

# Quantification of Steels and Alloys using a dual source multidetector system

## Part I: XRF-EDS and SEM-EDS analysis



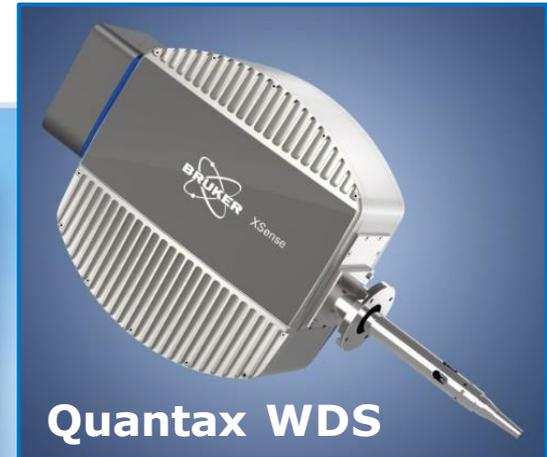
Bruker Nano Analytics, Berlin, Germany  
Webinar, March 23<sup>th</sup>, 2021



Quantax micro-XRF

At. No.	Line S.	EDS Mass [%] Norm.	XRF Mass [%] Norm.	Comb. Mass [%] Norm.	Certified Val. /M%
8	K-Series	45.71	45.58	46.03	46.82
11	K-Series	10.54	10.32	10.61	10.68
12	K-Series	2.32	2.27	2.33	2.22
13	K-Series	1.34	0.89	0.89	33.70
14	K-Series	33.94	34.99	34.18	0.11
16	K-Series	0.16	0.12	0.12	0.34
19	K-Series	0.37	0.35	0.35	5.08
20	K-Series	5.30	5.34	5.34	0.01
22	K-Series	0.00	0.01	0.01	0.03
26	K-Series	0.00	0.03	0.03	0.04
33	K-Series	0.28	0.06	0.06	-
38	L-Series	0.04	0.04	0.04	-

XRF: Oxygen quantification by stoichiometry



Quantax WDS



Quantax EDS

# Presenters



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Bruker Nano Analytics, Berlin, Germany

## Why using different analytical techniques for steel analysis?

- Quantification of Steels and Alloys can be problematic due to the various elements of interest ranging from very low Z-numbers up to high Z-elements and the range of concentrations present from majors through to traces.
- **E-beam excitation / EDS detection:**
  - Very good low Z-element sensitivity / spatial resolution
  - Relatively high spectral background / limiting in sensitivity
- **X-ray excitation / EDS detection:**
  - Low spectral background / high sensitivity for higher Z-elements
  - Limitation in spatial resolution/ light element detection
- **E-beam excitation / WDS detection:**
  - High spectral resolution; improved P/B ratio compared to e-beam / EDS
  - Sequential analysis (one element at a time)

# Overview:

## Quantification of Steels and Alloys



**Part I** of this webinar series will focus on the dual-beam sources (electron and X-ray source), and how they interact with the samples of interest to generate X-rays which are identified and quantified using EDS.

**Part II** will then compare these results with the measurements using the WDS collected on the same system.

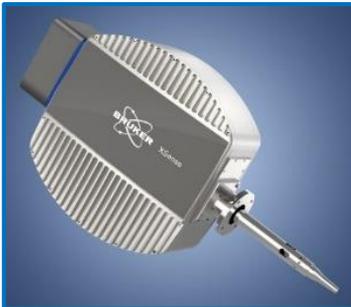
- Terminology
- MicroXRF Introduction
- Quantification Methodologies
- Application to Steels and Alloys
- Summary and Conclusion

# Introduction

## Terminology



**SEM-EDS:** Analysis based on the sample interaction with an electron beam source from the SEM and the resultant X-rays that are detected using an EDS (simultaneous element detection)

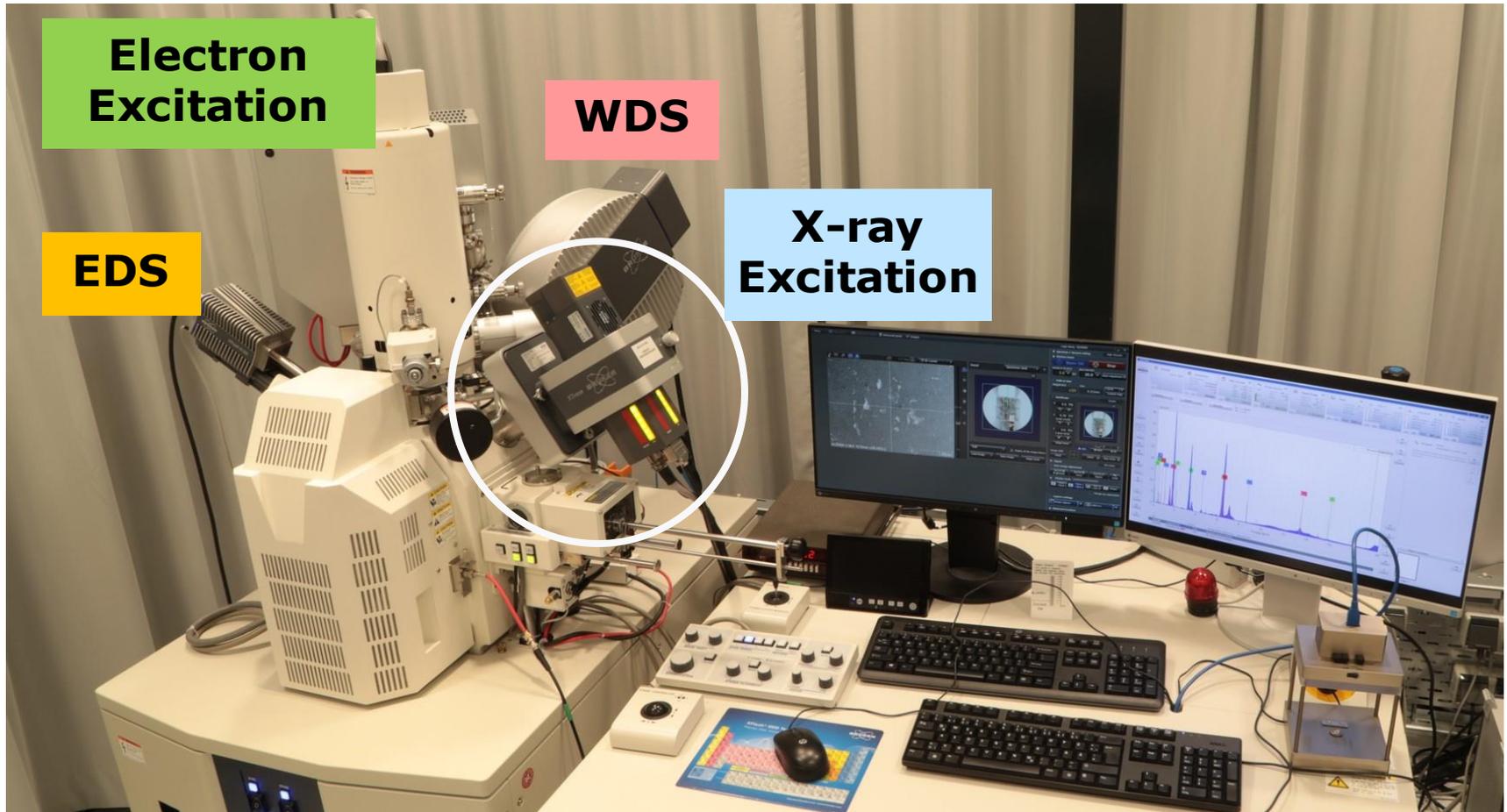


**SEM-WDS:** Analysis based on the sample interaction with an electron beam source from the SEM and the resultant X-rays that are detected using a WDS (sequential element detection)



**SEM-XRF-EDS: MicroXRF on SEM (XTrace):** Analysis based on the sample interaction with an X-ray beam source from the Micro XRF attached to the SEM and the resultant X-rays that are detected using an EDS (simultaneous element detection)

# Excitation: MicroXRF and Electron Detectors: EDS and WDS

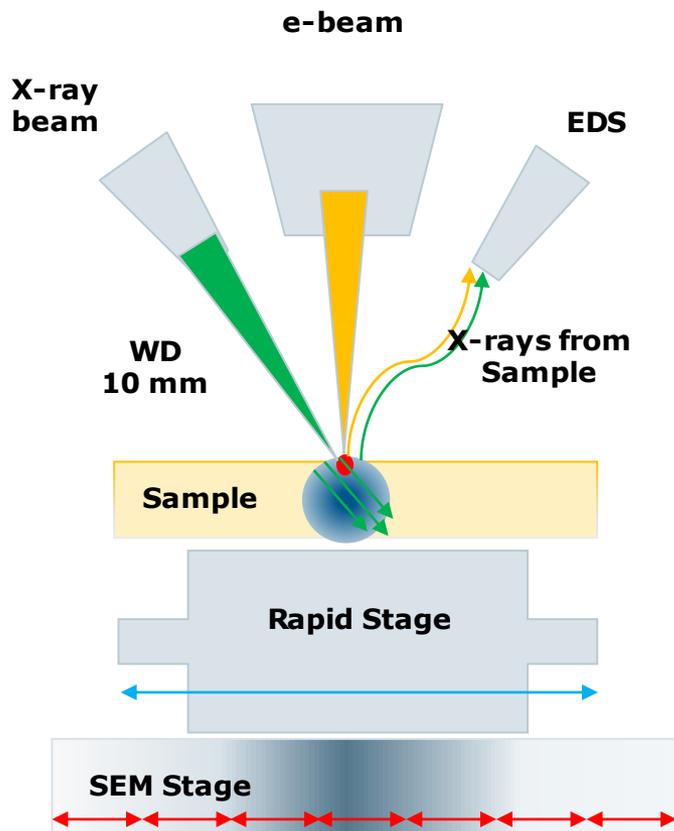


# Introduction

## What is Micro-XRF on the SEM?



### Enhanced SEM



### 2 Excitation Sources:

- Electron Beam (e-beam)
- Micro-XRF (X-ray beam)

### 1 Detector:

- Energy Dispersive Spectrometer (EDS)

### 2 Stages:

- SEM Stage
- Rapid Stage



# Micro-XRF and Electron Excitation Analysis

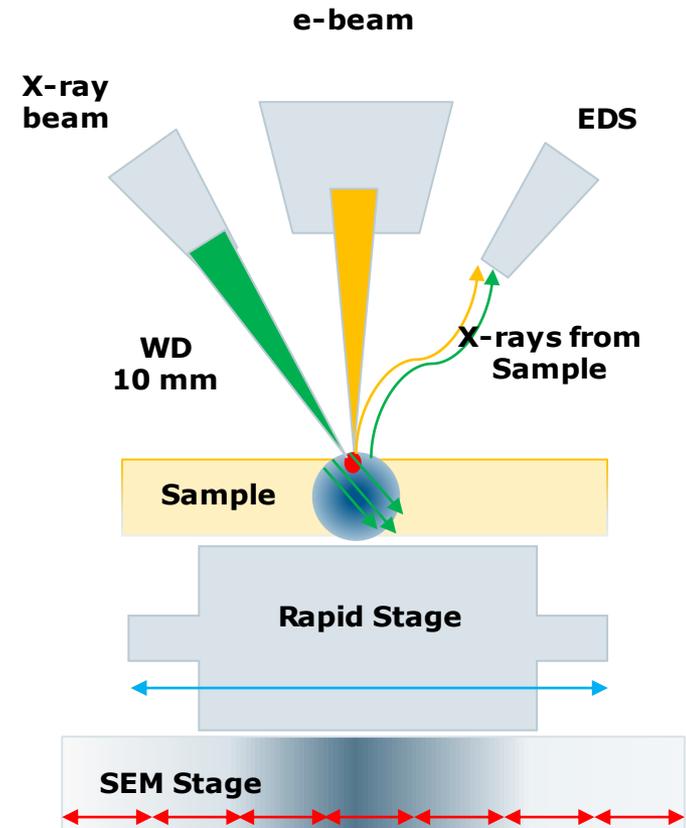


Excitation with either electrons or X-rays generate fluorescence radiation of the irradiated material.

