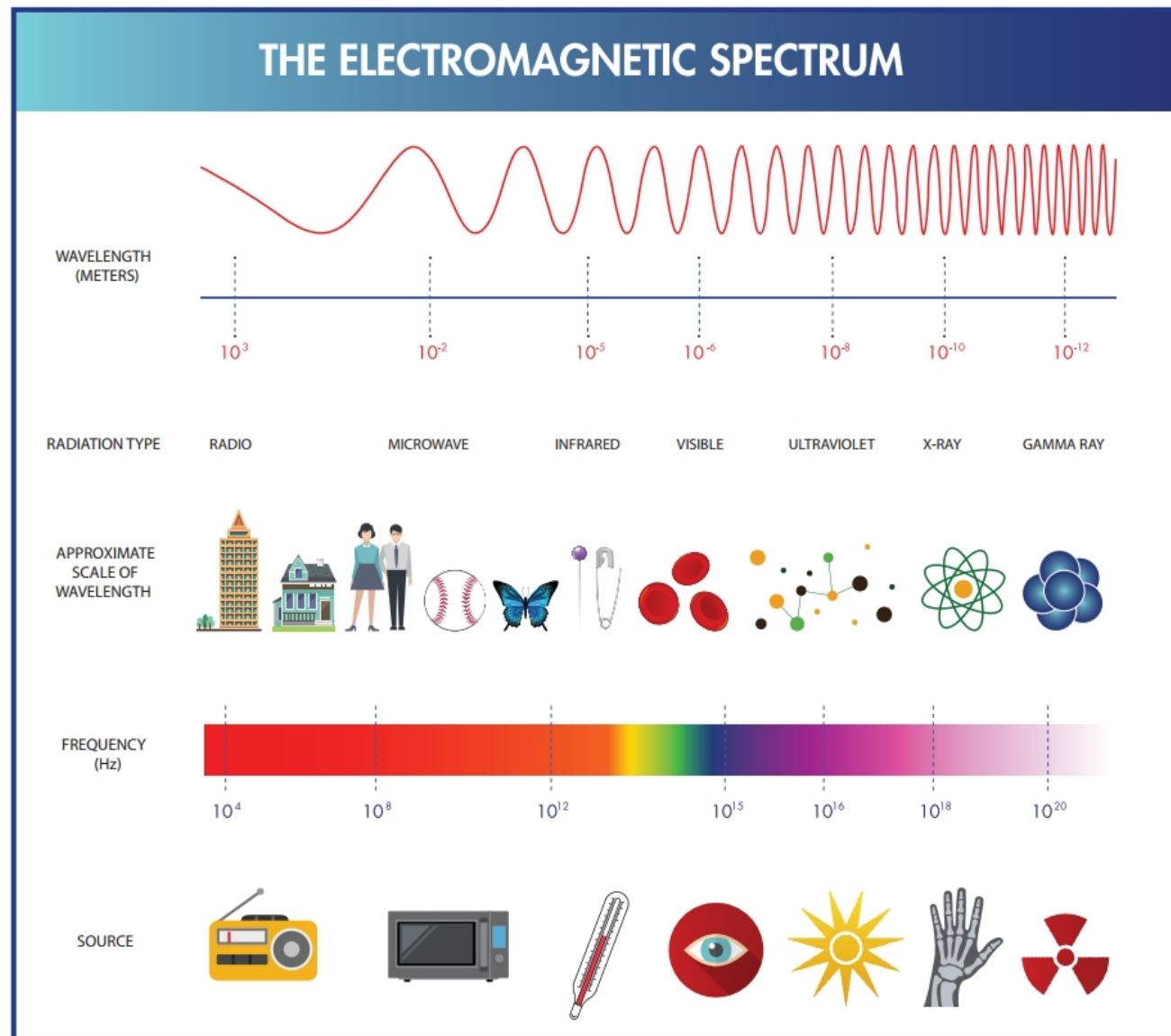


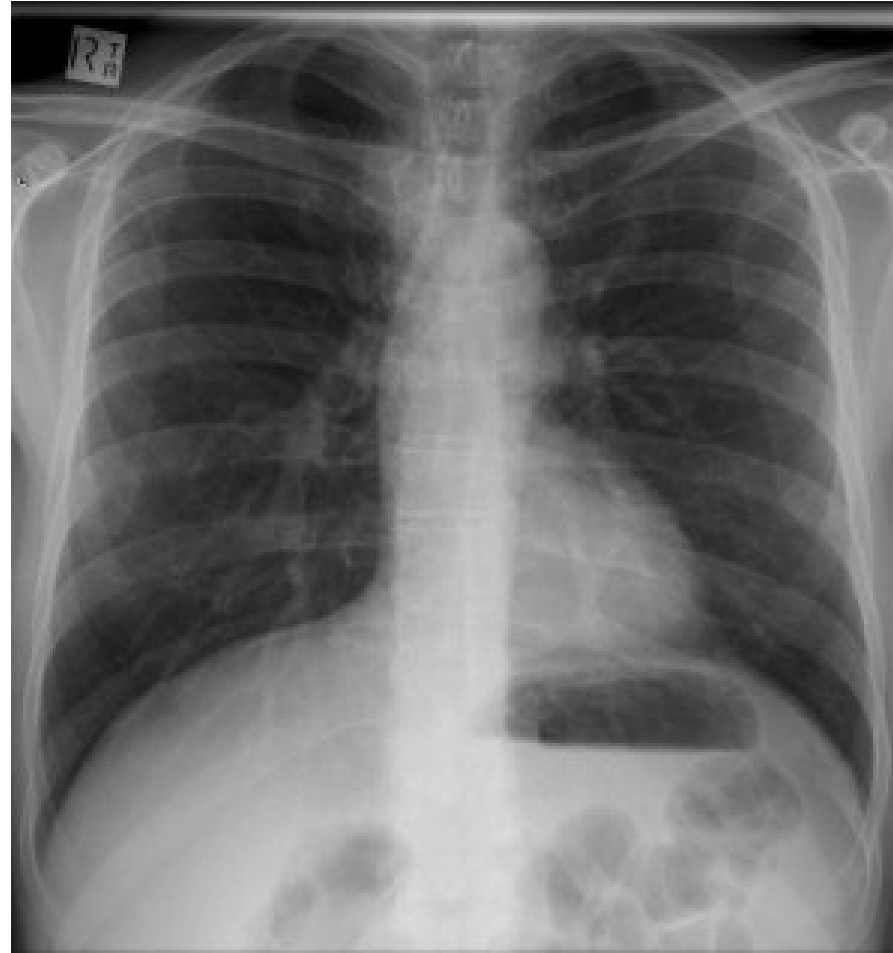
Image from: Anikó Bezur, Lynn Lee, Maggi Loubser, and Karen Trentelman, 2020. Handheld XRF in Cultural Heritage: A Practical Workbook for Conservators. Getty Conservation Institute. ISBN 1937433625

Generation of X-rays & Excitation Potentials

What are X-rays?



- The same X-rays used in medical diagnostics
- The difference – energies used are a lot lower when used on humans
 - *we do not want to excite our atoms!*



- If a spectrometer is added and not just the contrast film, can measure
 - Ca and P (from bones)
 - Fe (from Hemoglobin in blood)

Generation of X-rays & Excitation Potentials

Properties of X-rays



Binding energy of an electron

- The binding energy is the minimum X-ray energy required to expel an electron from a given atom sub-shell (K,L,M shell).
- The basic unit of binding energy is the kiloelectron volt (keV).
- The binding energy has the same numerical value as the Excitation Potential (units of kV)

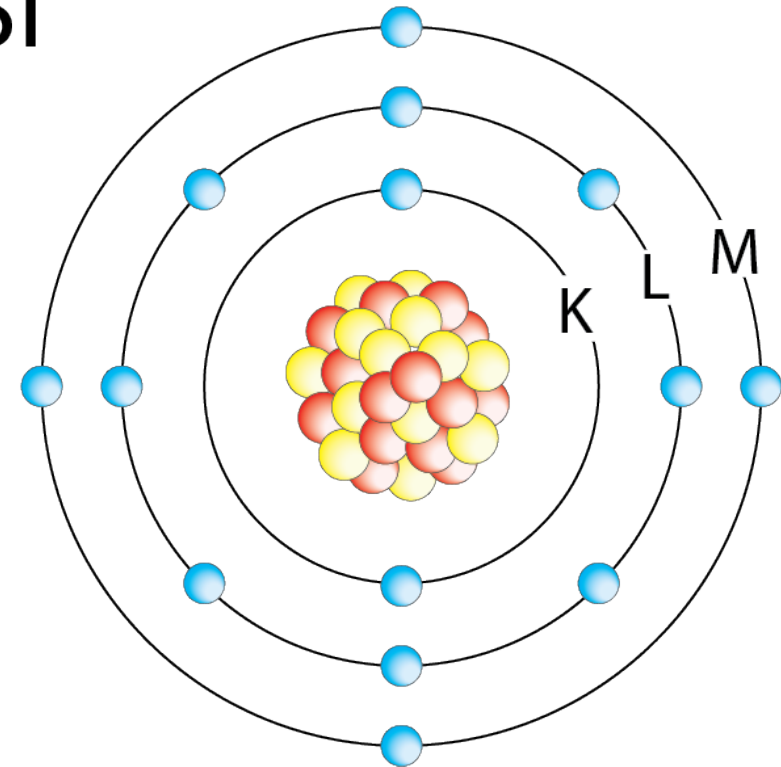
Wavelength of an X-ray photon

- Units of measurement:
 - Angström unit (Å) or nanometers (nm)

$$1 \text{ \AA} = 10^{-10} \text{ m}$$

$$10 \text{ \AA} = 1 \text{ nm}$$

Si



Generation of X-rays & Excitation Potentials

Properties of X-rays

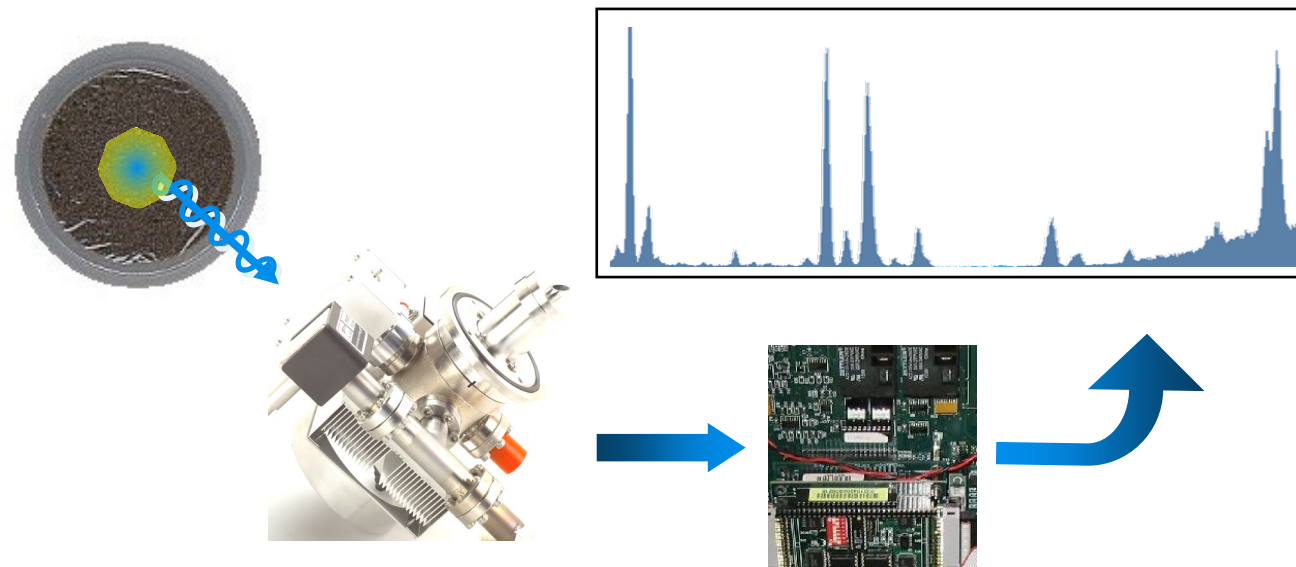


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Intensities of X-rays

- Measured based on the amount of ionization they cause
- This occurs in a detector where the ionization results in electron pair formation that can produce a pulse
- X-ray intensity is reported as the number of counts measured per unit of time,
 - e.g., counts per second (cps)
 - kilocounts per second (kcps)



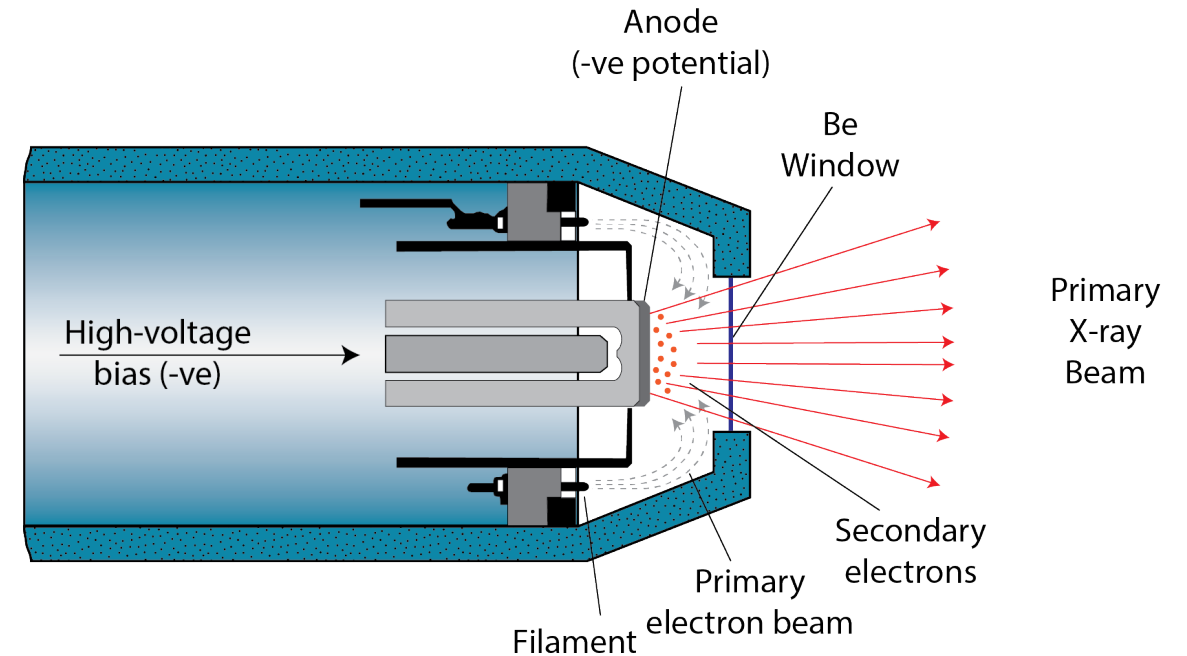
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Generation of X-rays & Excitation Potentials

How do we generate primary X-rays?



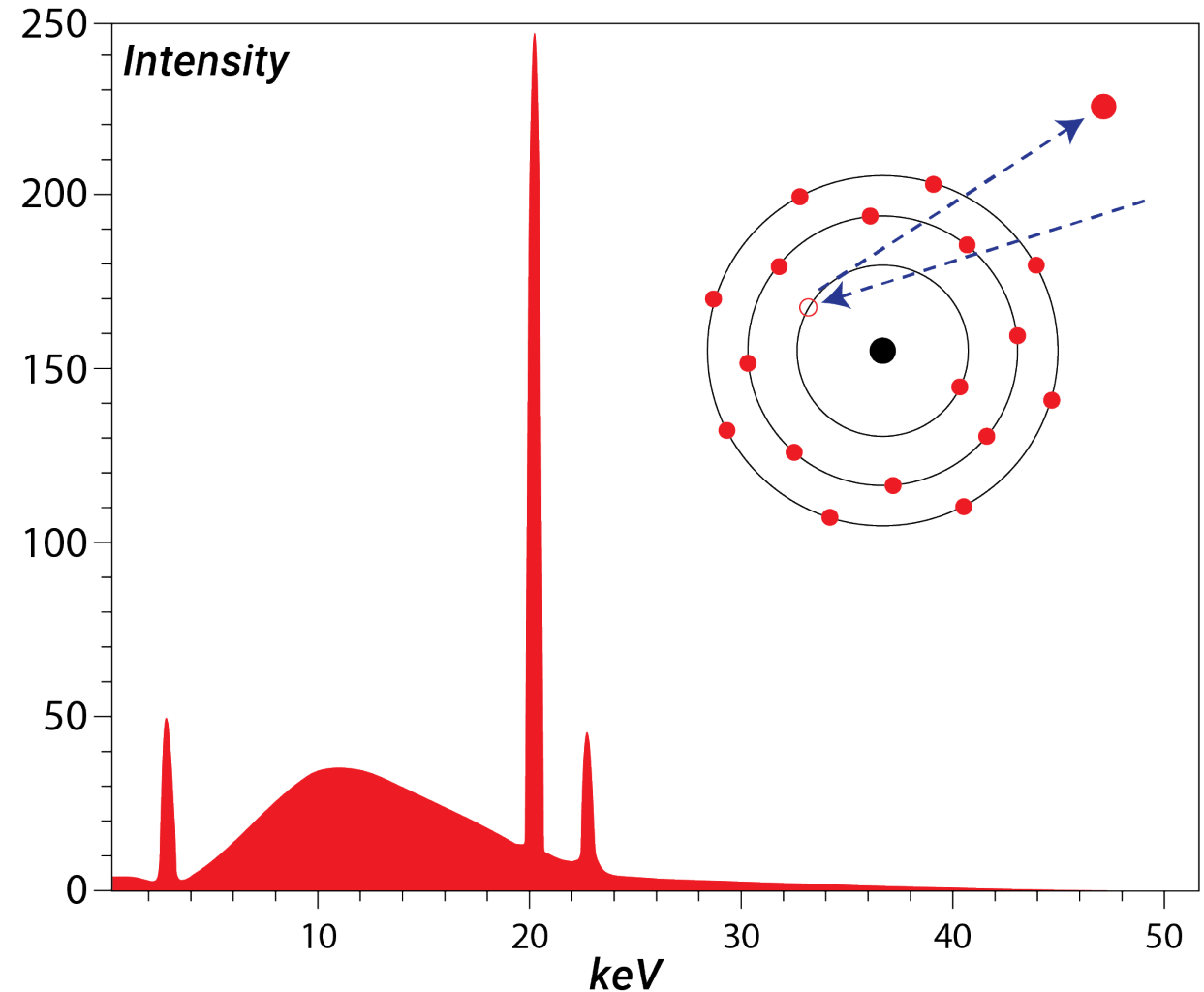
- Example of a miniature X-ray tube



Generation of X-rays & Excitation Potentials

How do we generate primary X-rays in the X-ray Tube?