Detection is normally performed with energy dispersive spectrometers (EDS), independent of the excitation source. Signal collection and spectral presentation is identical, but quantification is different.

Important differences:

- Spot Size
- Information depth
- Elemental Range
- Limits of detection
- Spectral Background
- Sample Handling



# Analytical Parameters and Conditions

## SEM-EDS vs SEM-WDS vs SEM-XRF

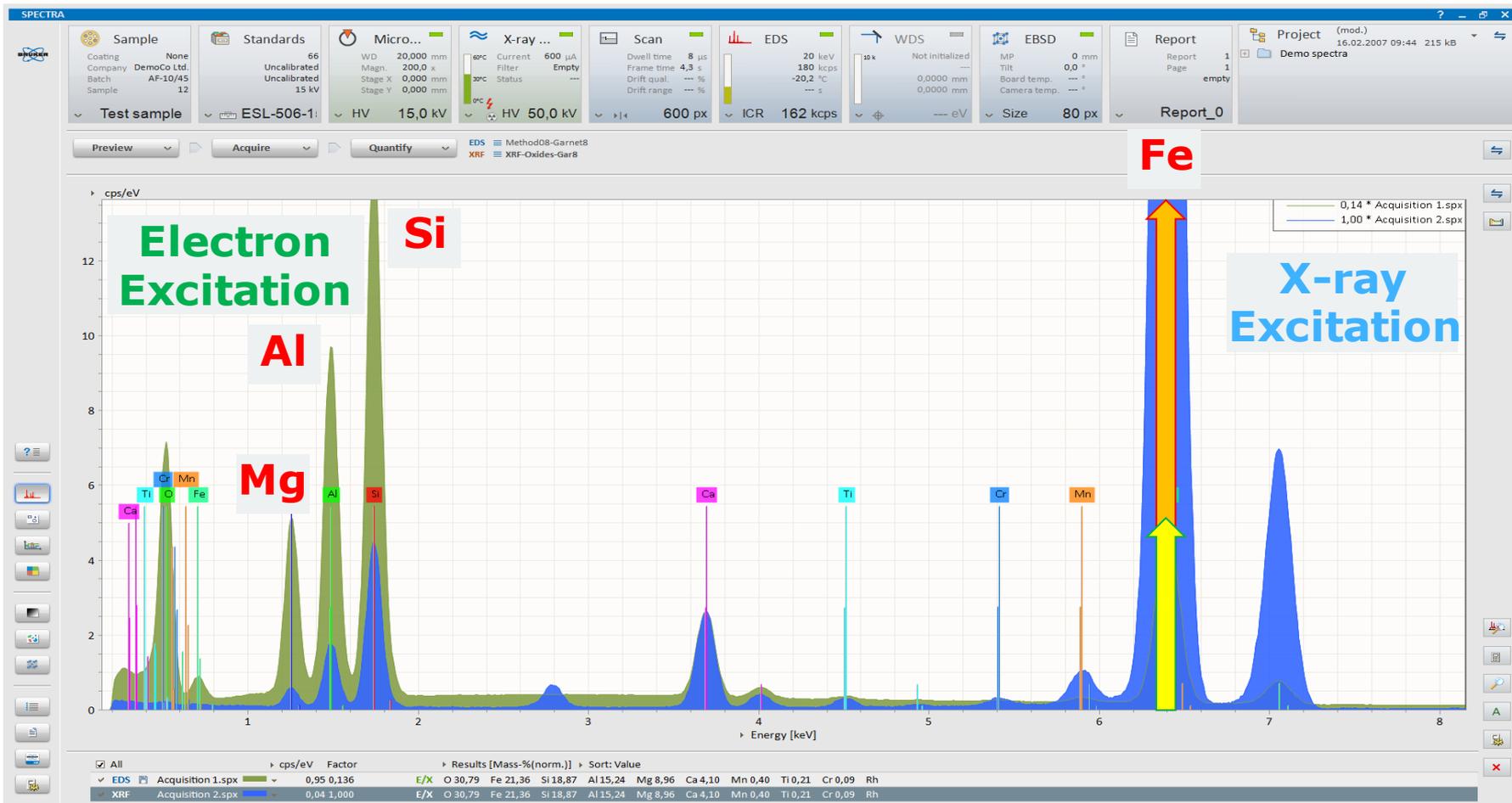


Parameter	EDS: E-beam (SEM-EDS)	WDS: E-beam (SEM-WDS)	EDS: Micro-XRF (SEM-XRF-EDS)
<b>Analyzed Volume</b>	Ø: few µm Information depth: µm; (depending primarily on electron energy)	Ø: few µm Information depth: µm; (depending primarily on electron energy)	Ø: 15-30 µm Information depth: µm to mm; (depending on analysed element and matrix)
<b>Detectable Elements</b>	Atomic number $Z \geq 4$ (beryllium)	Atomic number $Z \geq 4$ (beryllium)	Atomic number $Z \geq 6$ (carbon)
<b>Energy range</b>	K- L -M - Lines ( up to 20 keV)	70 eV – 3.6 keV (L- M- Lines)	K- L -M - Lines ( up to 40 keV)
<b>Concentration Range</b>	Down to 1000 ppm	Down to 100 ppm	Down to 10 ppm
<b>Quantification</b>	Standard less and Standard based	Standard based	Standard less and standard based
<b>Data collection</b>	Simultaneously	Sequentially	Simultaneously
<b>Sample Preparation</b>	Sample needs to be electrically conductive (commonly carbon-coated), polishing required	Sample needs to be electrically conductive (commonly carbon-coated), polishing required	Electrical Conductivity not required, samples doesn't need to be polished
<b>Sample Stress</b>	Heating due to absorbed electrons	Heating due to absorbed electrons	Minimal
<b>Typical SEM beam current</b>	Variable	Variable > 10 nA	N/A

# Energy Dispersive Spectra Comparison: Electron vs X-ray Excitation



**Garnet Spectra: Light element signal intensity higher for electron excitation.**



# Introduction

## Historic and Current Webinars



[www.bruker.com/events/webinars.html](http://www.bruker.com/events/webinars.html)

Filter: EDS, WDS, EBSD, Micro-XRF on SEM

High Speed Mapping Using  
Micro-XRF on SEM



Bruker Nano Analytics, Berlin, Germany  
Webinar, November 6<sup>th</sup>, 2019

Advanced elemental analysis of geological  
samples using QUANTAX WDS for SEM



Bruker Nano Analytics, Berlin, Germany  
Webinar, April 25, 2019

Microanalysis with high spectral resolution:  
the power of QUANTAX WDS for SEM



Bruker Nano Analytics, Berlin, Germany  
Webinar, September 14, 2017

Advancements in Microanalysis with  
micro-XRF on SEM



Bruker Nano Analytics, Berlin, Germany  
Webinar, April 18<sup>th</sup> 2018

WDS

Innovation with Integrity

Element	At. No.	Line S.	WDS Ratio (%) Norm.	WDS Ratio (%) Norm.	WDS Ratio (%) Norm.
Oxygen	8	K-Series	45.71	49.38	48.01
Carbon	6	K-Series	18.54	18.82	18.81
Chlorine	17	K-Series	2.52	2.27	2.01
Magnesium	12	K-Series	2.24	1.89	1.63
Manganese	25	K-Series	33.94	34.99	34.29
Iron	26	K-Series	0.39	0.17	0.14
Nickel	28	K-Series	0.37	0.20	0.20
Copper	29	K-Series	0.36	0.26	0.26
Zinc	30	K-Series	0.80	0.61	0.51
Calcium	20	K-Series	0.28	0.05	0.05
Titanium	22	K-Series	0.28	0.08	0.08
Vanadium	23	K-Series	0.04	0.04	0.04
Chromium	24	K-Series	0.04	0.04	0.04
Strontium	38	L-Series	0.02	0.02	0.02

Micro-XRF on SEM

27.04.2020

# Analysis of Steels and Alloys: Combined Quantification



- If both electron and X-ray excitation are available, the benefits of both methods can be combined. That is:
  1. better light element sensitivity of electron excitation, e.g., from C to Si typically have smaller statistical error and better sensitivity,
  2. better trace element sensitivity for heavy elements of X-ray excitation

*Thus, the results for each quantification method can be calculated separately, and then the results for the elements with better sensitivity and accuracy are used to calculate an improved combined quantification.*

**Note:** the sample has to fulfil the requirements for both excitation types, i.e., the sample needs to be conductive and polished and has to be homogeneous

# Benefit of combined SEM-EDS and micro-XRF



## Quantification of Cast Iron

Elem	Cert	EDS	XRF
C	2.51	2.84	
Al	0.013		0.023
Si	0.829	0.708	1.11
P	0.027		0.016
S	0.01	0.053	0.364
Ti	0.022		0.024
V	0.036		0.034
Cr	0.507	0.520	0.476
Mn	1.94	1.84	1.91
Fe	93.7	94.0	95.7
Ni	0.03		0.023
Cu	0.075		0.063
Zn	0.023		0.013
As	0.016		0.025
Nb	0.051		0.061
Mo	0.071		0.072
Sn	0.011		0.005
Sb	0.025		0.005
W	0.066		0.066
Pb	0.011		0.045
Bi	0.014		0.001

### Problems when using a single method:

- Electron excitation is not good for trace concentrations
- Trace element concentrations possible with X-ray excitation but light elements are not ideal.
- Thus the (normalized) quantification results for the main components could be wrong
- Deconvolution of overlapping peaks, e.g. S due to the overlap of S-K with Mo-L and Pb-M

Haschke, M, and Boehm, S, 2017, Micro-XRF in Scanning Electron Microscopes. In: Advances in Imaging and Electron Physics, Hawkes, P. W. (ed.), vol 199, Academic Press, pp. 1-60

# Benefit of combined SEM-EDS and micro-XRF

## Quantification of Cast Iron



Elem	Cert	EDS	XRF	Comb
C	2.51	2.84		2.84
Al	0.013		0.023	0.022
Si	0.829	0.708	1.11	0.910
P	0.027		0.016	0.044
S	0.01	0.053	0.364	0.036
Ti	0.022		0.024	0.025
V	0.036		0.034	0.033
Cr	0.507	0.520	0.476	0.495
Mn	1.94	1.84	1.91	1.84
Fe	93.7	94.0	95.7	93.0
Ni	0.03		0.023	0.02
Cu	0.075		0.063	0.058
Zn	0.023		0.013	0.013
As	0.016		0.025	0.023
Nb	0.051		0.061	0.057
Mo	0.071		0.072	0.067
Sn	0.011		0.005	0.005
Sb	0.025		0.005	0.005
W	0.066		0.066	0.077
Pb	0.011		0.045	0.042
Bi	0.014		0.001	0.002

### Solution:

In the combined analysis the concentration of the heavy elements (Mo-K, Pb-L) determined with X-ray excitation can be used to calculate their peak intensity in the range of S-K radiation and then only the difference is caused by S

→ Combined analysis allows the determination of mass fraction of all elements with a better accuracy than the quant only from a single excitation.

Haschke, M, and Boehm, S, 2017, Micro-XRF in Scanning Electron Microscopes. In: Advances in Imaging and Electron Physics, Hawkes, P. W. (ed.), vol 199, Academic Press, pp. 1-60

# Basics of FP Quantification

## Sherman's Equation



Quantitative X-ray Fluorescence analysis is based on sample properties and the physical processes of known probability.

These are known as the **Fundamental Parameters (FP)**.

The Sherman Equation is the basis for all XRF FP quantification

$$I_{fl,i} = K_i \cdot \int_{E_{abs,i}}^{E_{max}} \int_{x=0}^D I_0(E) \cdot e^{\frac{-\mu_S(E) \cdot \rho_S \cdot x}{\sin(\varphi_{in})}} \cdot C_i \cdot \tau_{i,E} \cdot Q_i(E, E_{fl}) \cdot e^{\frac{-\mu_S(E_{fl}) \cdot \rho_S \cdot x}{\sin(\varphi_{out})}} \cdot \frac{\Omega}{4\pi} \cdot \varepsilon^D dx dE$$

Excitation spectrum (as it reaches the sample)

Attenuation of excitation radiation when penetrating into the sample

Sample composition and interaction probabilities

Attenuation of fluorescence radiation when leaving the sample

Solid angle of detection and detector sensitivity

Instrument sensitivity for the respective element

# Analysis of Steels and Alloys

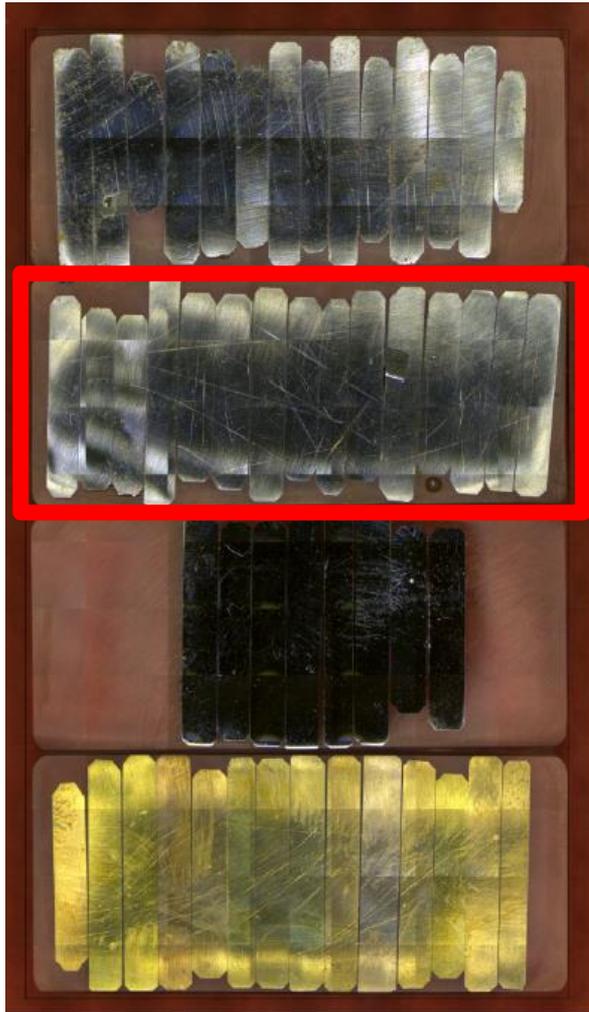
## Example



- It is important for the manufacturers of components using steels, alloys or other metals to know the composition of the material to be sure that the end-product can fulfil demanded specifications.
- The addition of heavy metals is essential for the mechanical and thermal properties of alloy steels.
- If it does not meet these demands, it might cause damage to the component or equipment.
- SEM-EDS analysis is challenging due to the
  - insufficient detection limit
  - line overlapping in the low energy range
  - specimen inhomogeneity
- Understanding steels / alloys:
  - Overall composition: Point Analysis
  - Inclusions: Mapping

# Analysis of Steels and Alloys

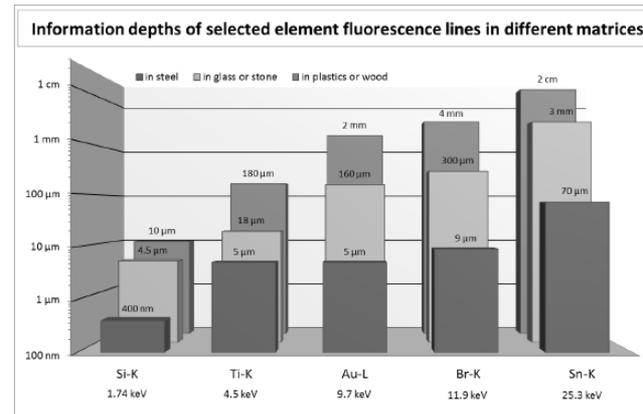
## Case Study



- SAMPLES: Set of 51 ARMI steels with known (certified) compositions
- Cast in Epoxy Resin and Polished
- High-Fe, **Cr-Ni steels**, Cu-alloys

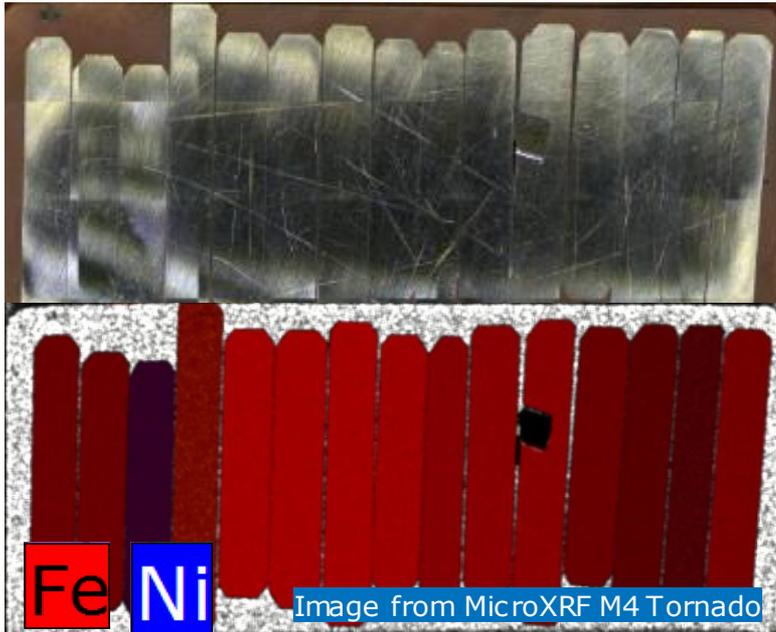
Steel samples have two main advantages:

- They are pretty homogeneous
- Even for the highest energy elements the information depth is restricted to below 100  $\mu\text{m}$  (up to Zn even  $< 5 \mu\text{m}$ )



→ Even small chips/flakes of steel can be quantified for classification

# Analysis of Steels and Alloys Samples



## ➤ Major Elements:

Cr, Fe, Mn, Ni

## ➤ Minor and Trace Elements:

C, N, Al, Si, P, S, Ti, V, Co, Cu, Nb, Mo, Sn, W

Element	Minimum (%)	Maximum (%)	Range (%)
C	0.02	1.02	1.00
N	0.01	0.33	0.32
Al	0.00	1.16	1.16
Si	0.27	1.38	1.11
P	0.01	0.04	0.03
S	0.00	0.29	0.29
Ti	0.00	0.63	0.63
V	0.02	0.26	0.24
Cr	<b>11.72</b>	<b>23.60</b>	<b>11.88</b>
Mn	0.35	9.31	8.96
Fe	<b>41.29</b>	<b>86.23</b>	<b>44.95</b>
Co	0.02	0.18	0.17
Ni	<b>0.11</b>	<b>35.84</b>	<b>35.73</b>
Cu	0.03	0.47	0.44
Nb	0.00	0.72	0.72
Mo	0.06	2.30	2.25
Sn	0.00	0.01	0.01
W	0.01	1.10	1.09
<b>Total</b>	<b>100.00</b>	<b>100.00</b>	<b>0.00</b>

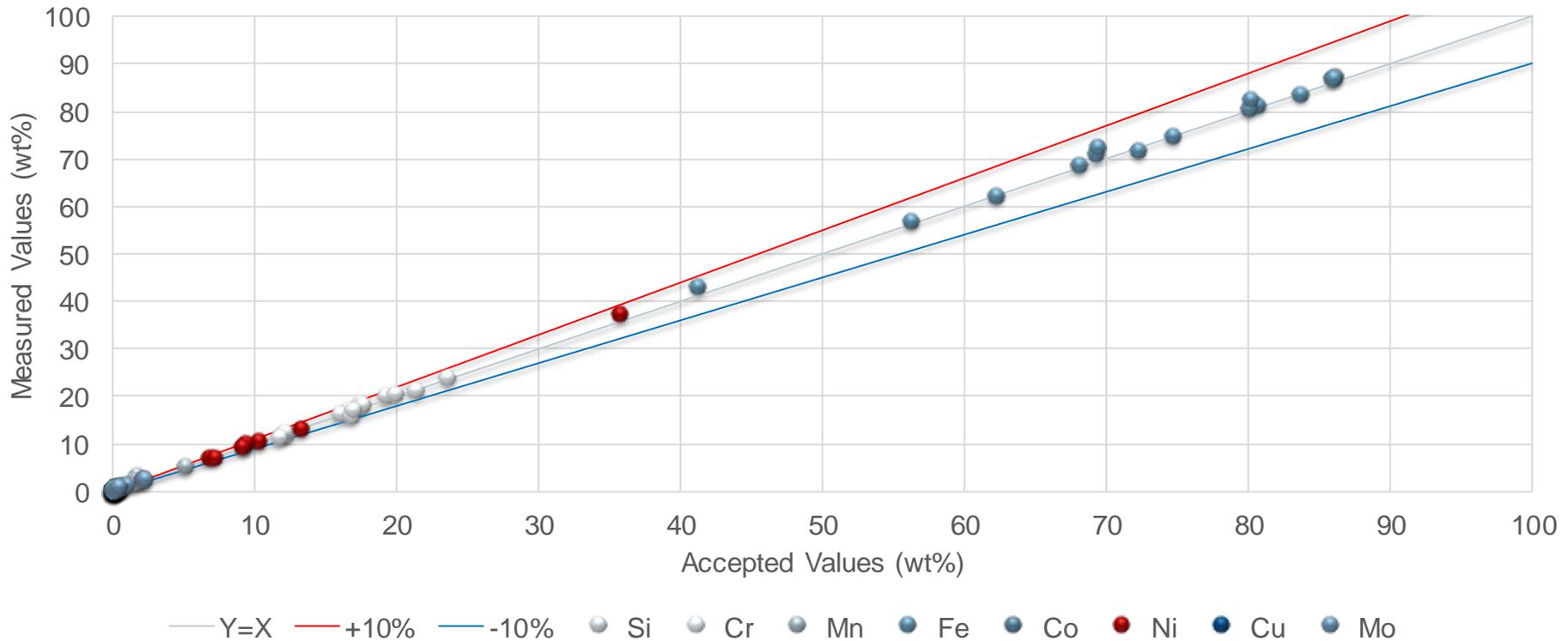
# Analysis of Steels and Alloys

## Excitation: Micro-XRF; Detector: EDS



Analytical Conditions Point Analysis: 50 kV, 600 uA, No Filter, 130 kcps, under vacuum, Working Distance 12 mm, 120 seconds