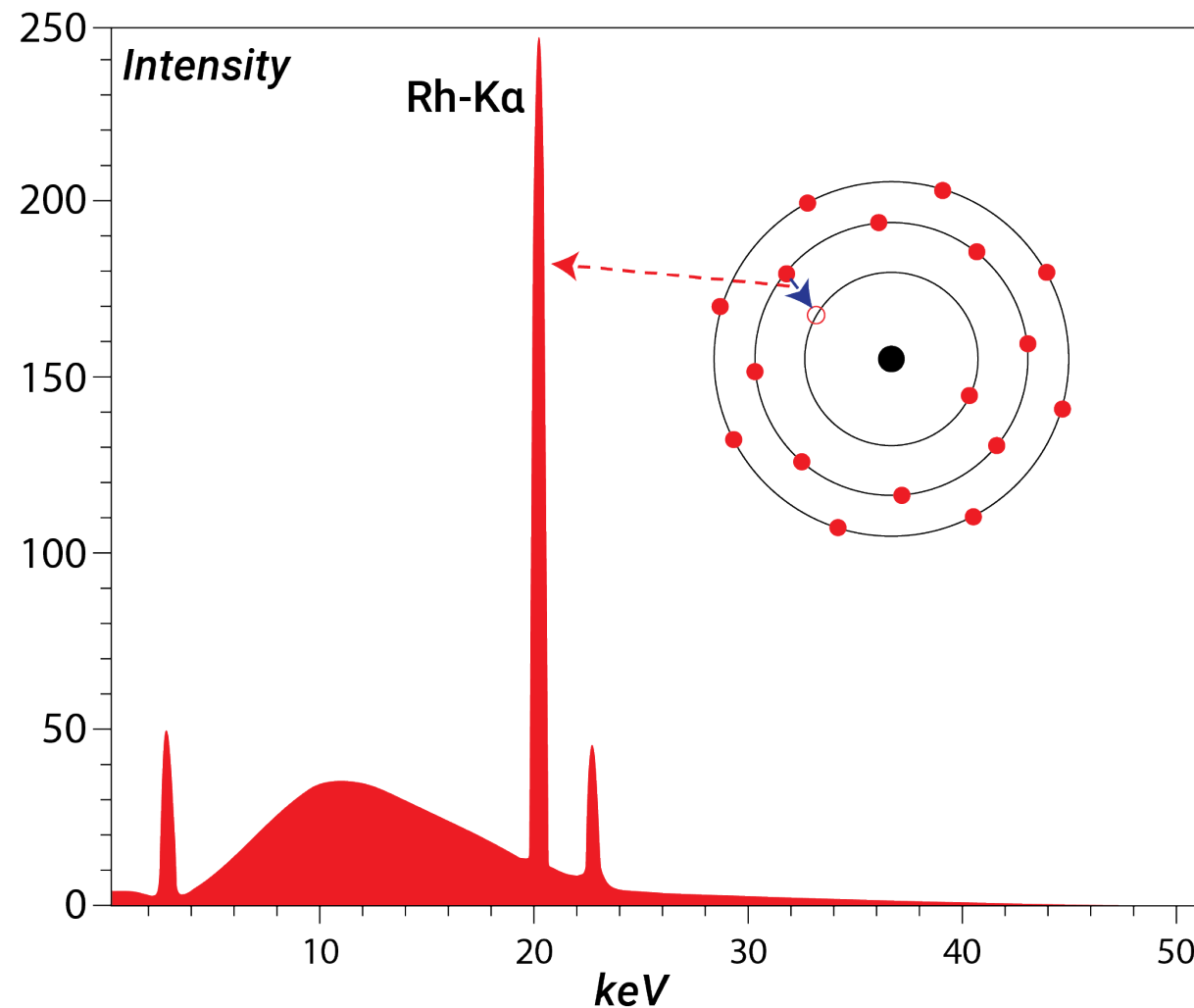
- Generating X-rays / photons
 - Electrons from the source filament displace inner shell electrons
 - Vacancy created is filled by other electron shells
 - This transition to replace the vacancy produces characteristic X-rays (photons)
 - The transition event determines the energy (and label) of the X-ray line
 - e.g.,
 - L to K = $K\alpha$
 - M to K = $K\beta$
 - M to L = $L\alpha$



Generation of X-rays & Excitation Potentials

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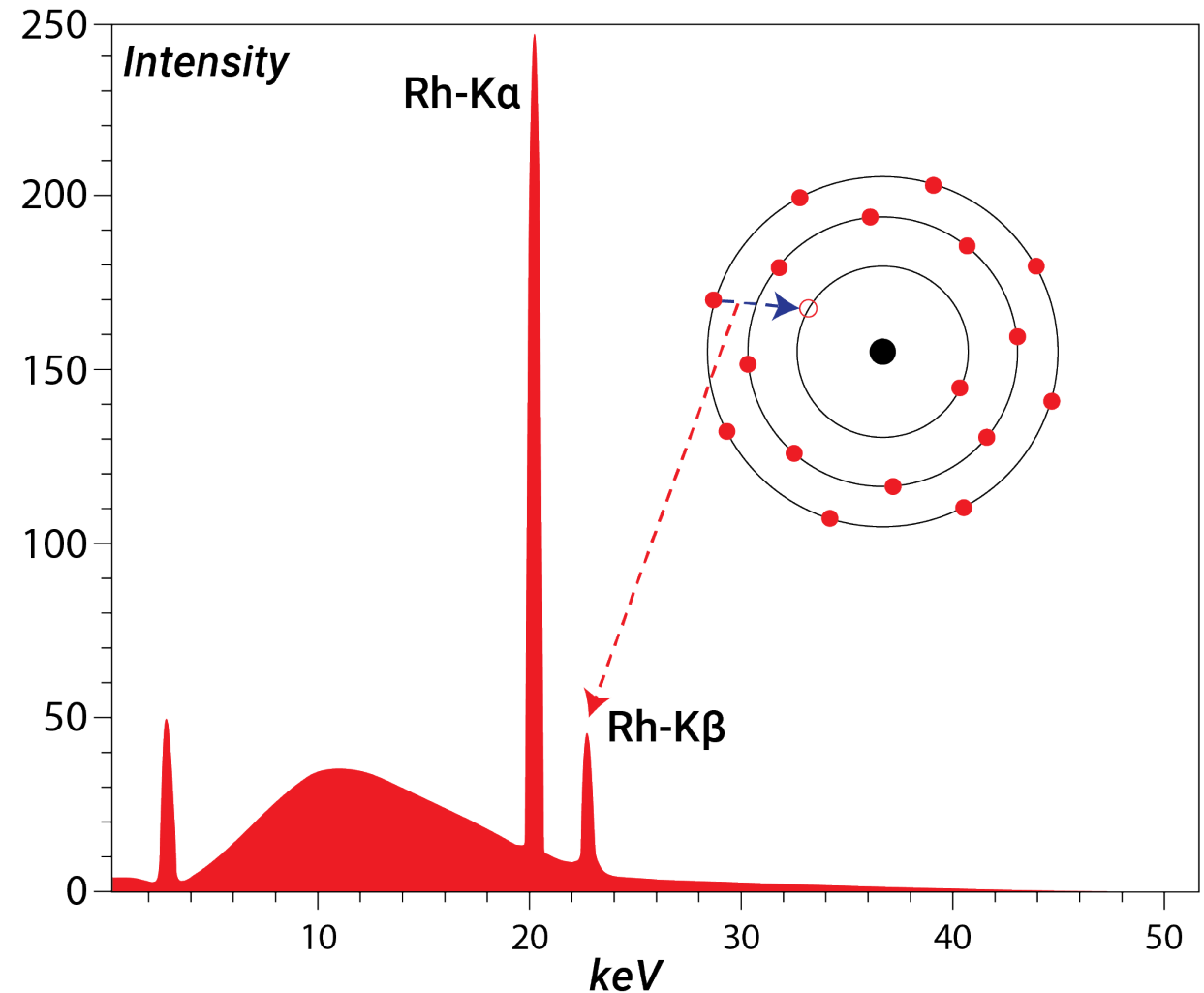
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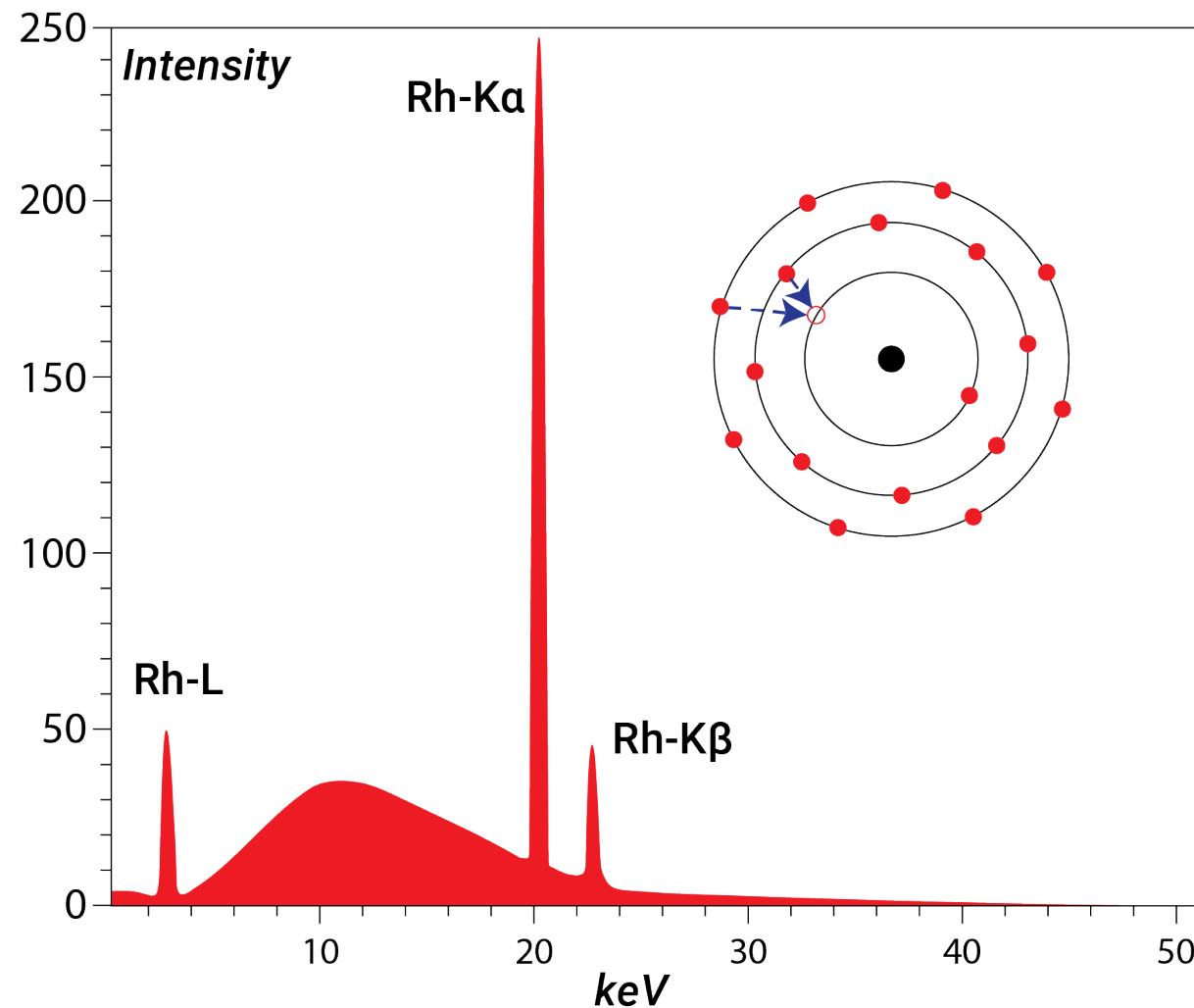
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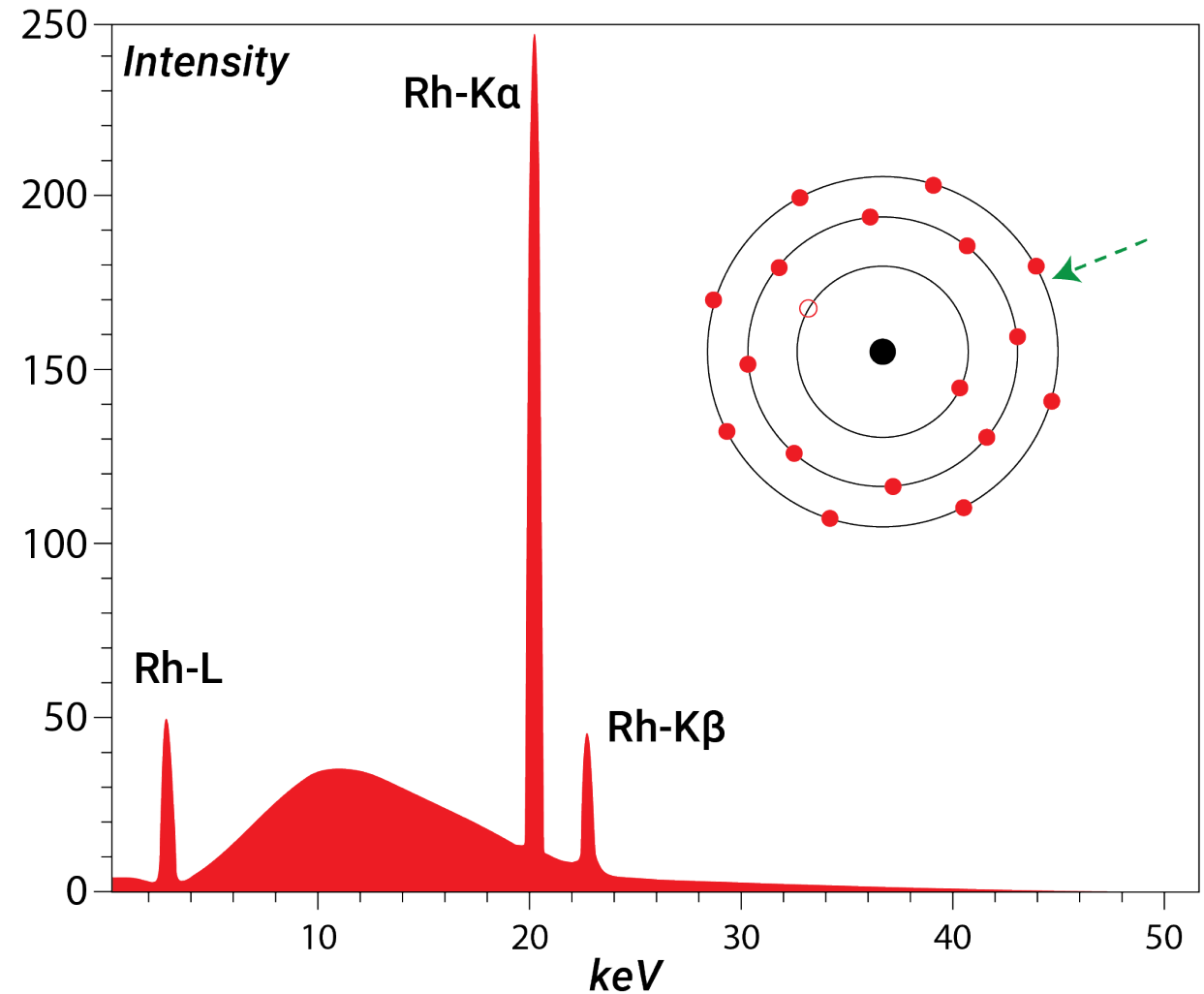
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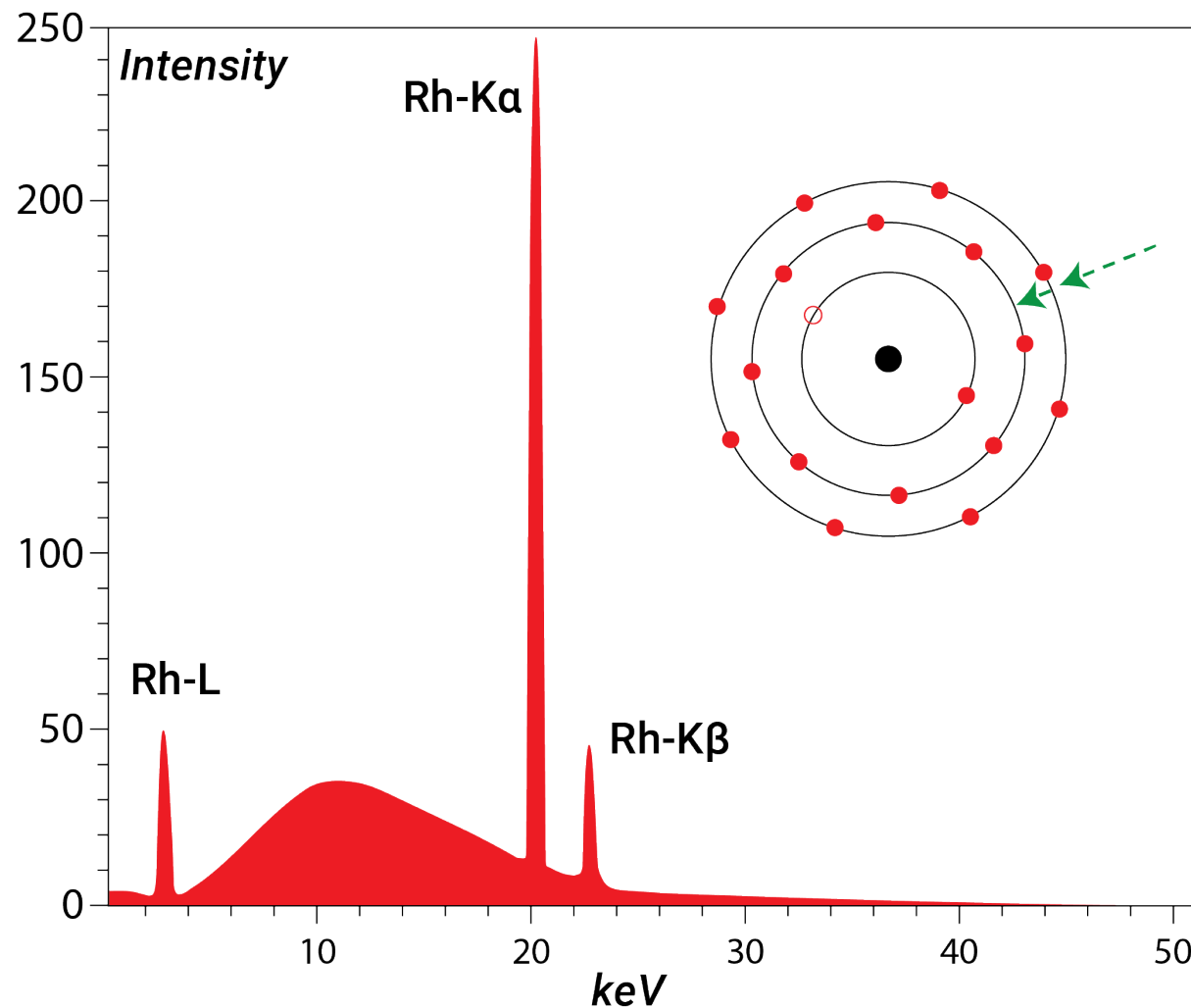
- The X-ray continuum
- Deceleration of primary electrons by an atom's electron field produces white or continuous radiation (Bremsstrahlung)