### All (wt%) Reconciliation

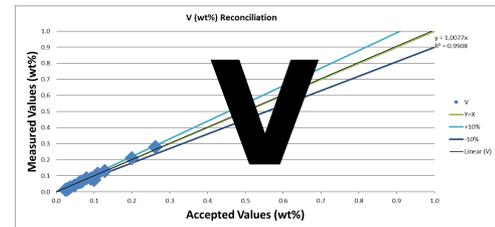
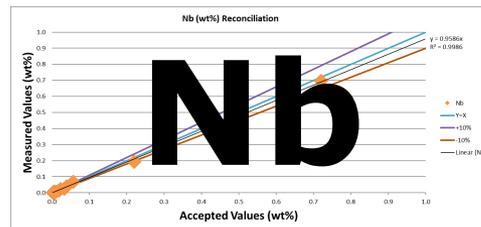
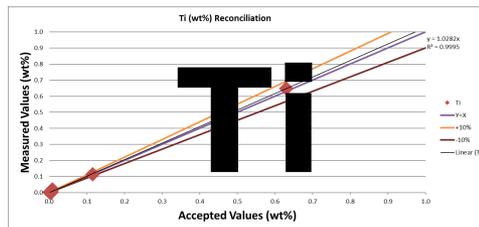
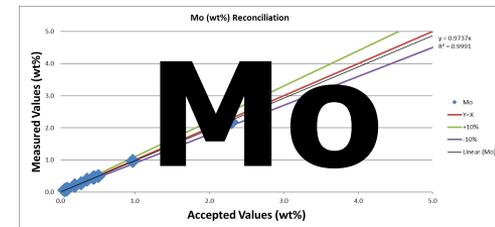
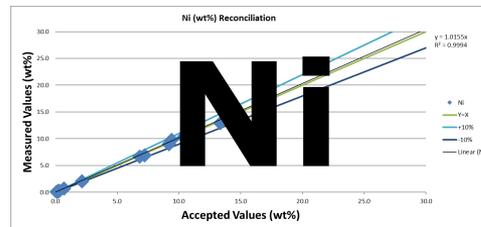
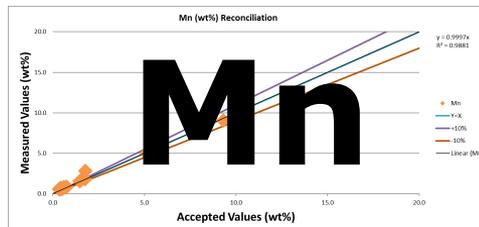
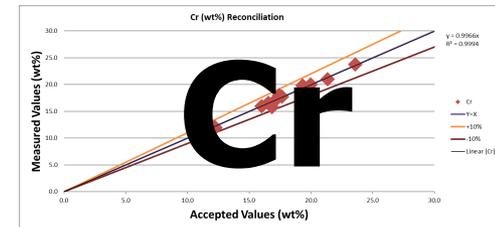
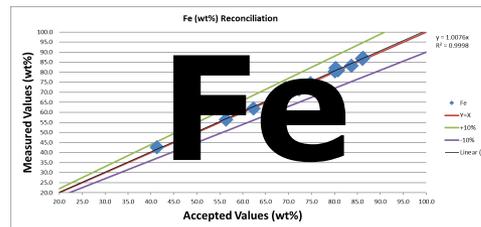
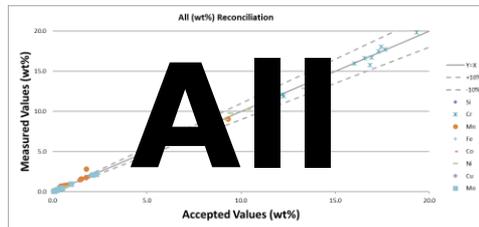


# Analysis of Steels and Alloys

## Individual Elements: Ti (Minor / Trace)

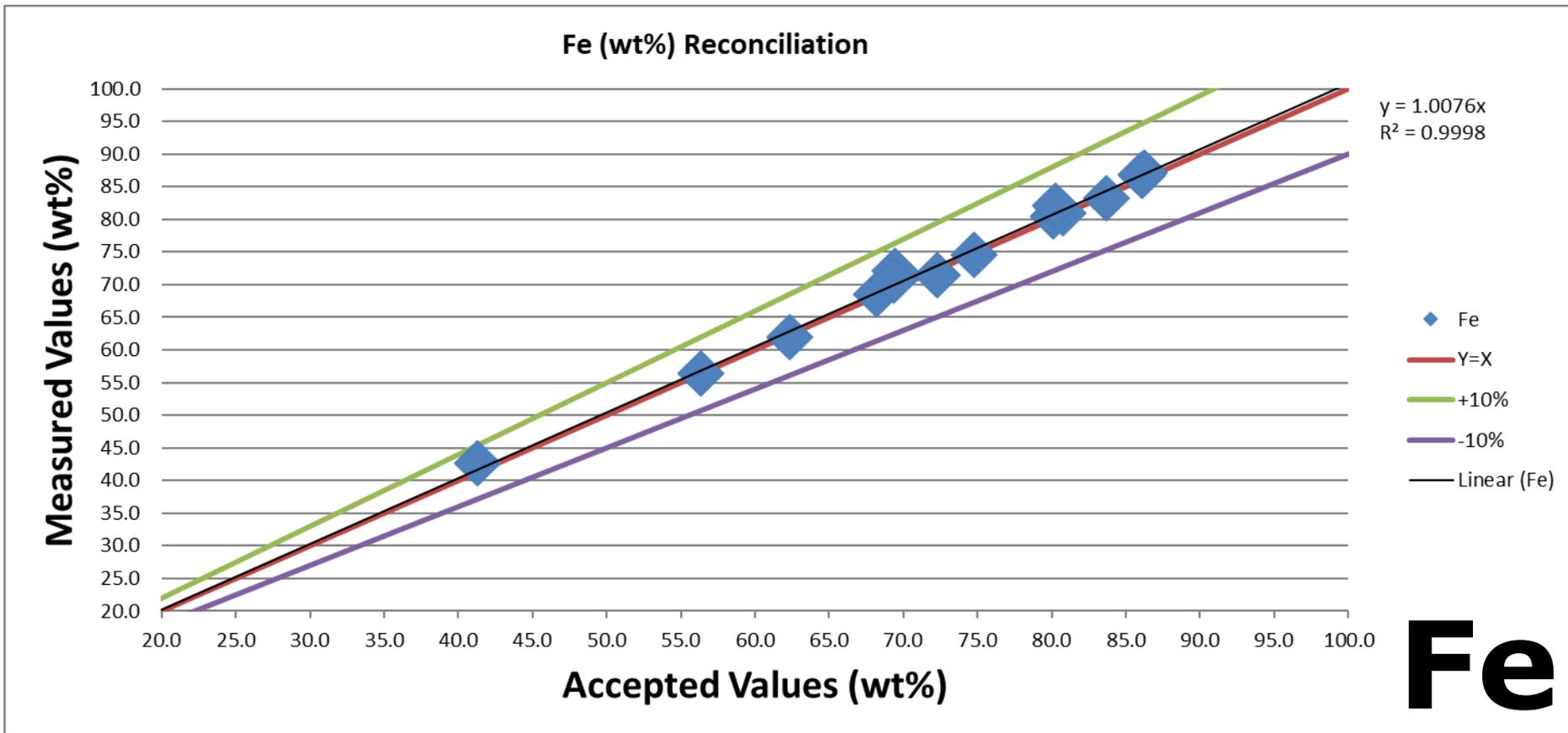


**Micro-XRF Excitation. EDS Detector. SEM-XRF-EDS.**



# Analysis of Steels and Alloys

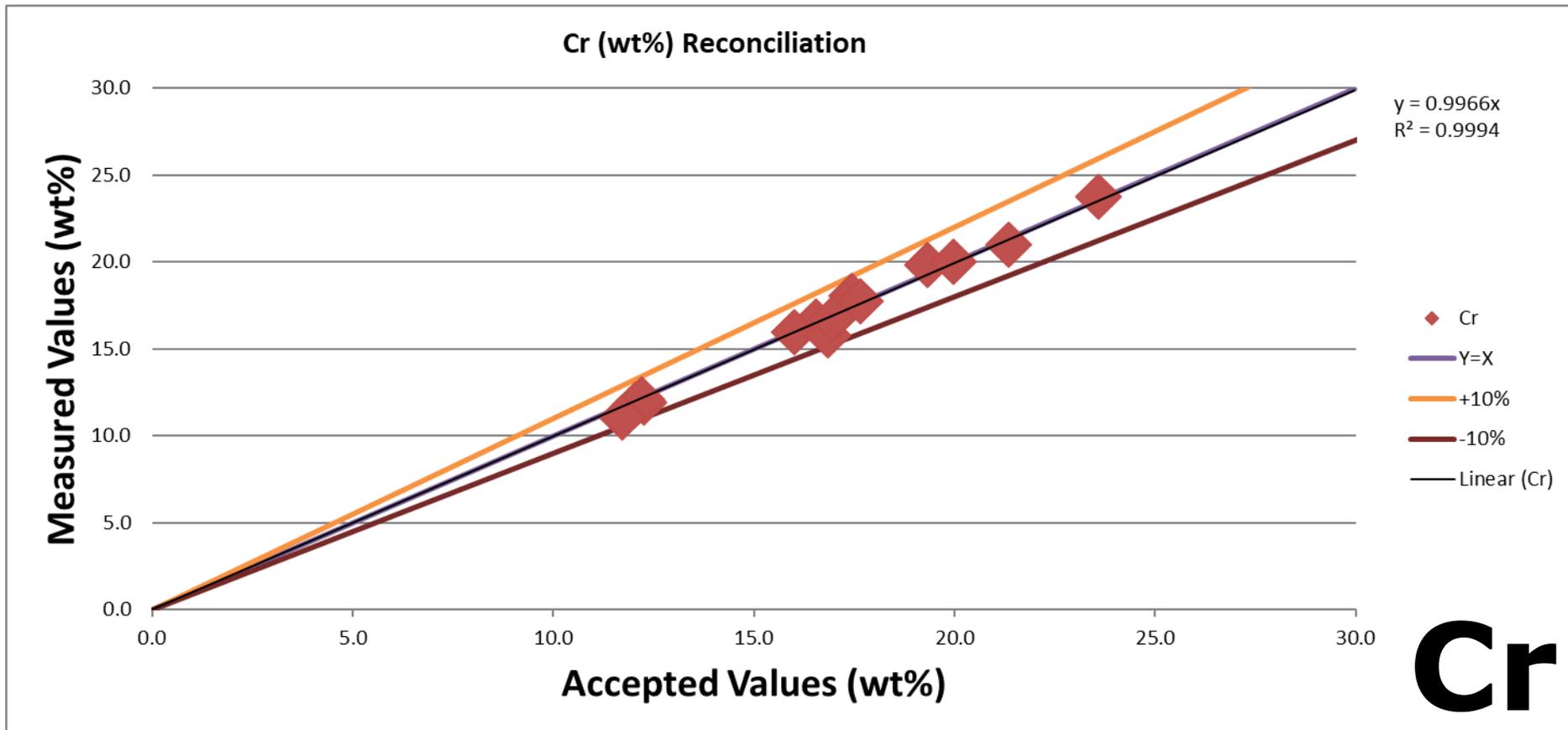
## Individual Elements: Fe (Major)



**Micro-XRF Excitation. EDS Detector. SEM-XRF-EDS.**

# Analysis of Steels and Alloys

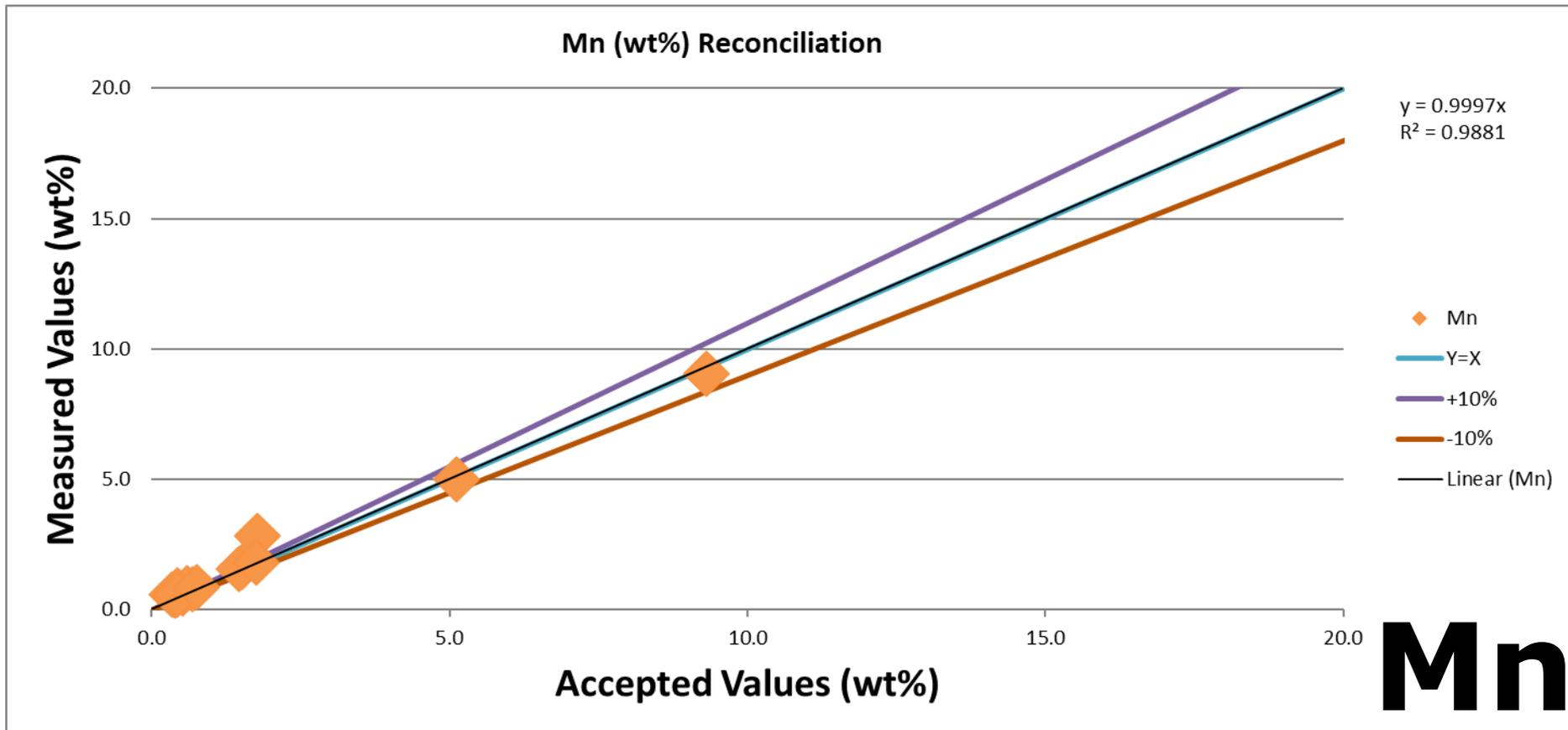
## Individual Elements: Cr (Major)