Generation of X-rays & Excitation Potentials

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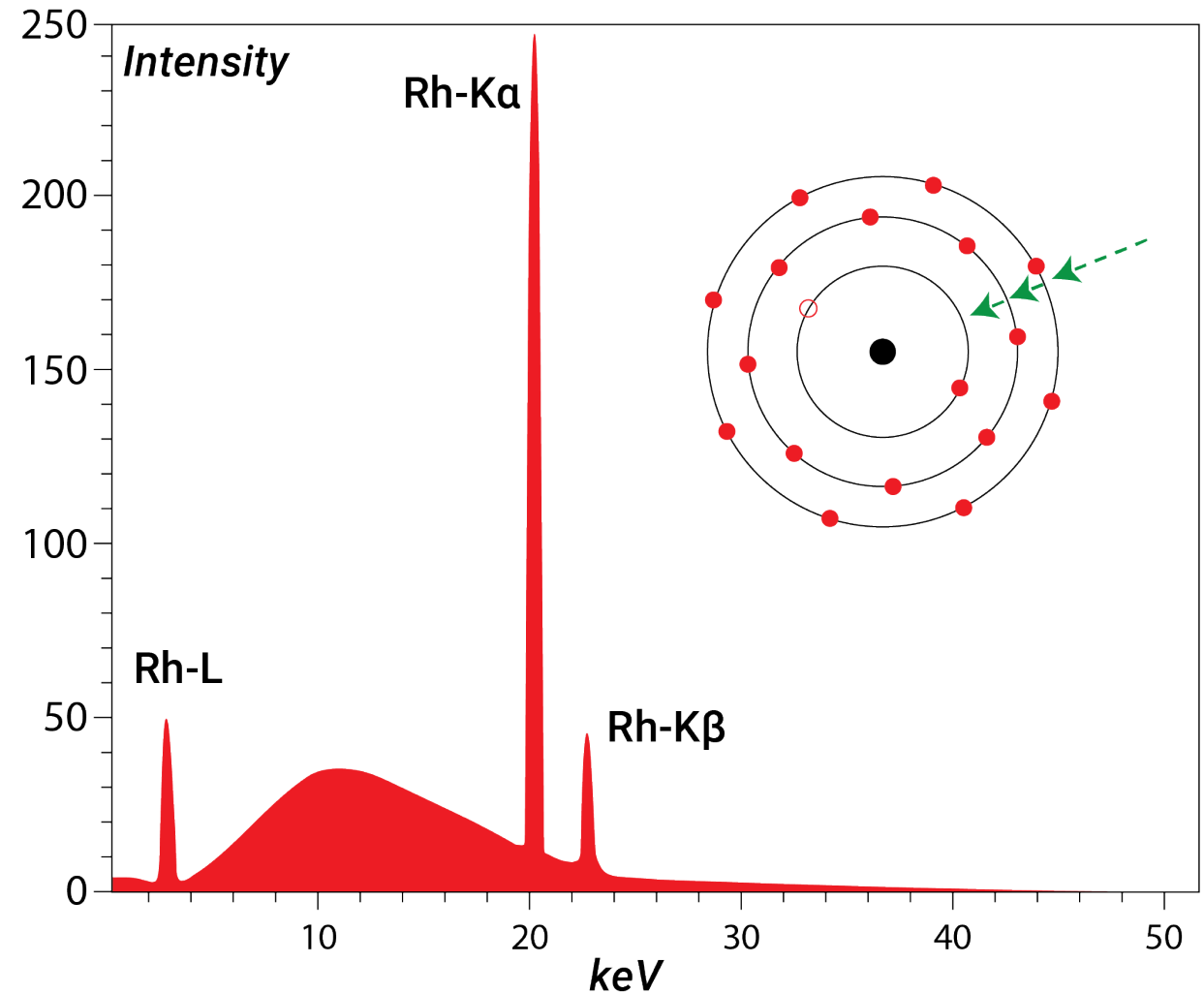
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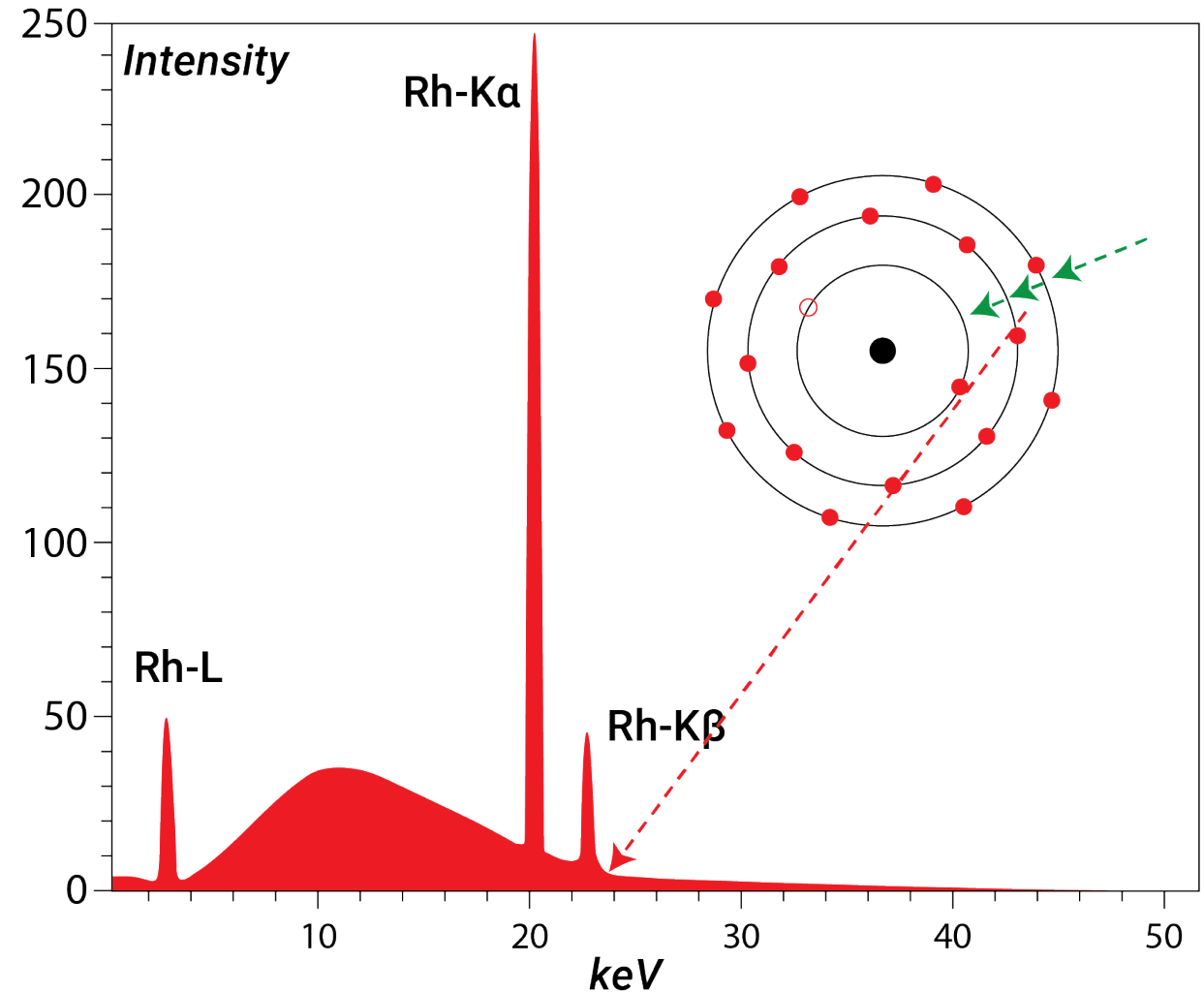
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Generation of X-rays & Excitation Potentials

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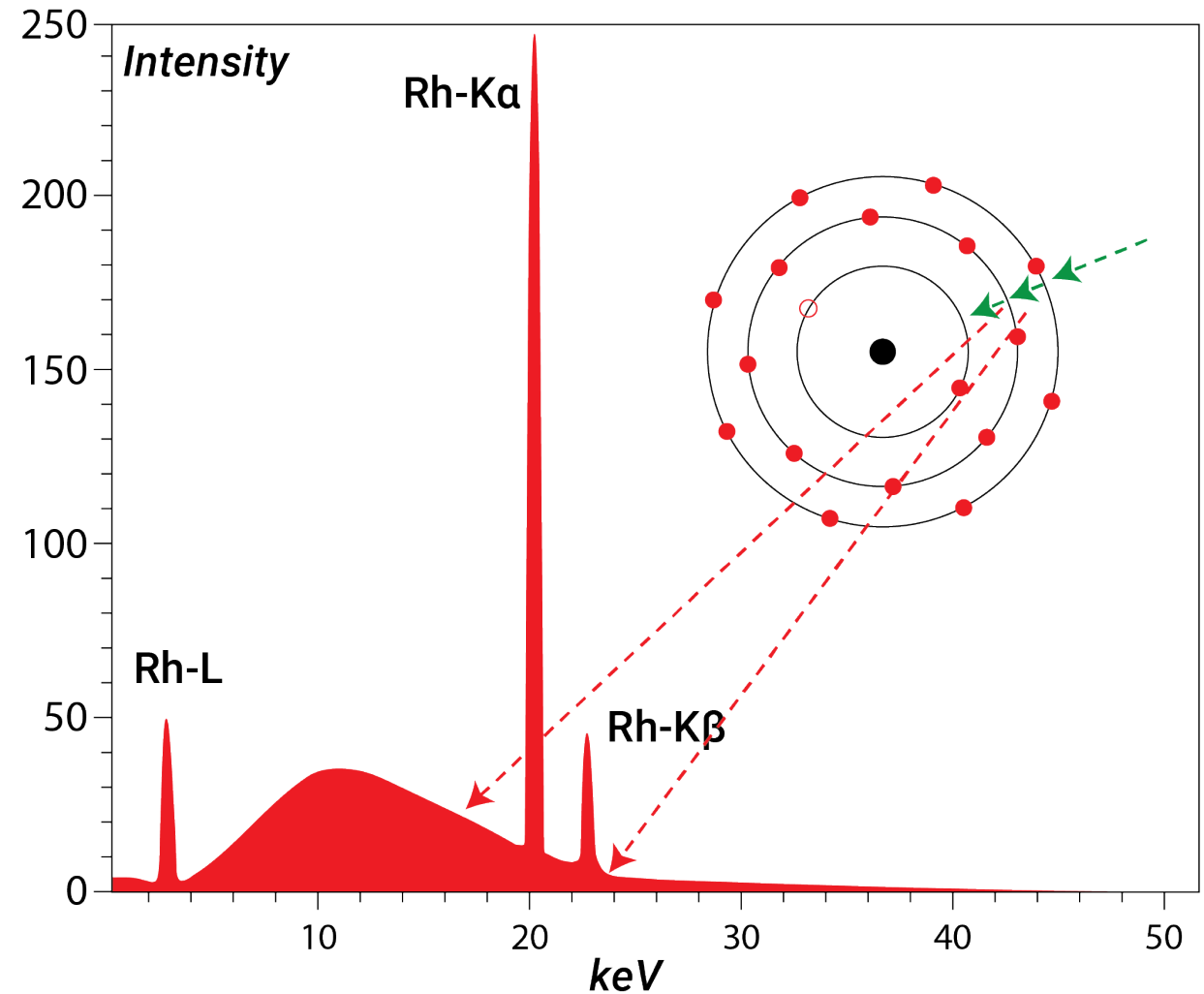
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Generation of X-rays & Excitation Potentials

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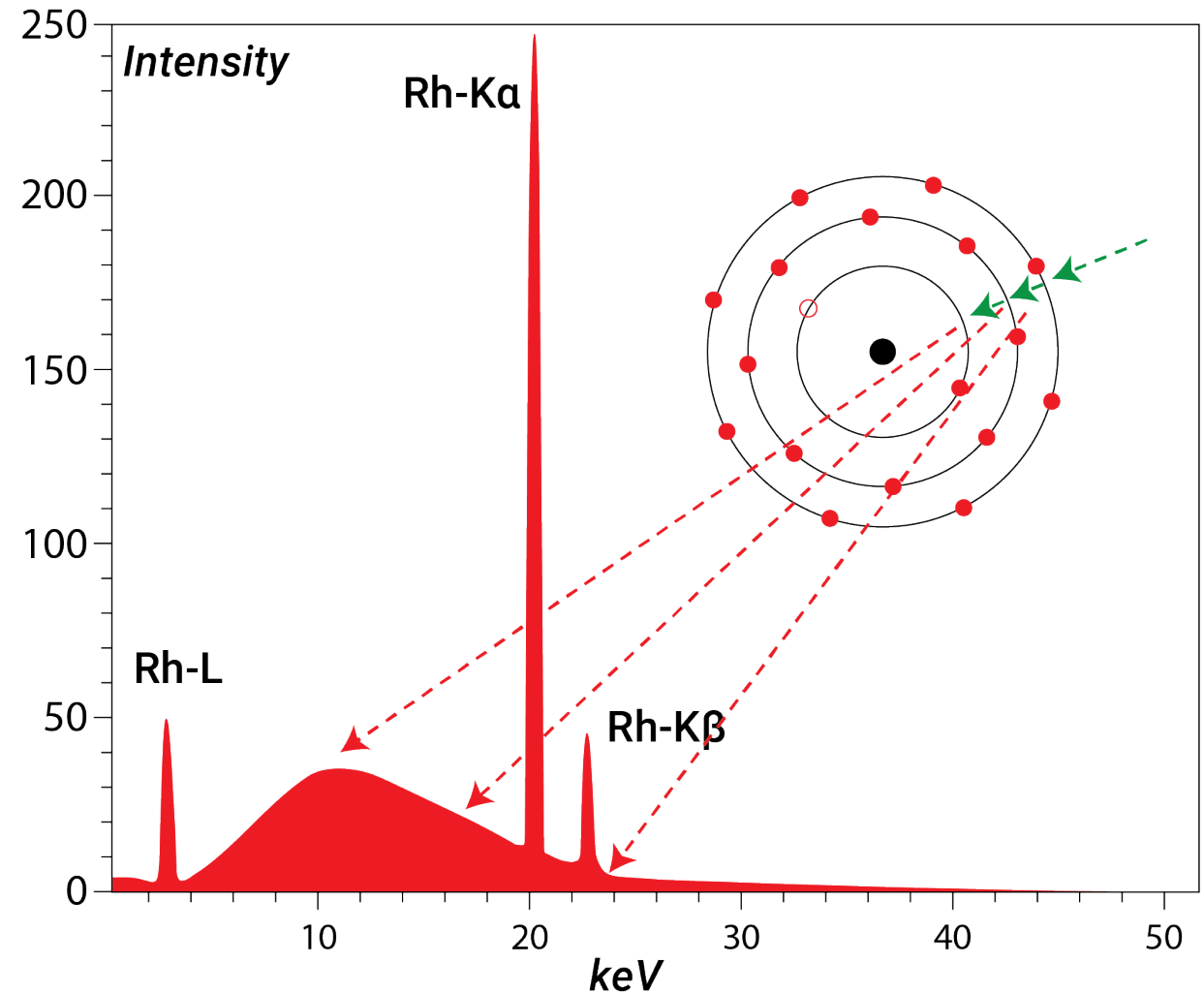
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Generation of X-rays & Excitation Potentials

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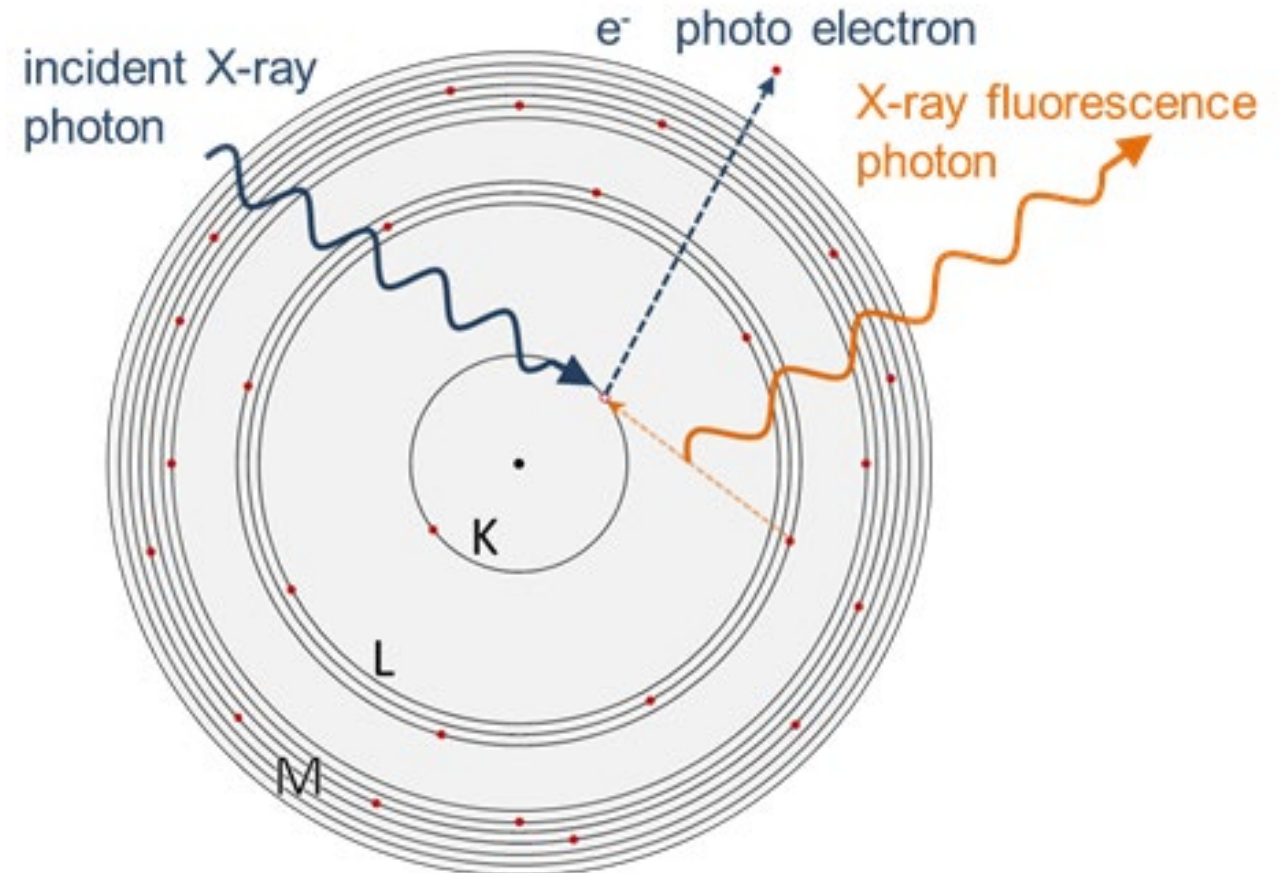


Generation of X-rays & Excitation Potentials

Generating secondary X-rays in a sample



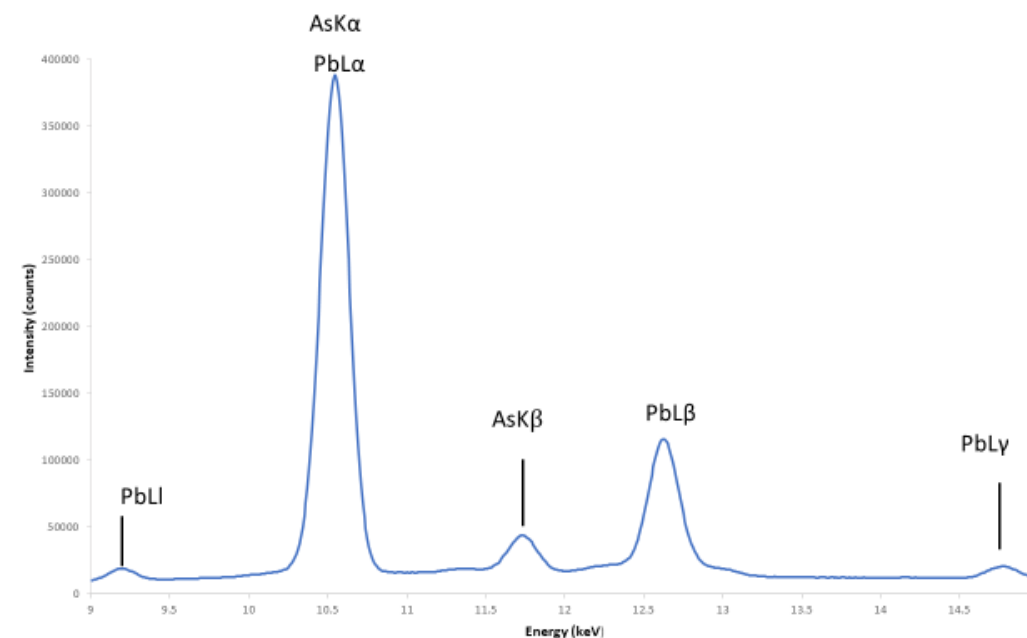
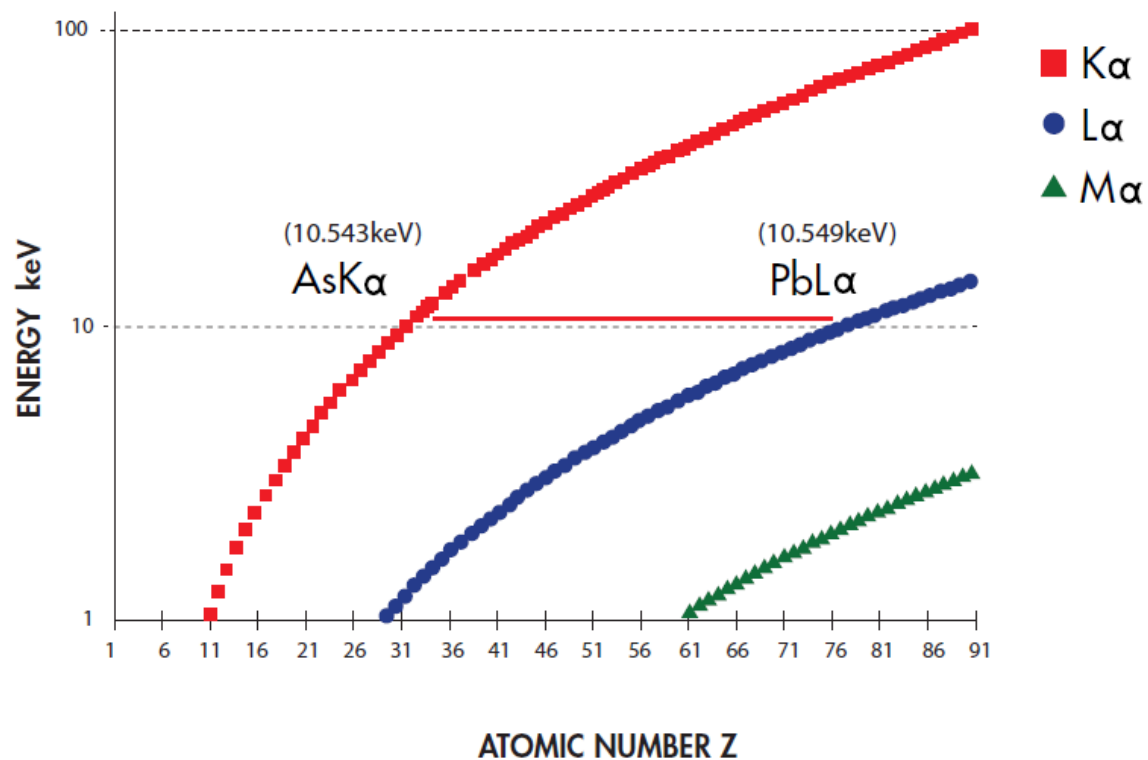
- Similar process to generating primary X-rays
 - X-ray (photon) expels the inner core electron to start the process, instead of an electron
 - We now have a complex array of different atoms instead of a single composition target



Generation of X-rays & Excitation Potentials

X-ray emission energy vs Atomic number (Z)

Moseley's Law - Relationship in practical terms





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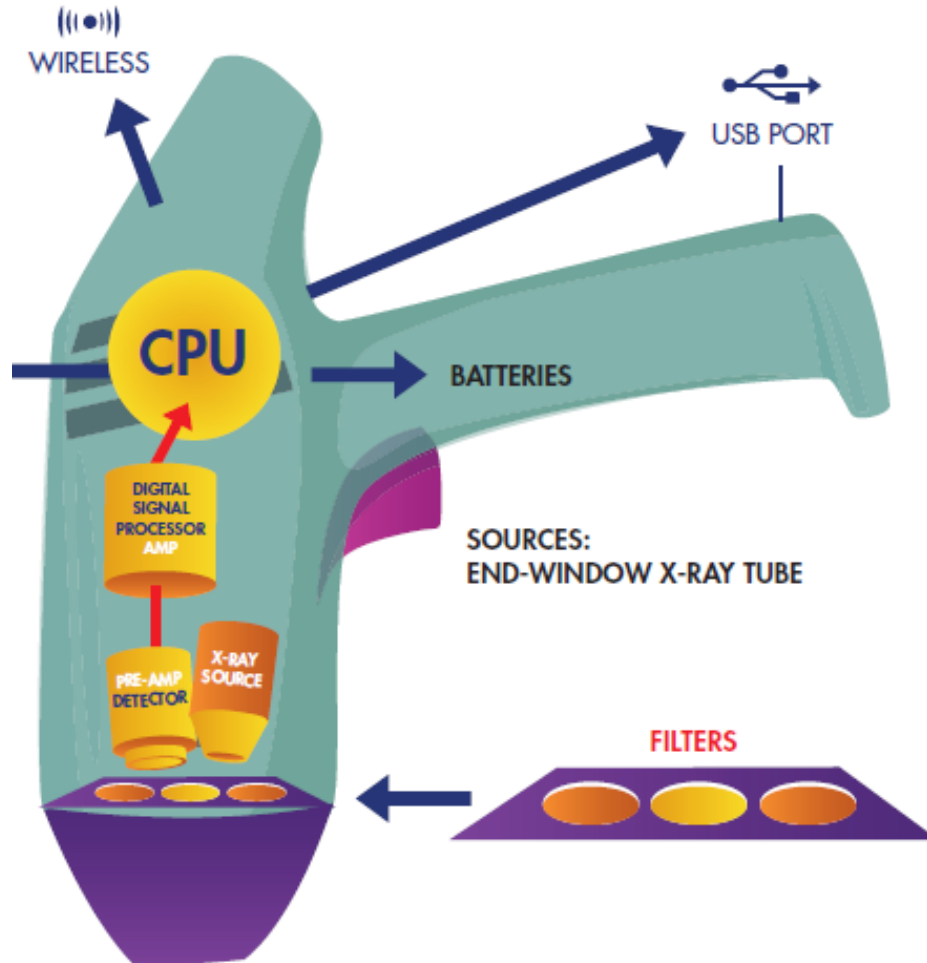
Back to basics in Handheld XRF

Modulation of primary X-rays – volts, current, filters

Modulation of X-rays – volts, current, filters

What controls the generation of characteristic X-rays from the sample?

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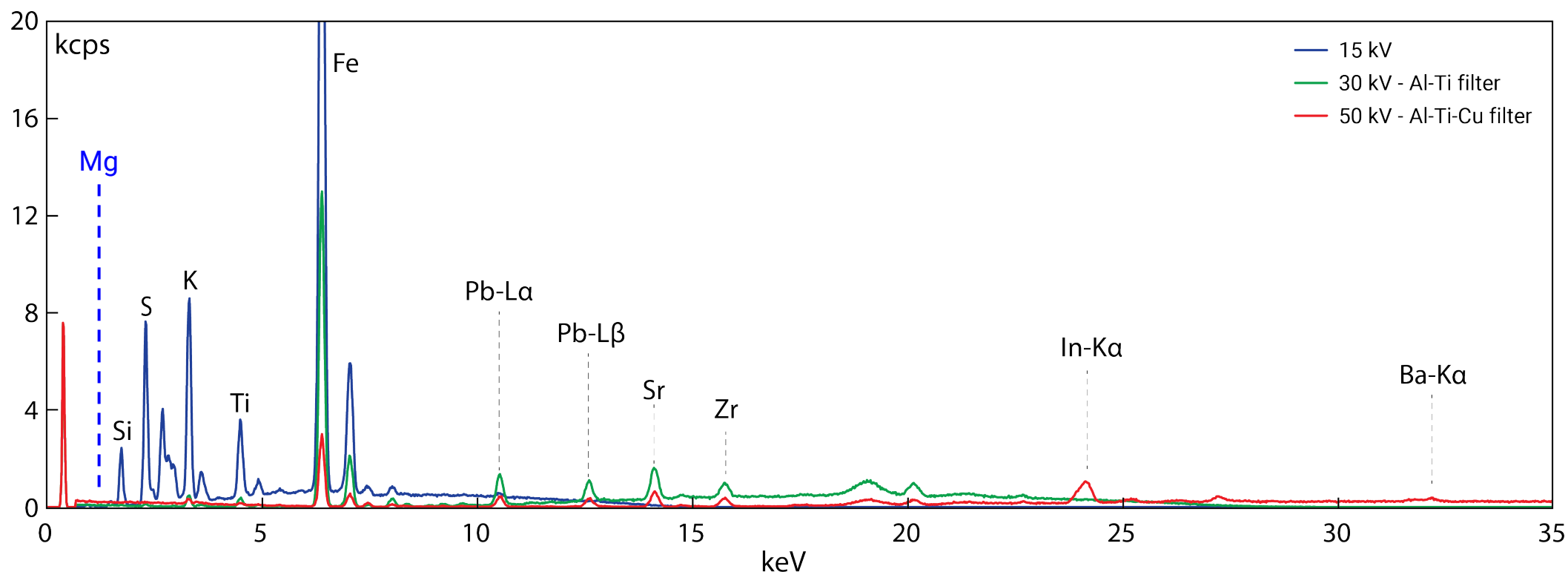
- Three ways to modulate the primary X-ray source to influence the response from the sample (as observed in the spectrum)
 - Voltage (kV)
 - Current (μA)
 - Primary beam filters

Modulation of X-rays – volts, current, filters

Range of measurable X-rays (HH-XRF)

Typical

- Mg-K α (1.254 keV)
 - medium, detector dependent
- Ba-K α (32.196 keV)