**Micro-XRF Excitation. EDS Detector. SEM-XRF-EDS.**

# Analysis of Steels and Alloys

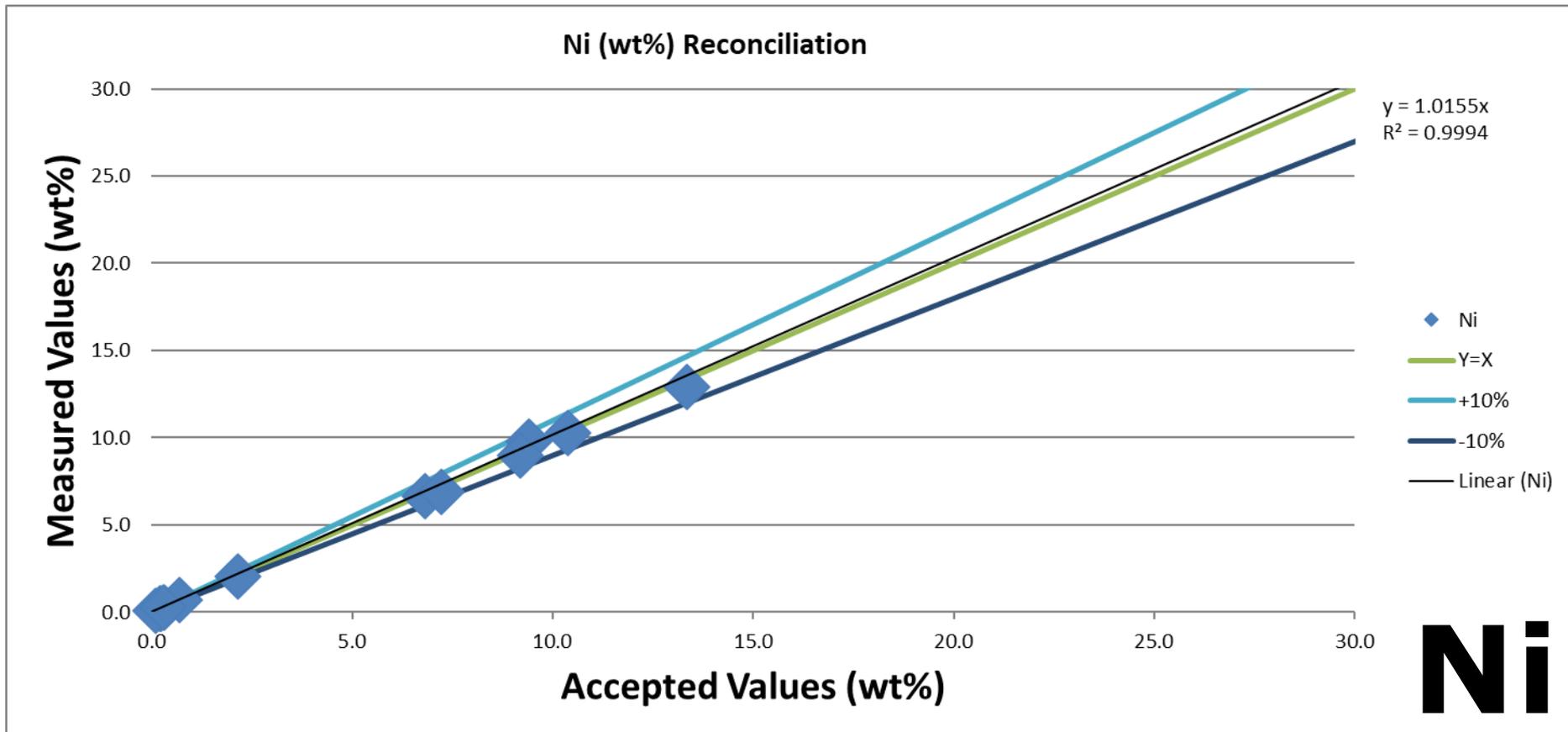
## Individual Elements: Mn (Major / Minor)



**Micro-XRF Excitation. EDS Detector. SEM-XRF-EDS.**

# Analysis of Steels and Alloys

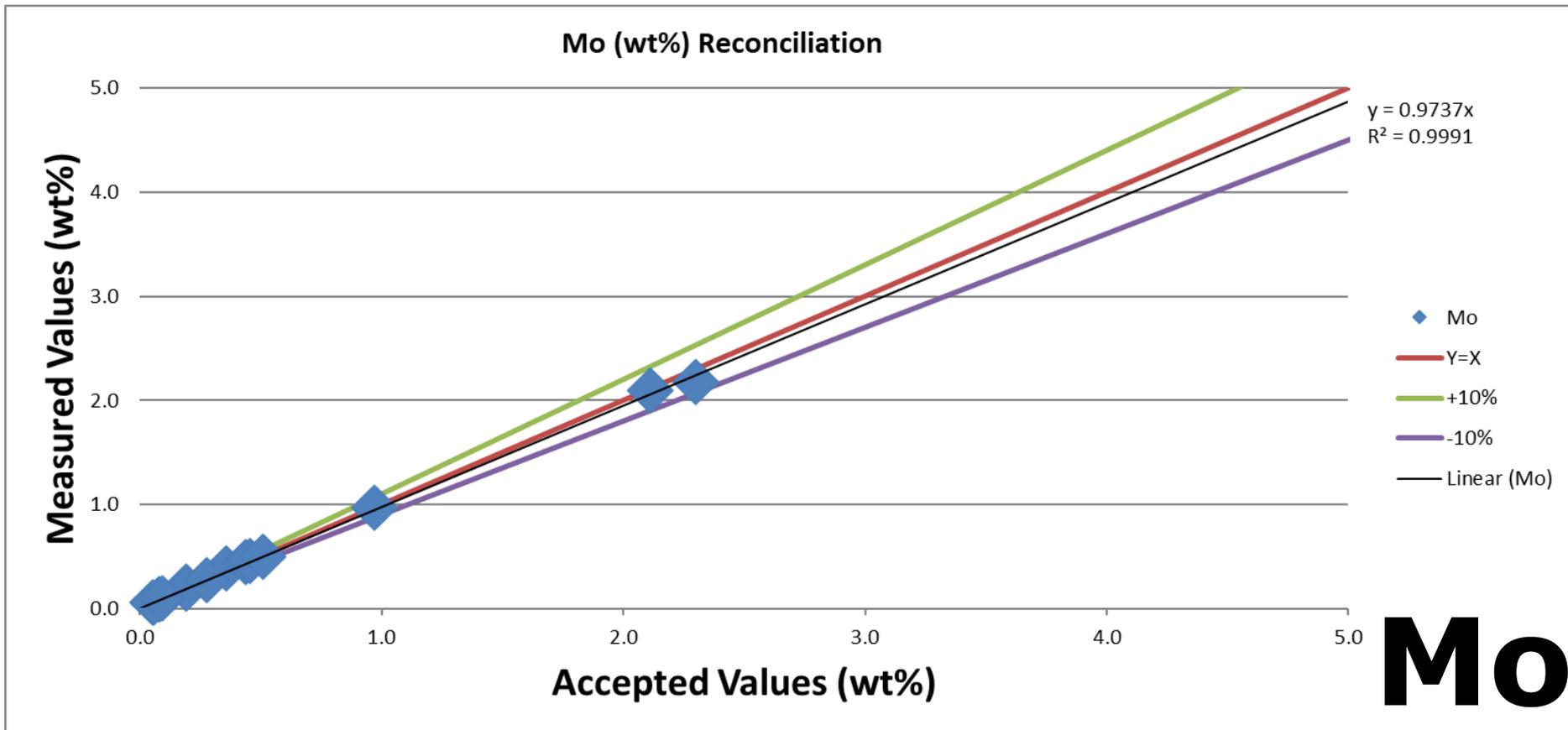
## Individual Elements: Ni (Major / Minor / Trace)



**Micro-XRF Excitation. EDS Detector. SEM-XRF-EDS.**

# Analysis of Steels and Alloys

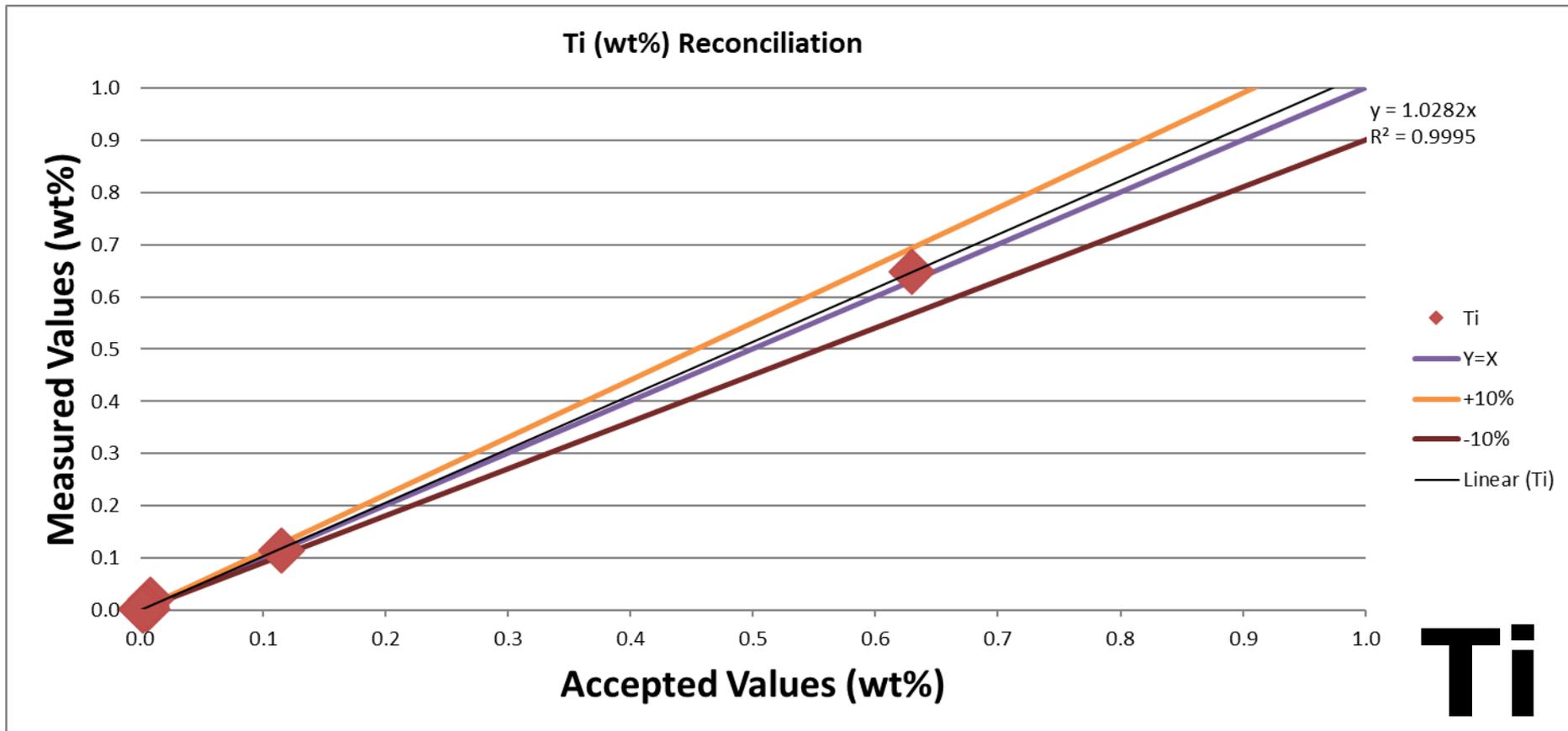
## Individual Elements: Mo (Minor / Trace)



**Micro-XRF Excitation. EDS Detector. SEM-XRF-EDS.**

# Analysis of Steels and Alloys

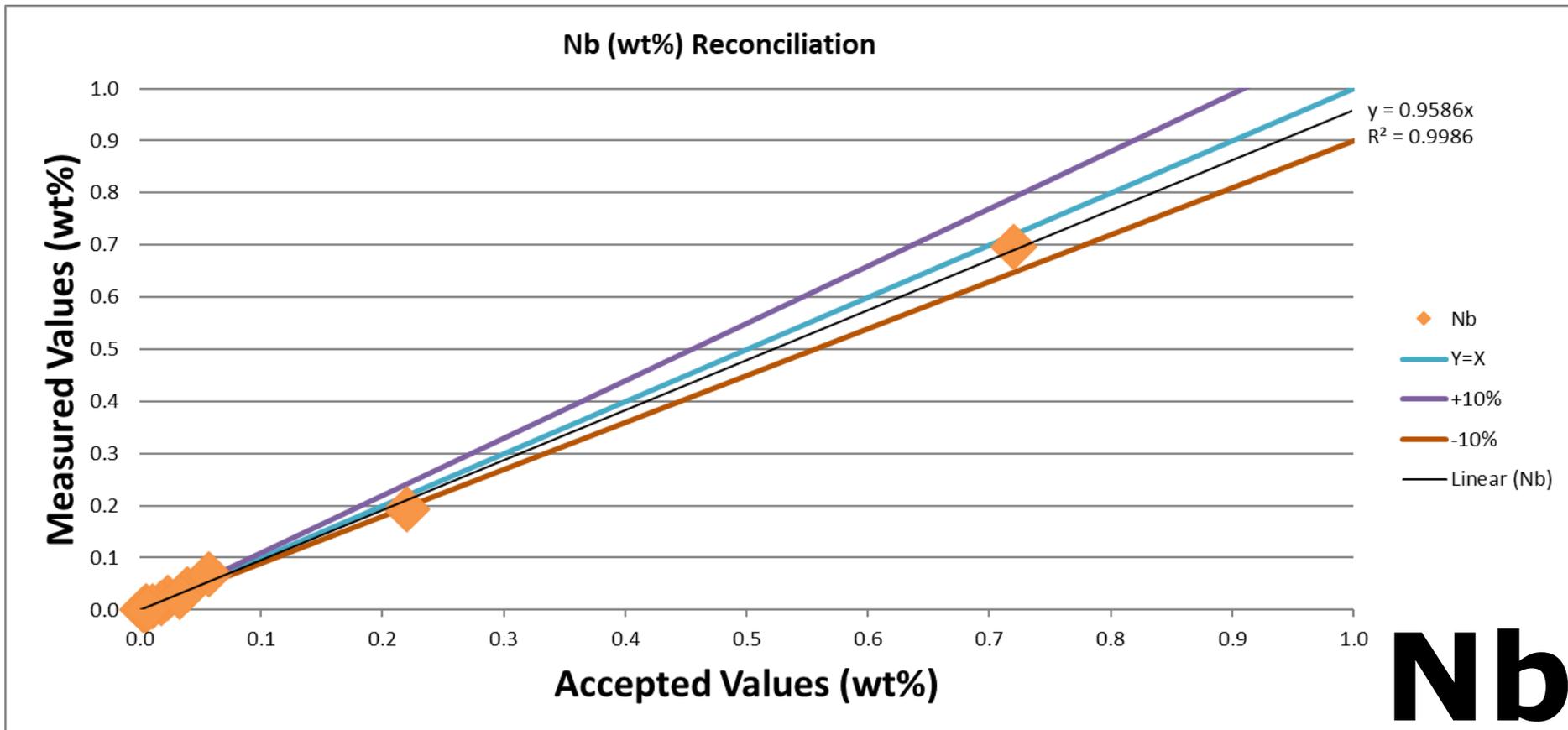
## Individual Elements: Ti (Minor / Trace)



**Micro-XRF Excitation. EDS Detector. SEM-XRF-EDS.**

# Analysis of Steels and Alloys

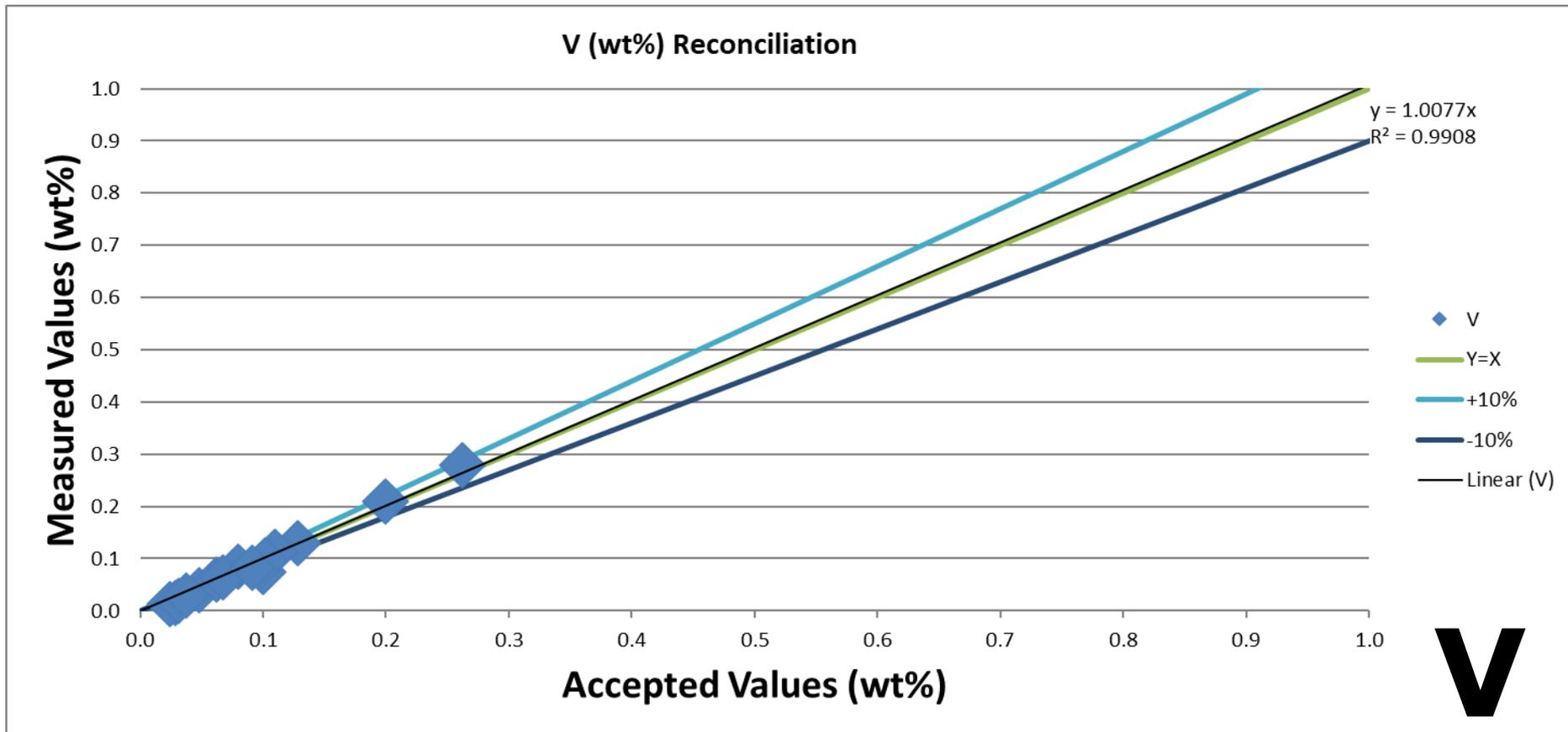
## Individual Elements: Nb (Minor / Trace)



**Micro-XRF Excitation. EDS Detector. SEM-XRF-EDS.**

# Analysis of Steels and Alloys

## Individual Elements: V (Minor / Trace)



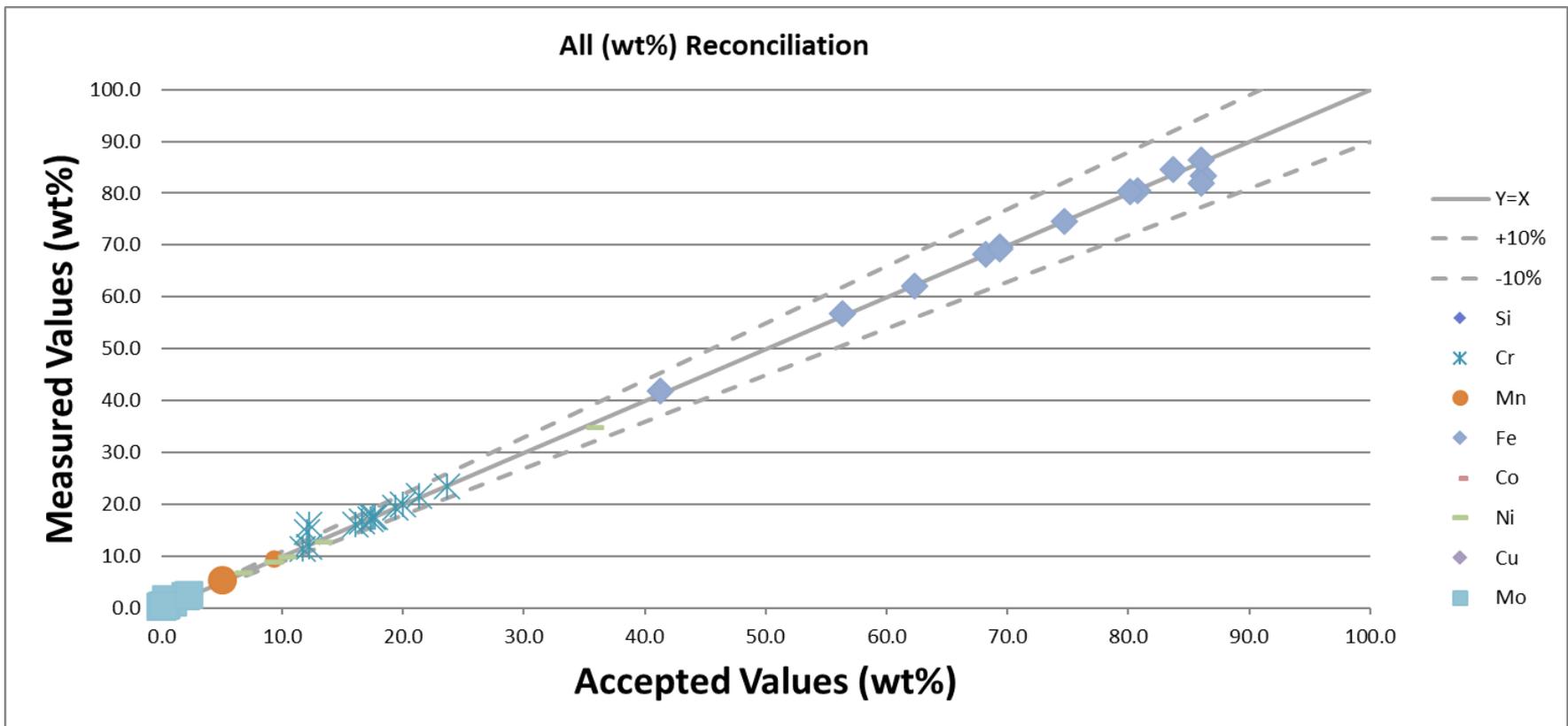
**Micro-XRF Excitation. EDS Detector. SEM-XRF-EDS.**