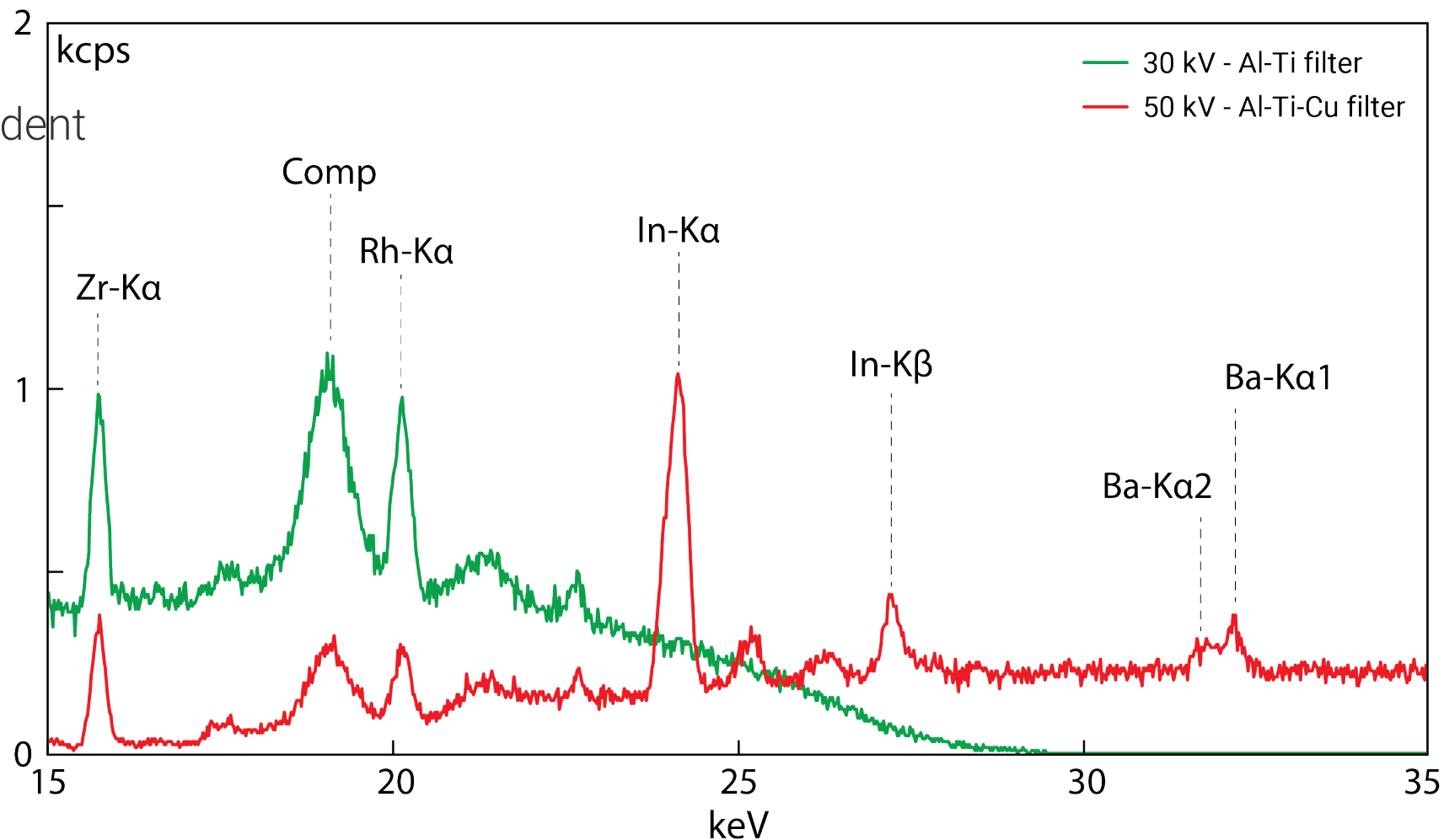


Modulation of X-rays – volts, current, filters

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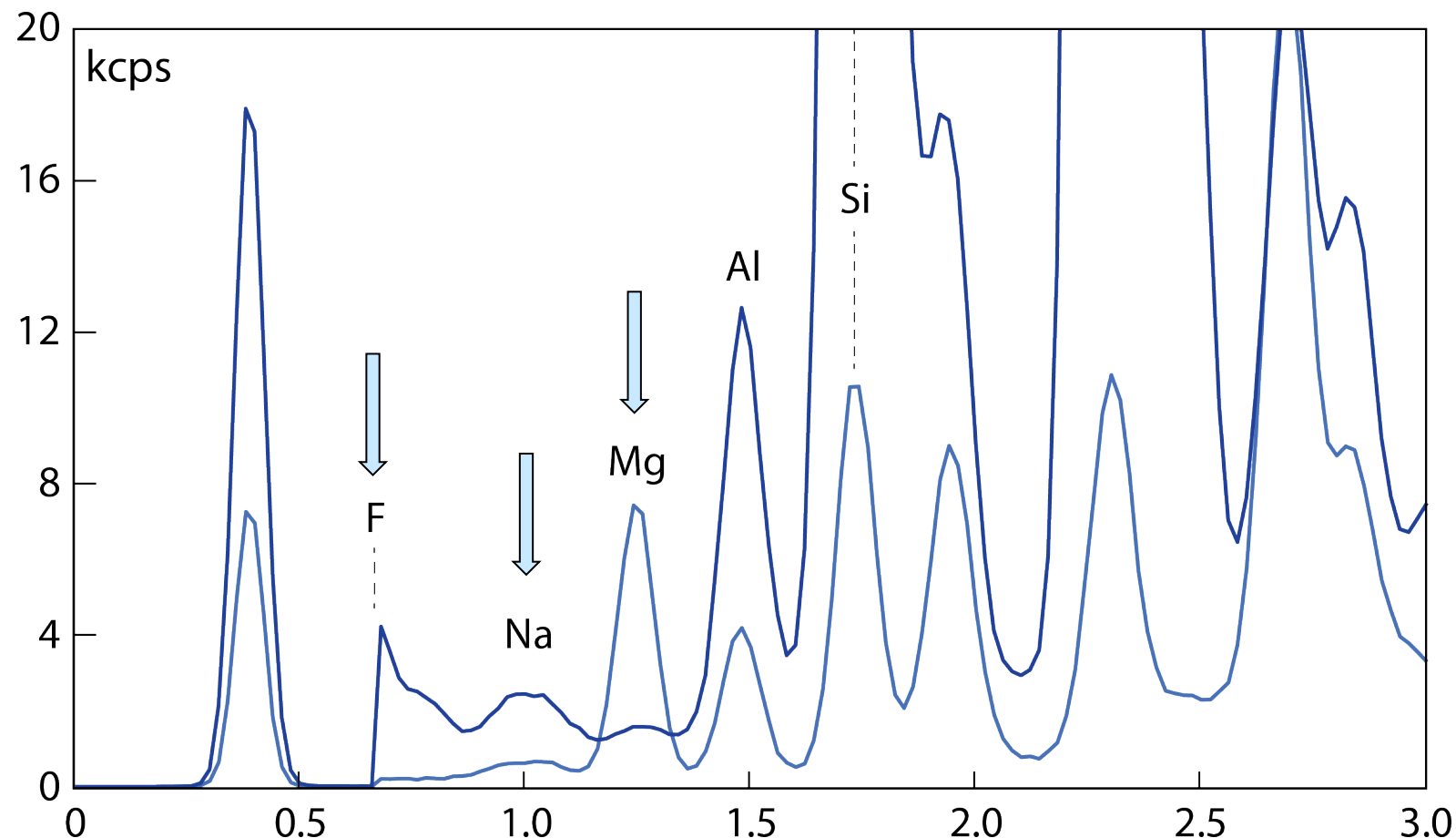


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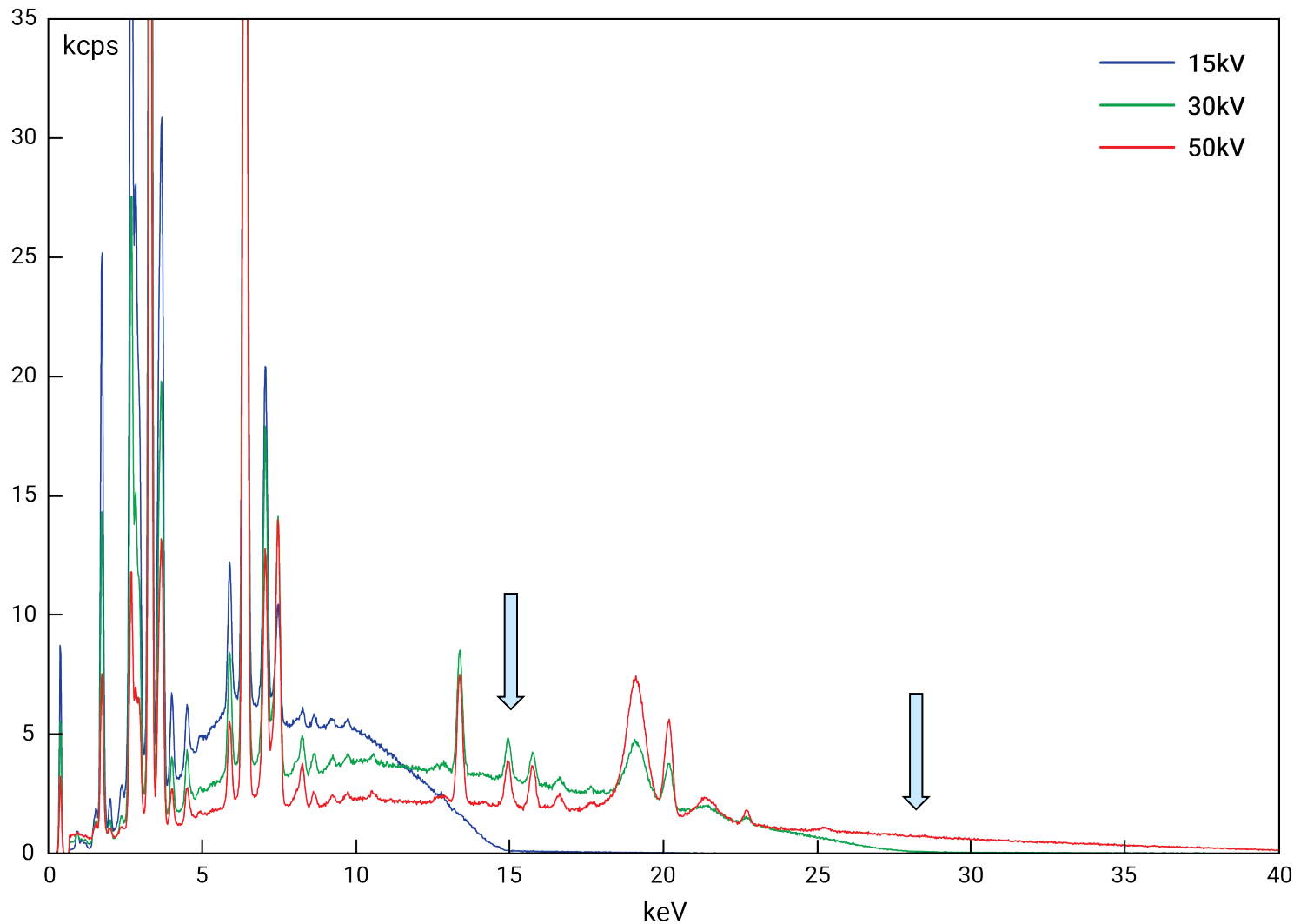
Possible

- Na-Kα (1.040 keV)
- F-Kα (0.677 keV)
 - *Requires enhanced transmission using He-flush +/- graphene detector)*
- The biggest limiting factor is escape depth from sample for these low energies



Modulation of X-rays – volts, current, filters

Using voltage to modulate excitation



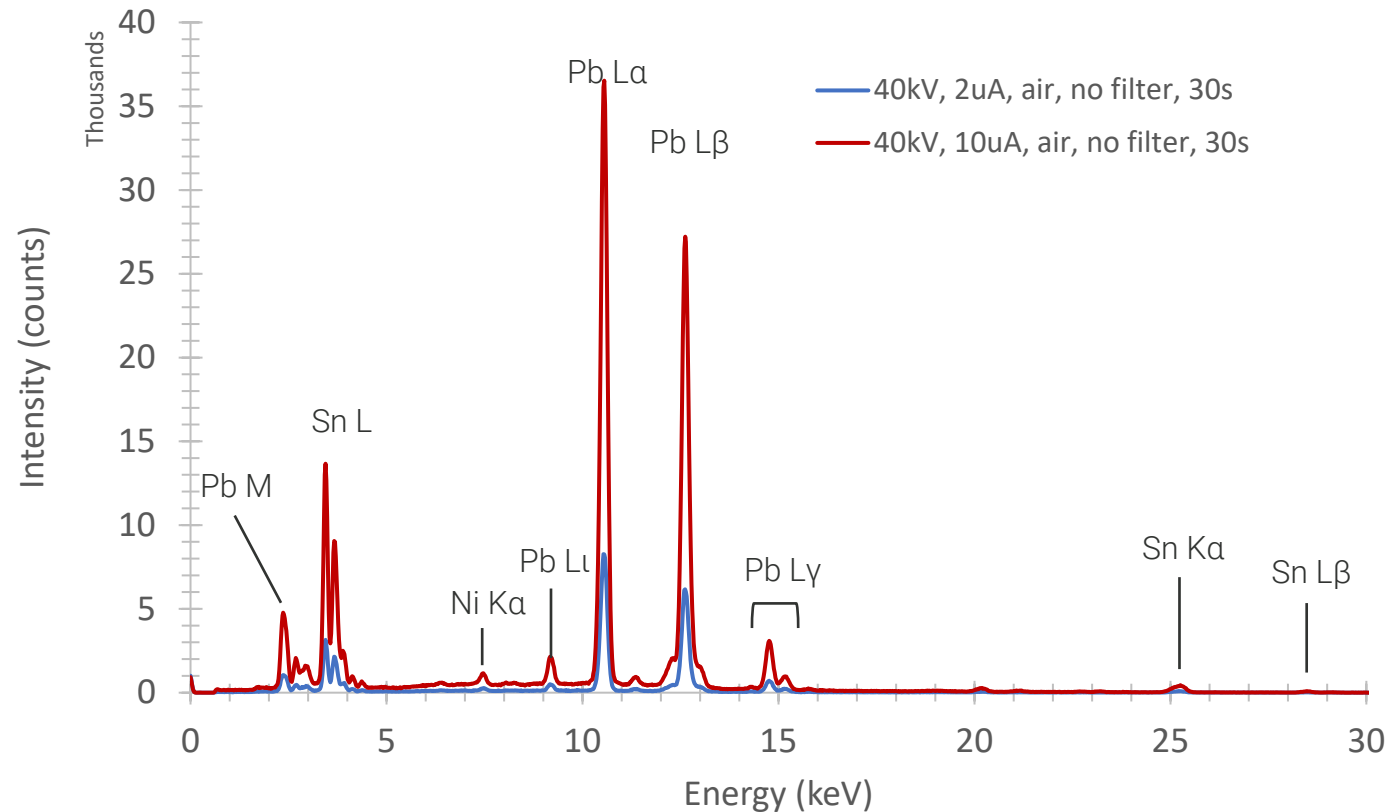
- Voltage applied to the generation of X-rays influences
 - Ability to excite X-rays from different shells in atoms
 - Resulting shape of the X-ray continuum due to Bremsstrahlung

Modulation of X-rays – volts, current, filters

Using current to modulate excitation

- Current influences the number of photons that are generated from the sample
 - Higher current = higher counts
- Maximum current will depend on X-ray tube wattage
- Current used balances highest count rate without overwhelming the detector (deadtime)