# Analysis of Steels and Alloys

## Excitation: Electron; Detector: EDS



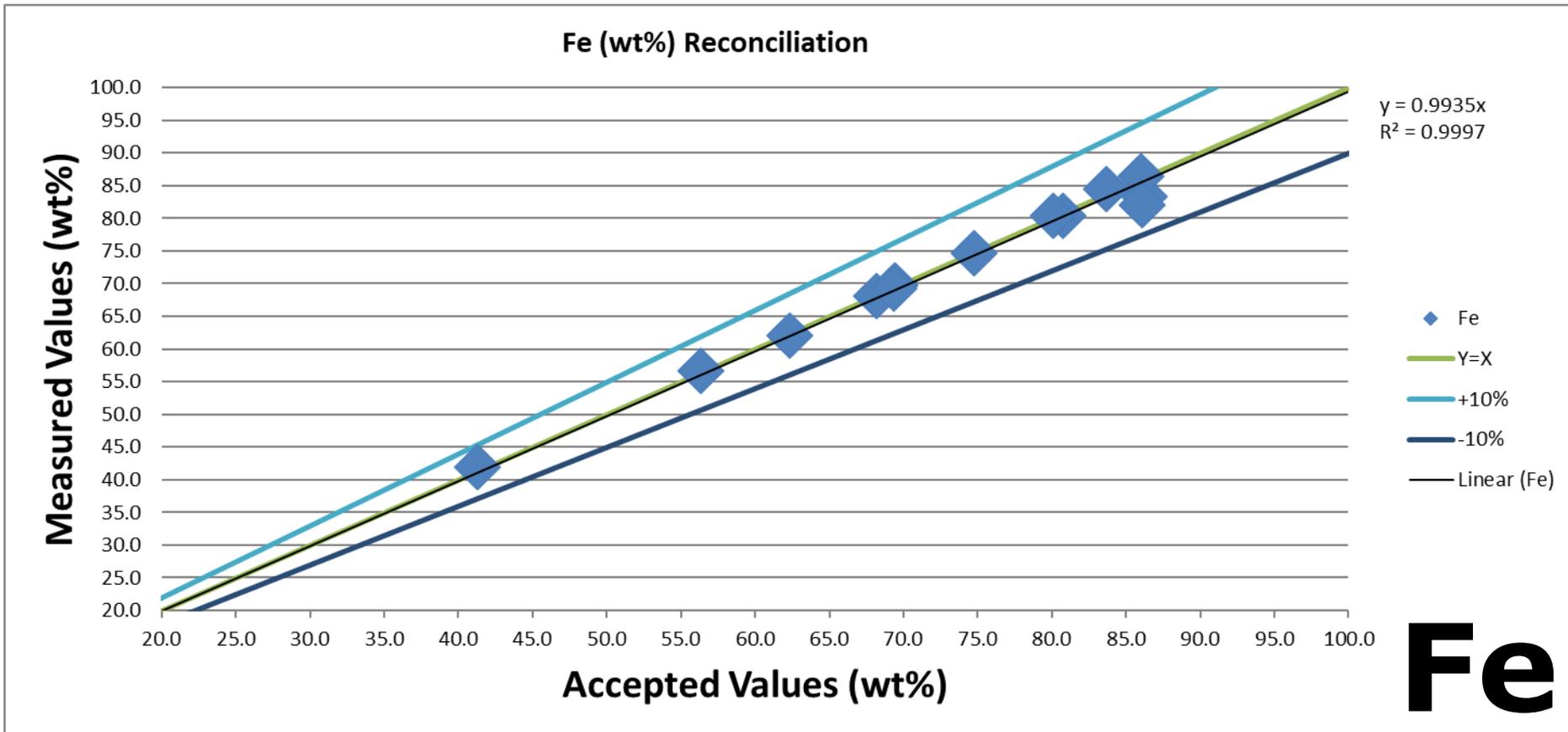
Analytical Conditions Point Analysis: 20 kV, 10 mA, 275 kcps, under vacuum, Working Distance 12 mm, 120 seconds



**Electron Excitation. EDS Detector. SEM-EDS.**

# Analysis of Steels and Alloys

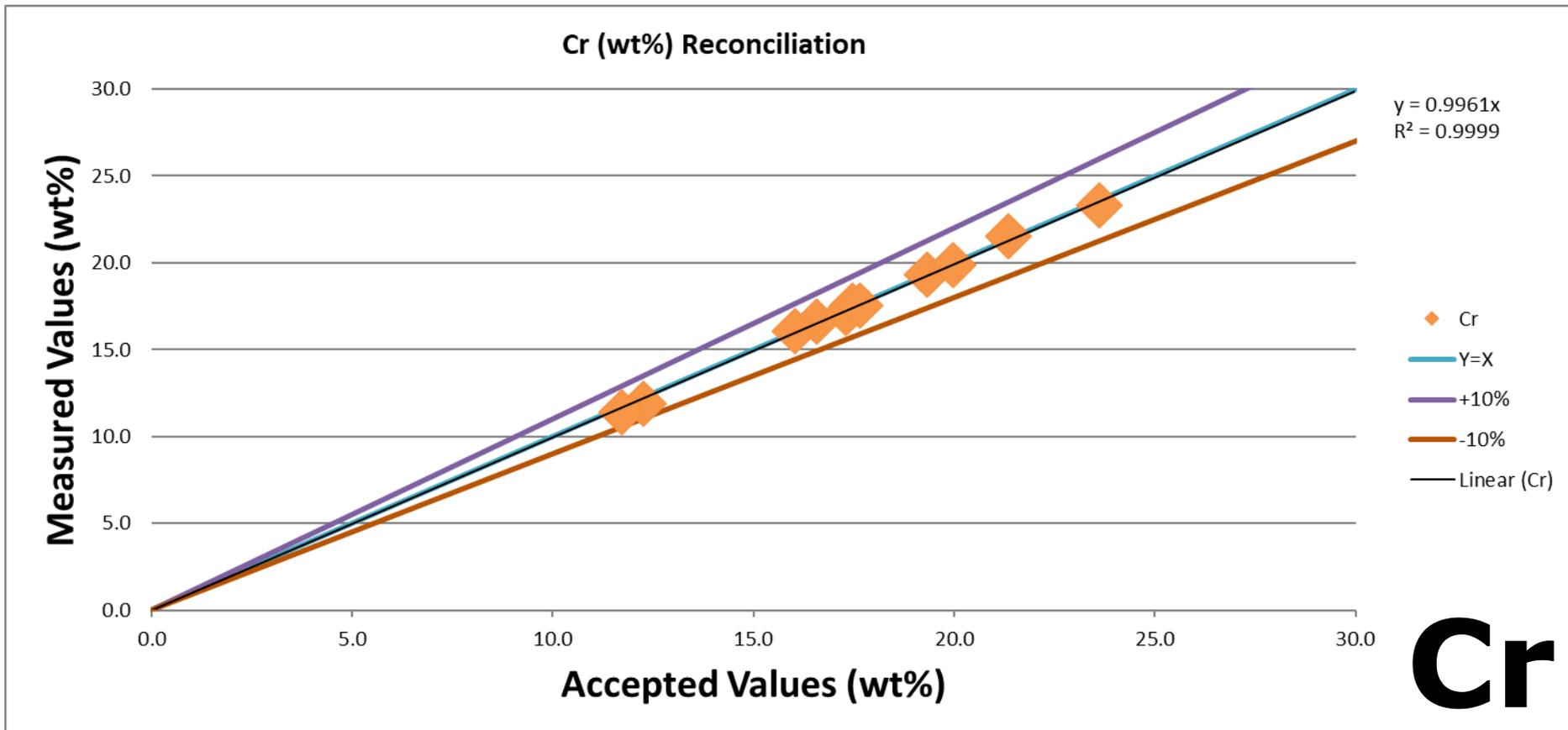
## Individual Elements: Fe (Major)



Electron Excitation. EDS Detector. SEM-EDS.

# Analysis of Steels and Alloys

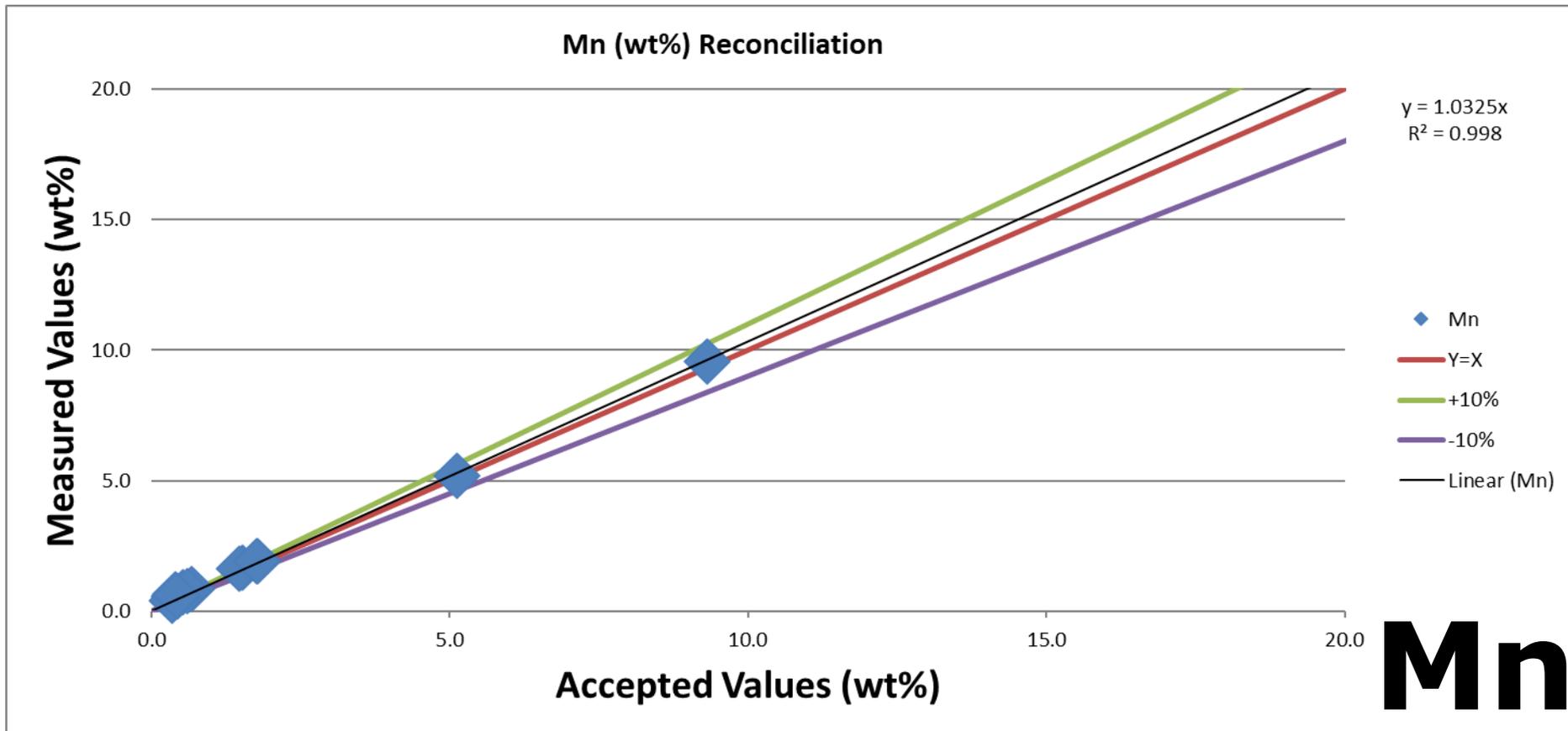
## Individual Elements: Cr (Major)