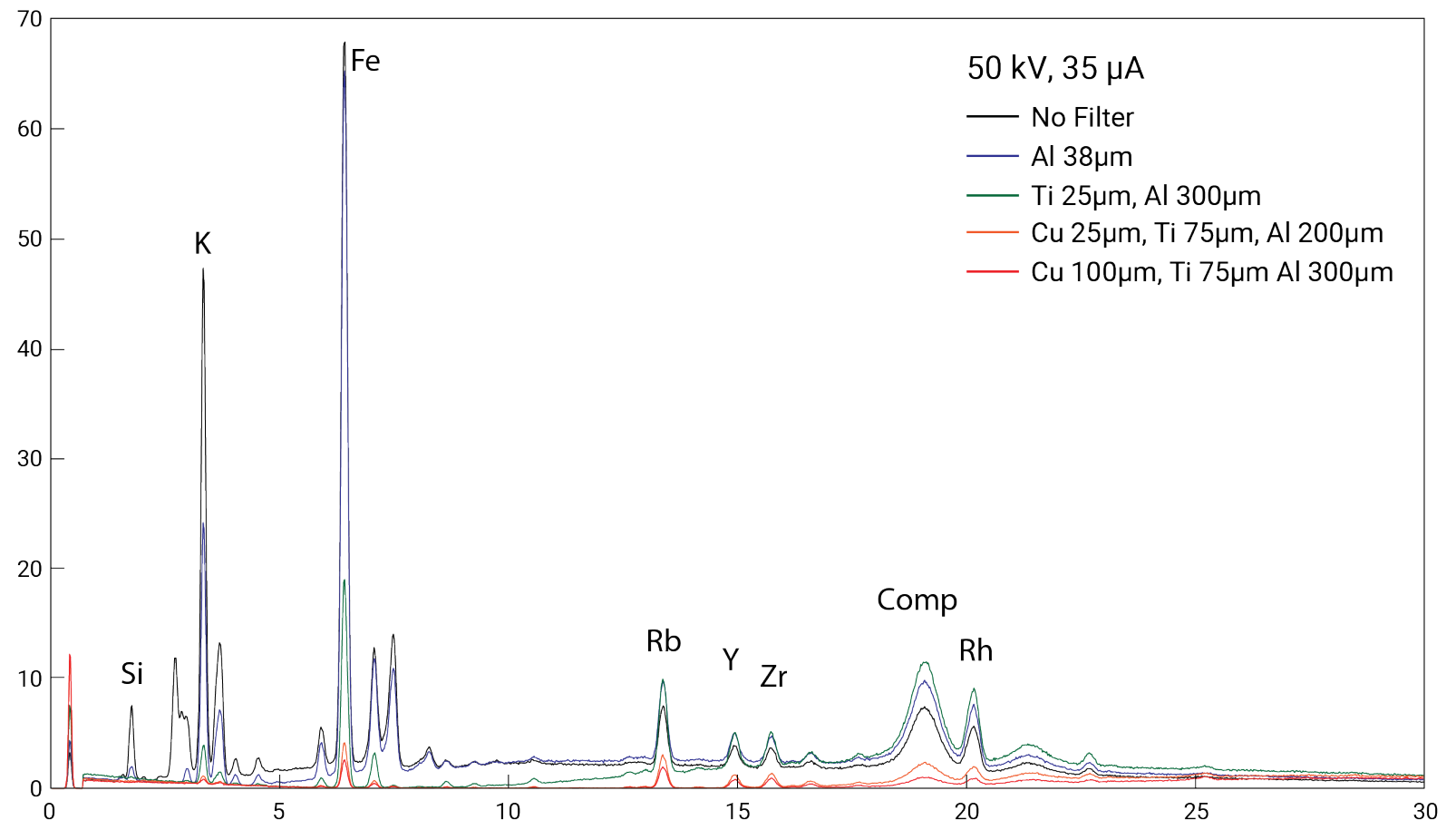
Lead-tin solder analysis
The effect of current on spectra (Bruker Tracer III-SD, Rh tube)



Modulation of X-rays – volts, current, filters

Using filters to modulate excitation

- Filters – thin metal sheets placed between the primary beam and the sample
 - Influence the energy "shape" of the primary beam by attenuating lower energy X-rays
- Used to reduce
 - Local background
 - Diffraction peaks and other artifacts
 - Saturation of detector





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HANDHELD XRF IN CULTURAL HERITAGE STUDIES

Back to basics in Handheld XRF

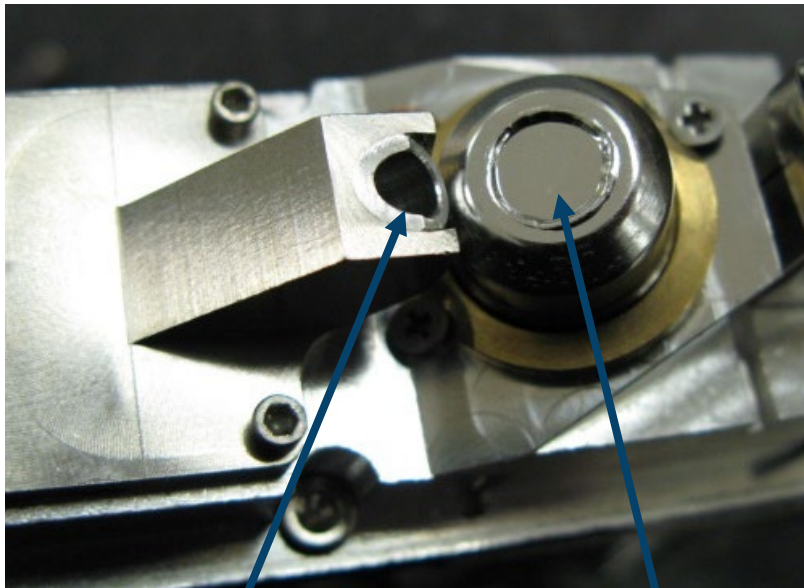
The spectrum: Detection of X-rays, artifacts

The spectrum: Detection of X-rays, artifacts

The X-ray detector



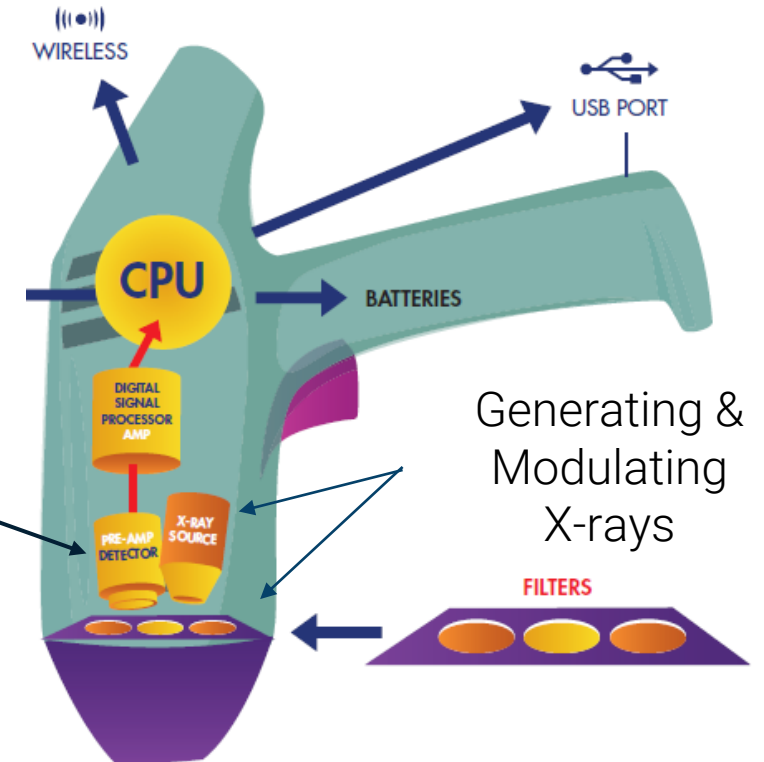
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Tube

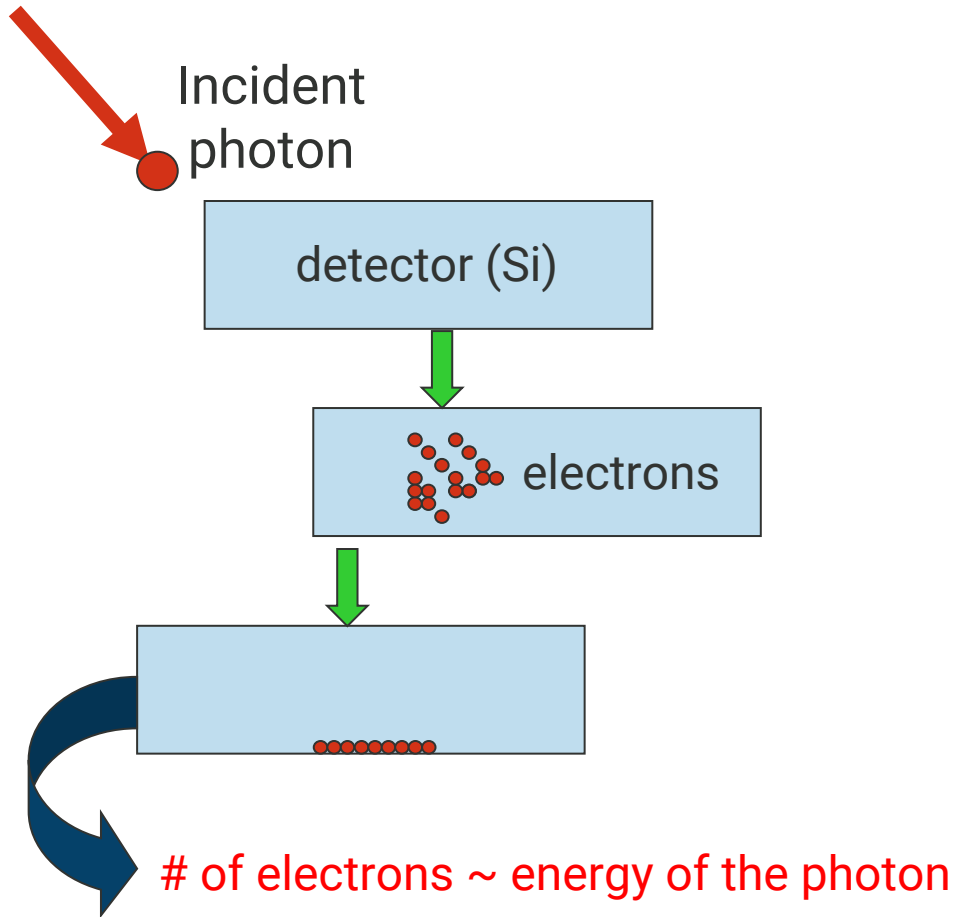
Detector

Detection of X-rays



The spectrum: Detection of X-rays, artifacts

The X-ray detector

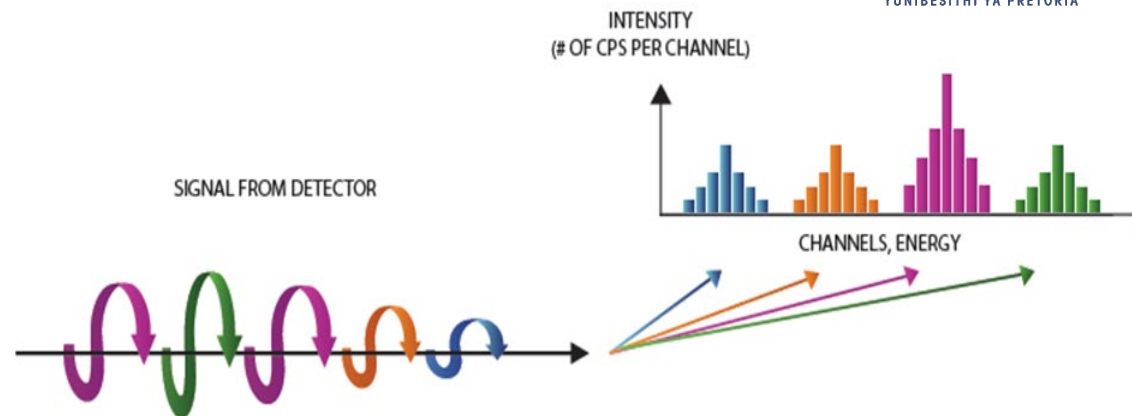
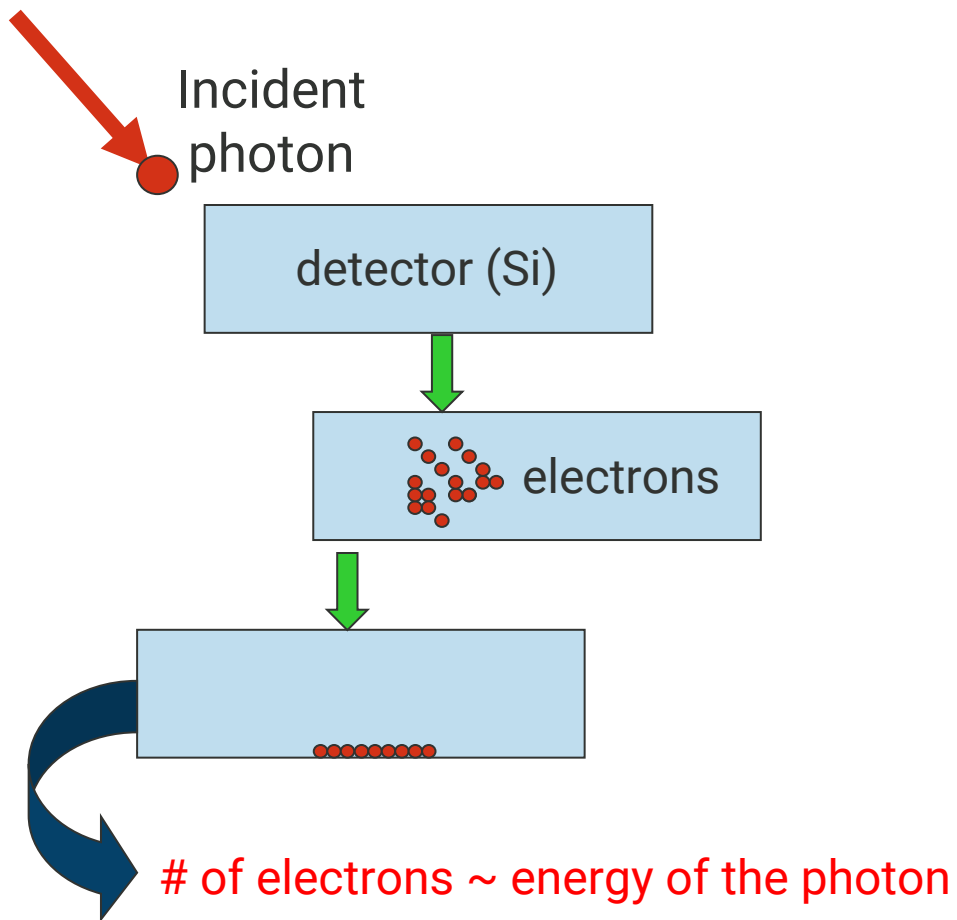


The spectrum: Detection of X-rays, artifacts

The X-ray detector

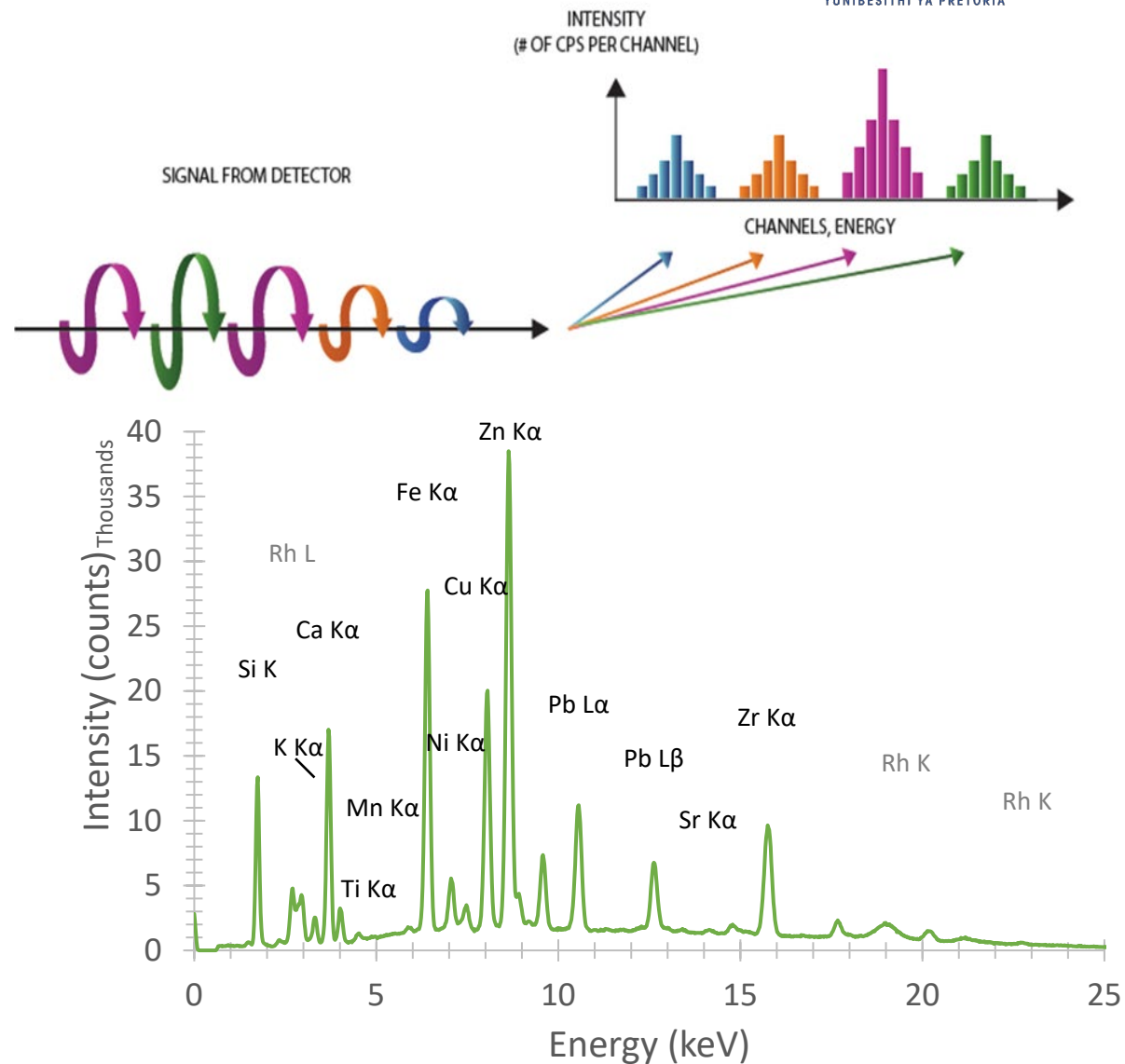
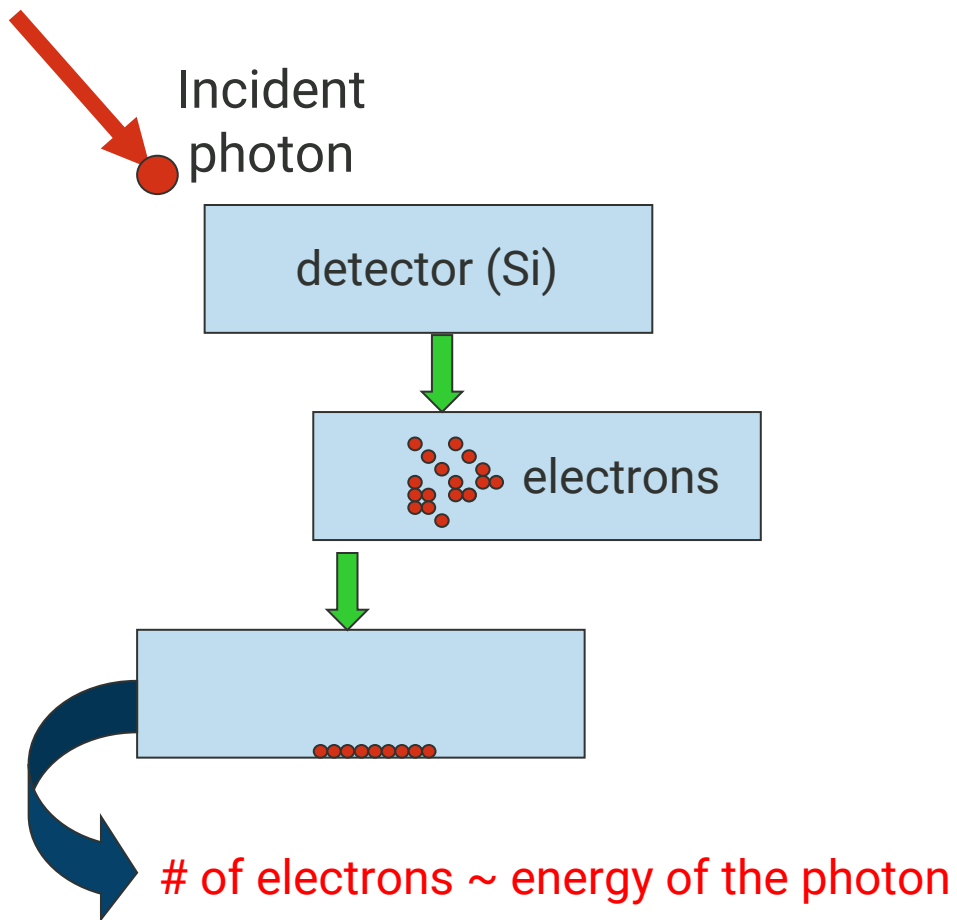


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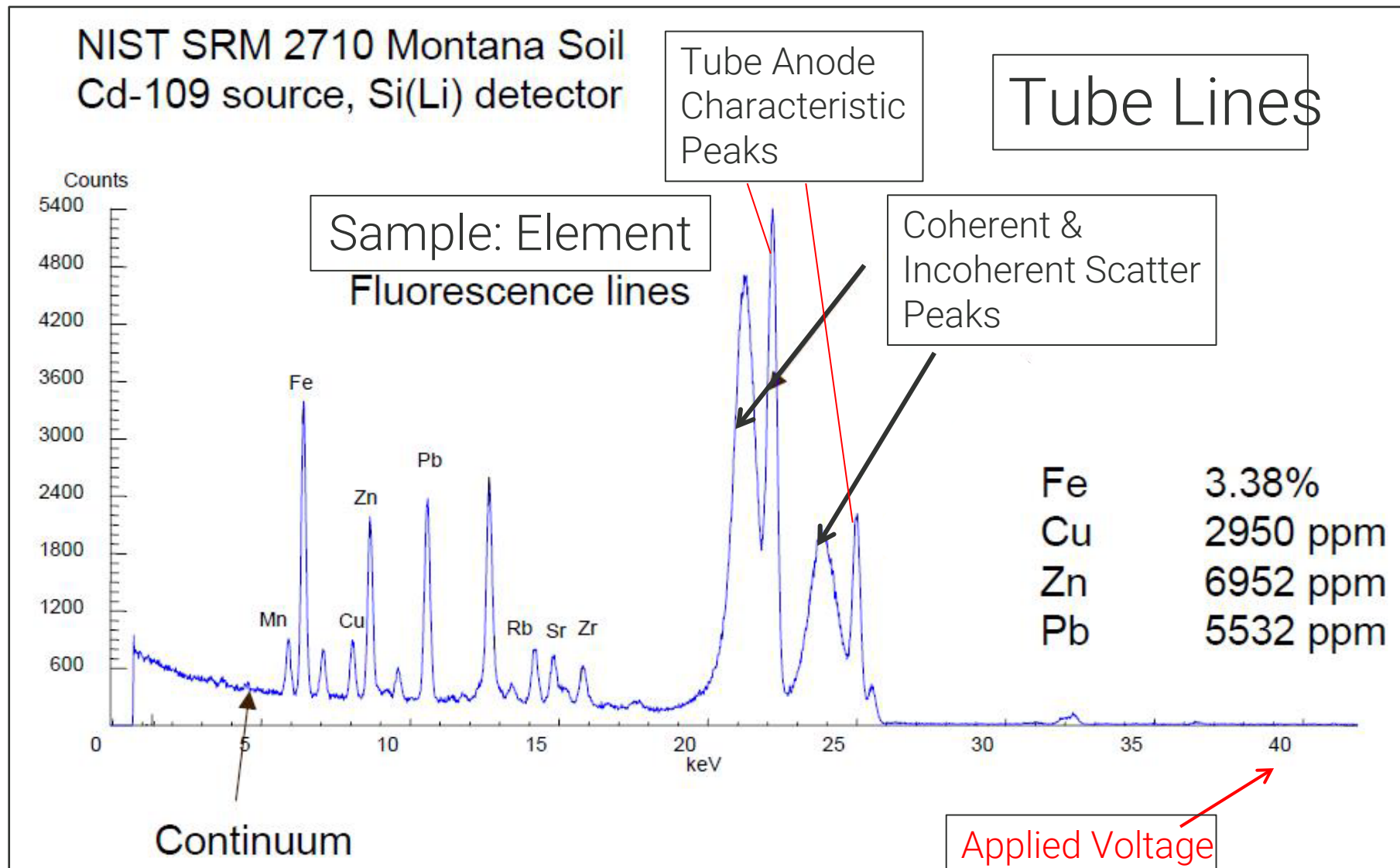
The spectrum: Detection of X-rays, artifacts

The X-ray detector



The spectrum: Detection of X-rays, artifacts

The spectrum we see

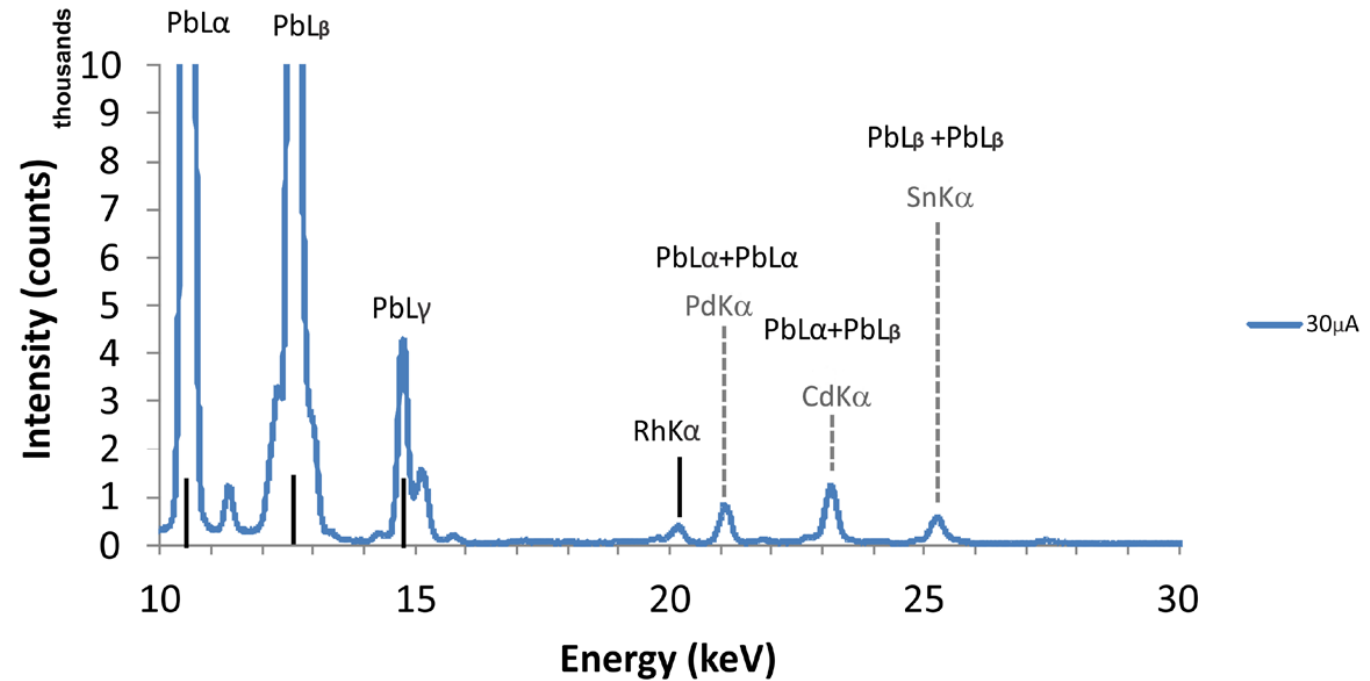


The spectrum: Detection of X-rays, artifacts

Artifacts in detection

Sum peaks

- Occur at very-high count rates when multiple photons reach the detector closer together in time than can be separated by the processing electronics
- e.g., two photons hit the detector simultaneously
- puts out a voltage pulse proportional to the sum of the energy from both photons



Element	Sum Peaks	Energy (keV)	Coincidence Peak (keV)	
Pb L lines	Pb L α + Pb L α	21.10	Pd K α	21.19
	Pb L α + Pb L β	23.16	Cd K α	23.16
	Pb L β + Pb L β	25.27	Sn K α	26.25

The spectrum: Detection of X-rays, artifacts

Artifacts in detection

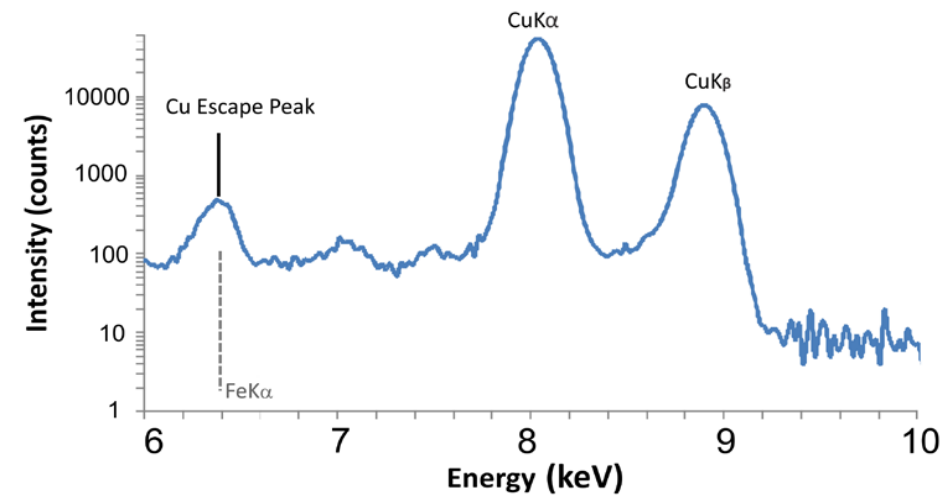
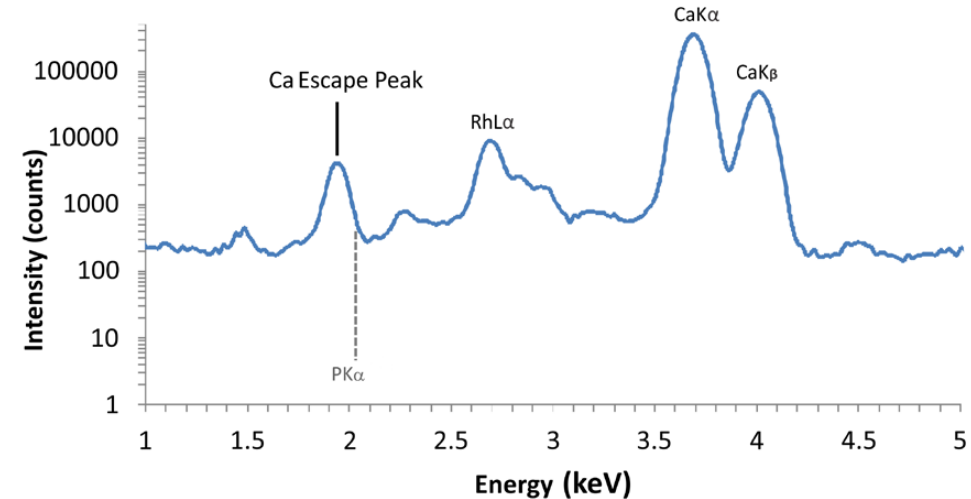


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Escape peaks

- When fluorescent X-rays from a sample hit the Si atoms in the detector
 - Si K α fluorescence photon is generated if the X-rays have an energy greater than the excitation potential of Si (1.84 keV)
- Si K α photon may be reabsorbed by the detector. Occasionally, it “escapes.”
 - the energy of the incoming X-ray photon is reduced by the amount of energy lost to the escaped Si K α photon (1.74 keV)
 - Ca K α (3.69 keV) would have an escape peak at $3.69 - 1.74 = 1.95$ keV



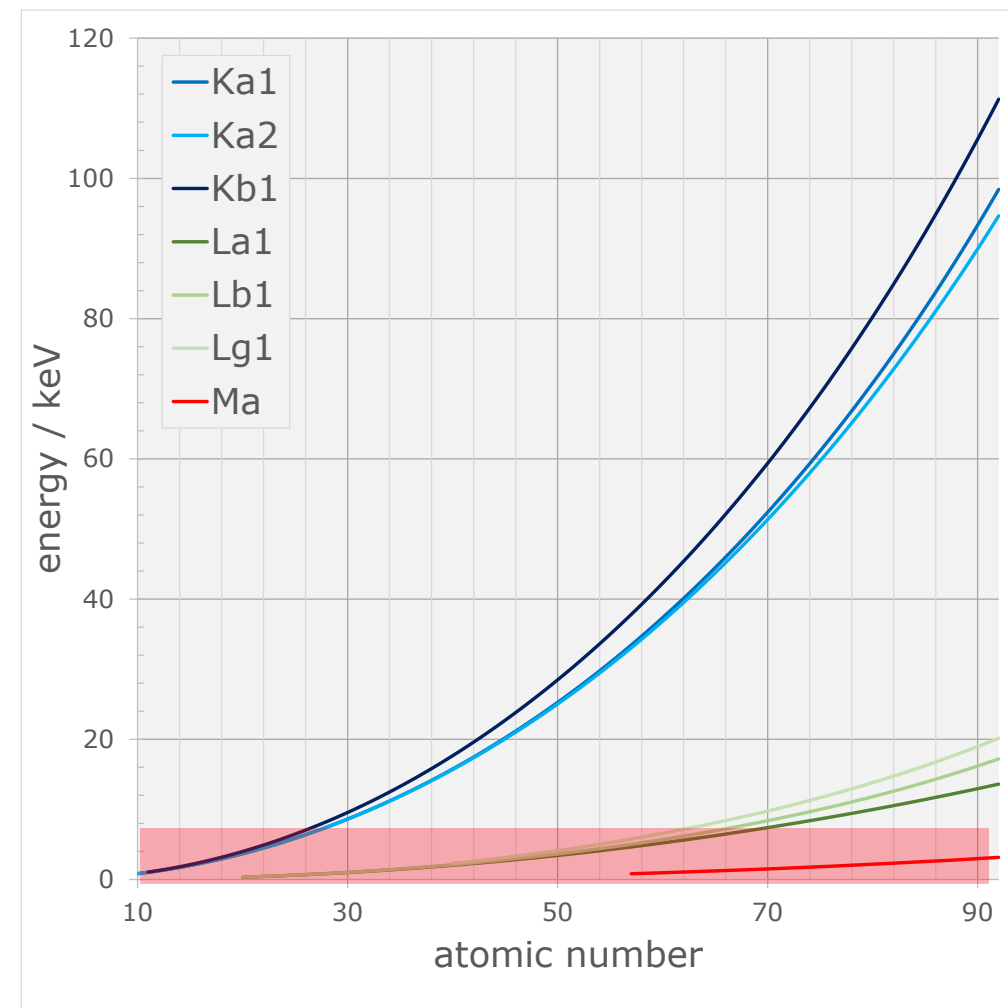
The spectrum: Detection of X-rays, artifacts

Overlapping peaks in the spectrum

- Overlaps are present whether we like it or not!
 - Close spacing of X-ray lines at low energies (Moseley's law)
 - Similar energies from different X-ray lines

	K α	K β		L α	L β		M α	M β
S	2.309	2.465	Mo	2.292	2.394	Pb	2.342	2.444
Ti	4.512	4.933	Ba	4.466	4.828			
V	4.953	5.428						
As	10.543	11.726	Pb	10.551	12.614			

- Strategies
 - Chose appropriate X-ray lines (e.g., K β over K α)
 - Deconvolution





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HANDHELD XRF IN CULTURAL HERITAGE STUDIES

Back to basics in Handheld XRF

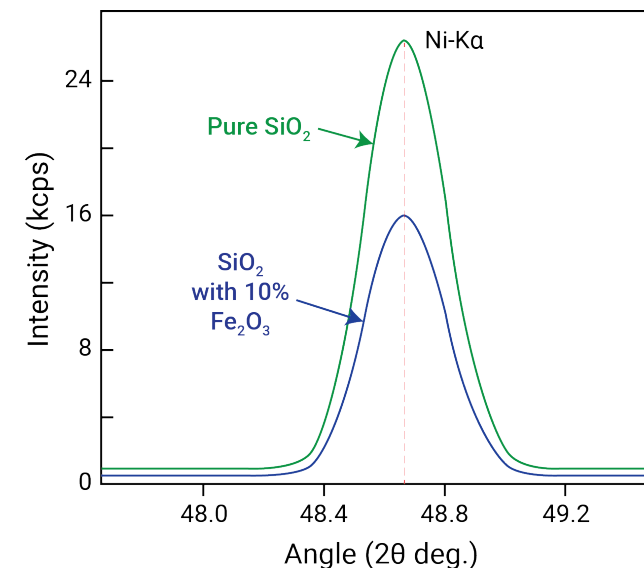
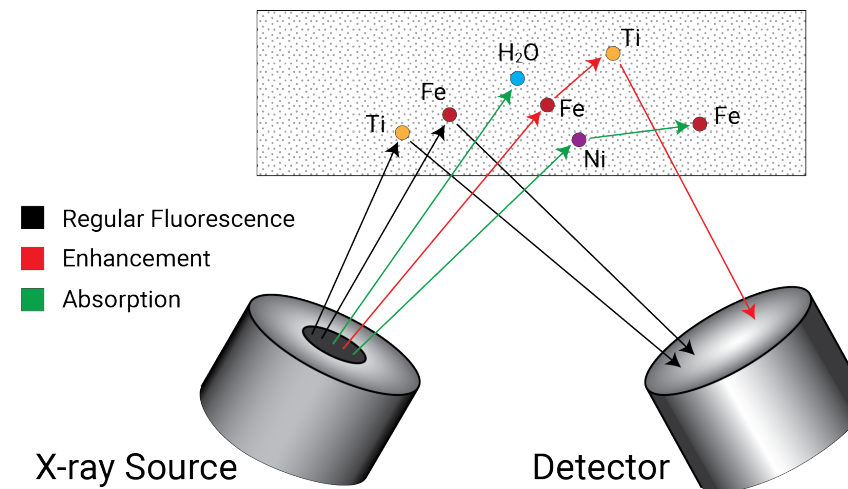
Other contributions to the result

Other contributions to the result

Matrix effects

- Relative intensity across the spectrum is subject to the sample matrix
- i.e., what happens to the excited photon while it travels out of the sample to the detector.
 - Inter-element effects
 - Absorption
 - Enhancement

Matrix	Ni-Kα (kcps)	Mass Absorp. Coeff. (λ cm ² /g)
SiO ₂	26.46	37.55
10% Fe ₂ O ₃ 90% SiO ₂	16.05	58.63



Other contributions to the result

Non-matrix factors



- **Sample thickness**
- **Surface topography and the impact of an air-gap**
- **Feature size**
- Moisture
- Sample cups - films
- Chemical effects
- Mineralogical effects

Other contributions to the result

Non-matrix factors



- **Sample thickness**

- Surface topography and the impact of an air-gap
- Feature size
- Moisture
- Sample cups - films
- Chemical effects
- Mineralogical effects

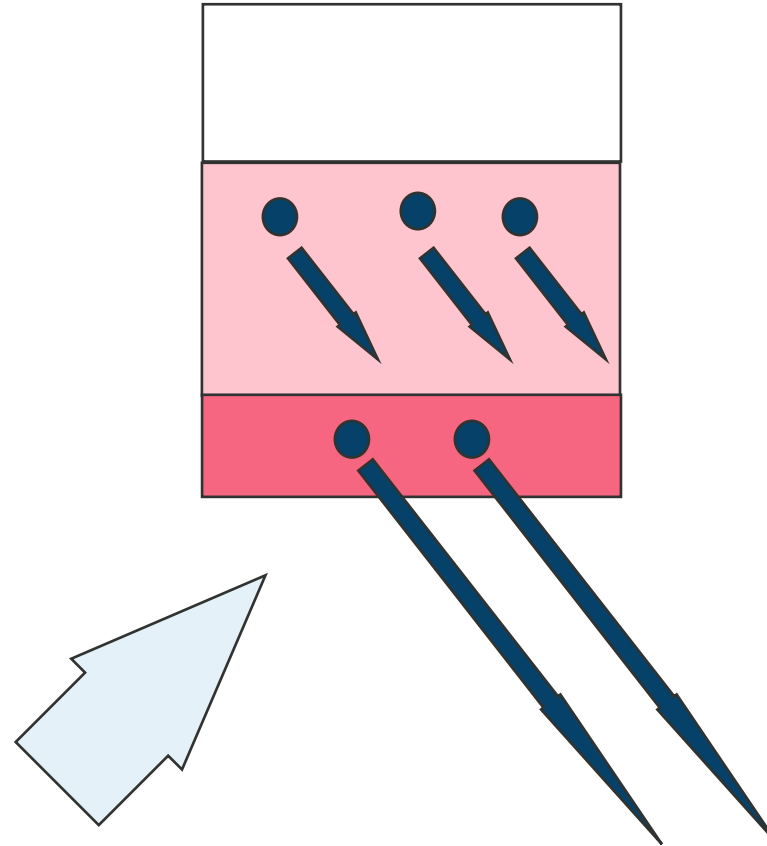
- What is my sampling depth using XRF?
 - Penetration Depth - how deep do primary X-rays transmit into the sample?
 - kV /mA (excitation potential of anode of X-ray tube)
 - density of the sample
 - Escape (Information) Depth - from how deep in the sample is X-ray fluorescence can be detected?
 - energy of the fluorescence radiation
 - composition / density of the sample
 - *depth within the sample from which up to 99% (or 99.9%) of the signal for an element from an infinitely thick sample can be obtained*

Other contributions to the result

Non-matrix factors



- **Sample thickness**
- Surface topography and the impact of an air-gap
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For infinitely thick samples

- No excitation of the inner parts of the sample
- An intermediate part can be excited, but the fluorescence radiation is absorbed within the sample
- The detectable fluorescence radiation comes from the parts close to the surface

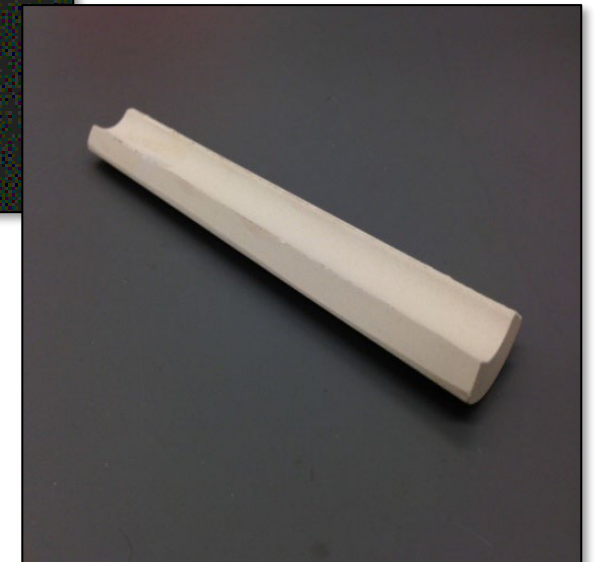
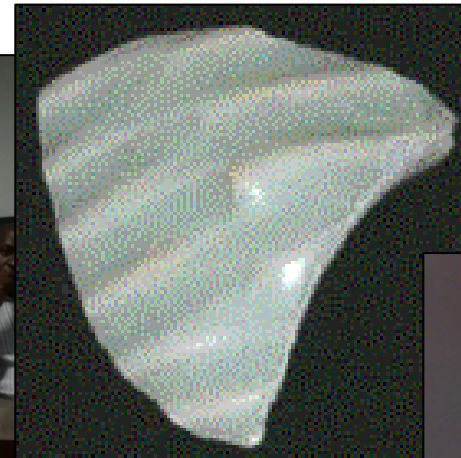
Other contributions to the result

Non-matrix factors



- Sample thickness
- **Surface topography and the impact of an air-gap**
- Feature size
- Moisture
- Sample cups - films
- Chemical effects
- Mineralogical effects

- Air Gap
 - Pronounced topography on the sample surface increases the distance between the sample and detector

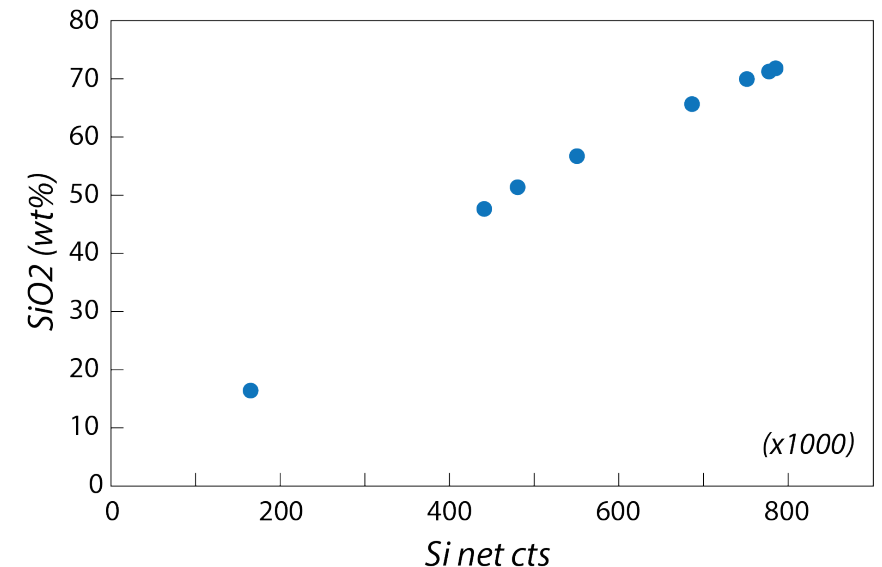
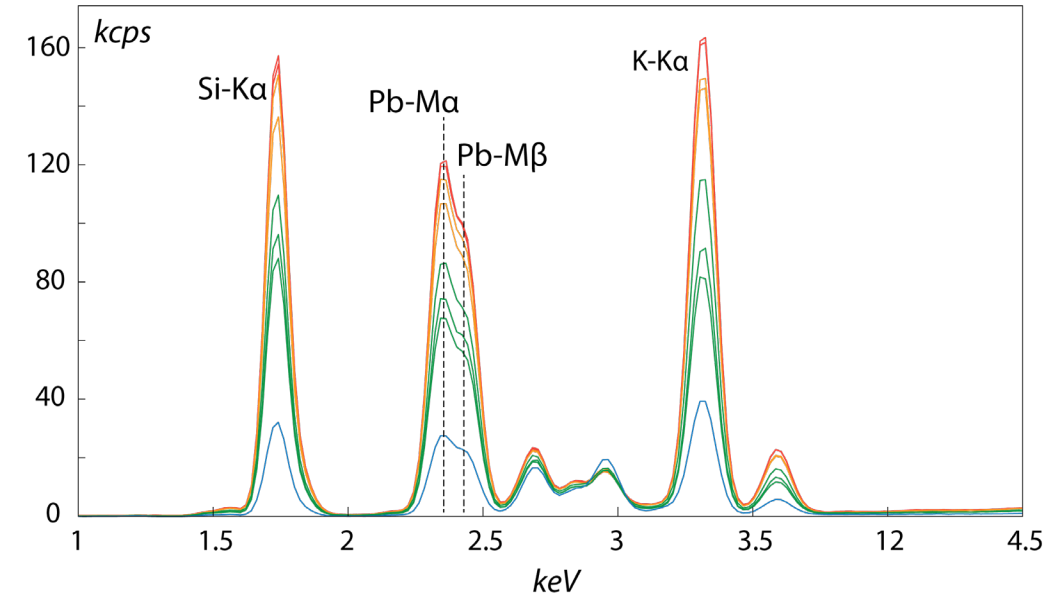


Other contributions to the result

Non-matrix factors

- Sample thickness
- **Surface topography and the impact of an air-gap**
- Feature size
- Moisture
- Sample cups - films
- Chemical effects
- Mineralogical effects

- Air Gap



Other contributions to the result

Non-matrix factors

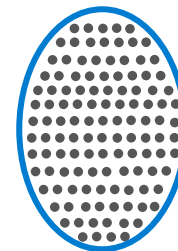
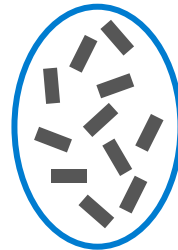
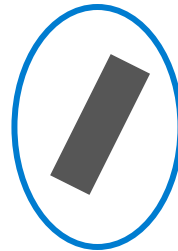


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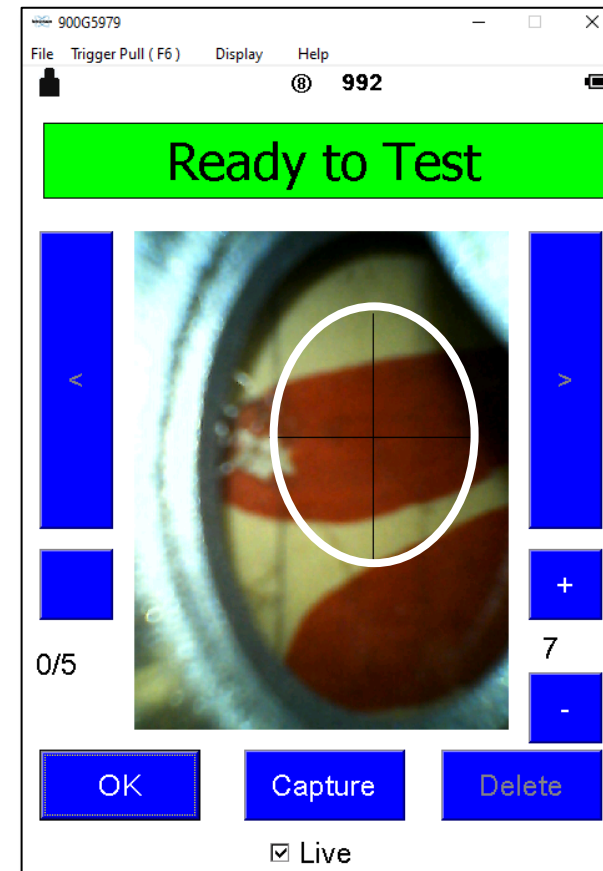


- Sample thickness
- Surface topography and the impact of an air-gap
- **Feature size**
- Moisture
- Sample cups - films
- Chemical effects
- Mineralogical effects

- Beam size vs Analytical target



8mm collimator



Painted wood fishing lure
(Indiana State Museum Collection)

Handheld XRF in Cultural Heritage Studies Summary

- For handheld XRF - a point and shoot technique will not result in meaningful information
- Measurements taken without thought to the underlying technique will result in weak results, or at worst, erroneous data
- Understanding of the instrument, how it operates, and how it interacts with the sample creates a powerful approach to investigation of cultural heritage materials



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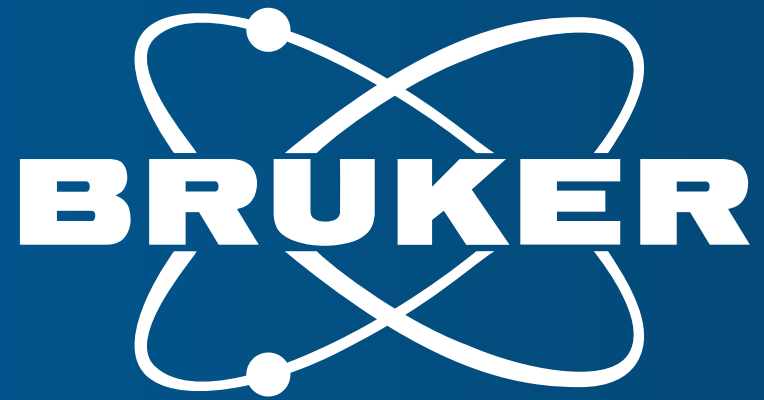




To learn more about practical usage of HH-XRF in cultural heritage work

Download from the Getty website here:

<https://gty.art/3a7Mjaa>



Innovation with Integrity