**Electron Excitation. EDS Detector. SEM-EDS.**

# Analysis of Steels and Alloys

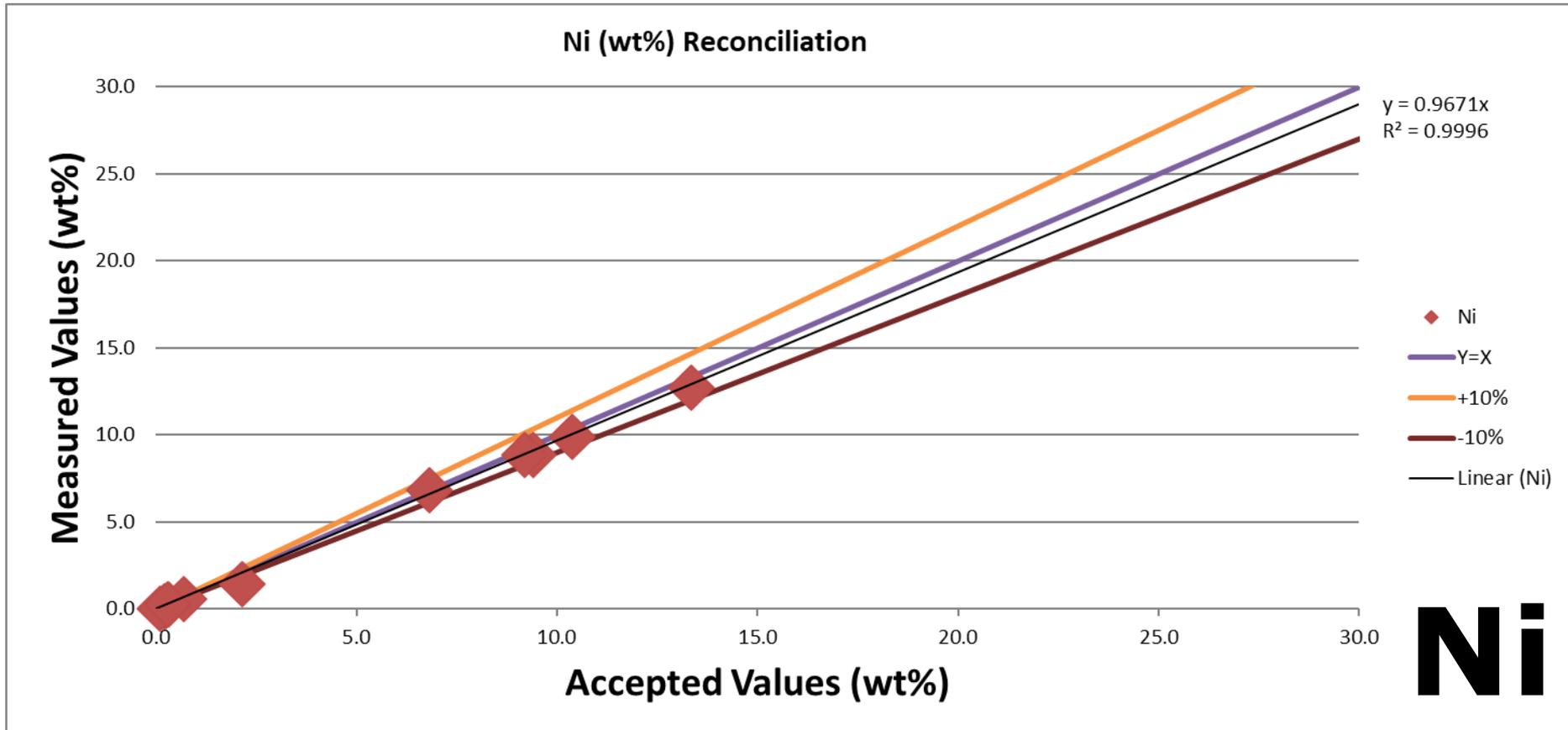
## Individual Elements: Mn (Major / Minor)



Electron Excitation. EDS Detector. SEM-EDS.

# Analysis of Steels and Alloys

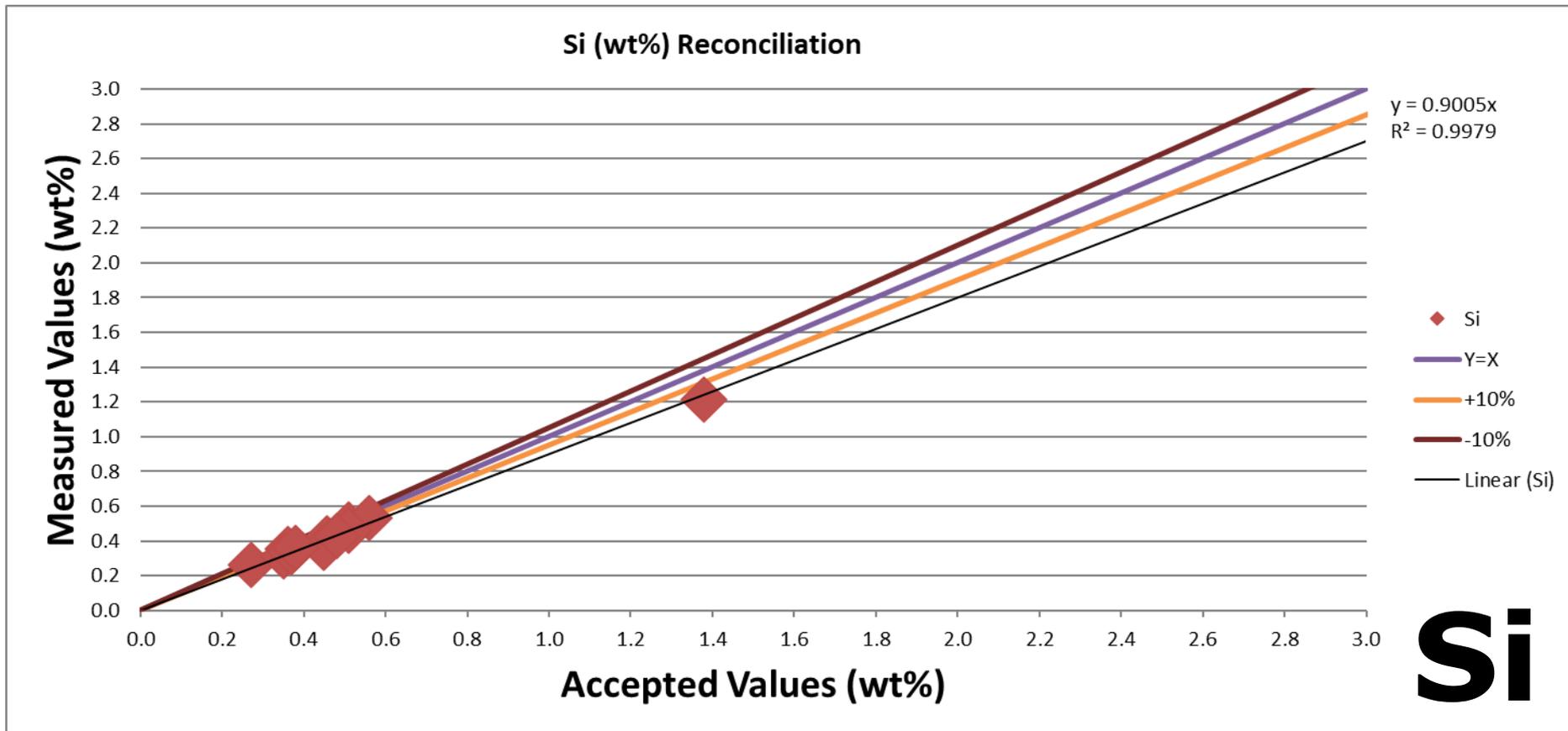
## Individual Elements: Ni (Major / Minor)



**Electron Excitation. EDS Detector. SEM-EDS.**

# Analysis of Steels and Alloys

## Individual Elements: Si (Minor)



Electron Excitation. EDS Detector. SEM-EDS.

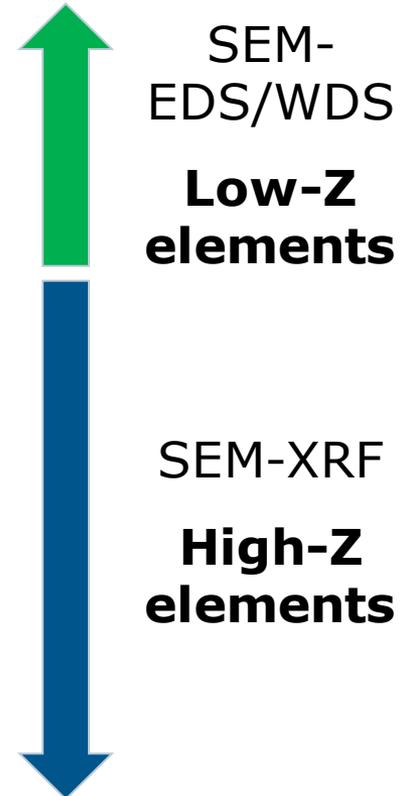
# Analysis of Steels and Alloys

## Combined Analysis



Sample 32: AISI 422-205B

Element	Certified	MicroXRF	SEM-EDS	Combined
C	0.22			
N	0.05			
Al	0.01			
Si	0.37		0.34	0.33
P	0.01			
S	0.00			
Ti	0.00	0.003		0.00
V	0.26	0.279		0.26
Cr	11.72	11.084	11.37	11.32
Mn	0.68	0.797	0.87	0.75
Fe	83.70	83.243	84.55	83.20
Co	0.03	0.024	0.49	0.02
Ni	0.70	0.692	0.54	0.67
Cu	0.15	0.177		0.15
Nb	0.02	0.012		0.01
Mo	0.97	0.970	0.95	0.94



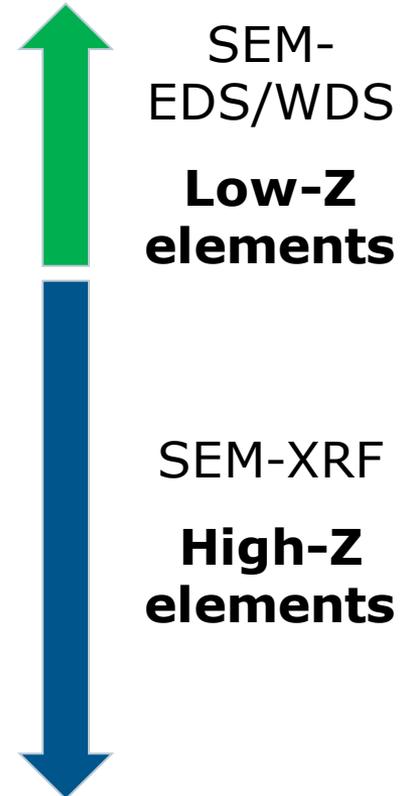
# Analysis of Steels and Alloys

## Combined Analysis



Sample 28: AISI 347-8D

Element	Certified	MicroXRF	SEM-EDS	Combined
C	0.05			
N	0.02			
Al	0.00			
Si	0.36		0.36	0.35
P	0.03			
S	0.03			
Ti	0.00	0.002		0.002
V	0.06	0.059		0.058
Cr	17.30	17.496	17.09	17.41
Mn	1.76	1.788	1.92	1.74
Fe	69.33	70.713	69.33	70.11
Co	0.14	0.154	0.60	0.14
Ni	9.19	9.012	8.86	9.00
Cu	0.47	0.501	0.50	0.49
Nb	0.72	0.695	0.74	0.67
Mo	0.44	0.448	0.44	0.42

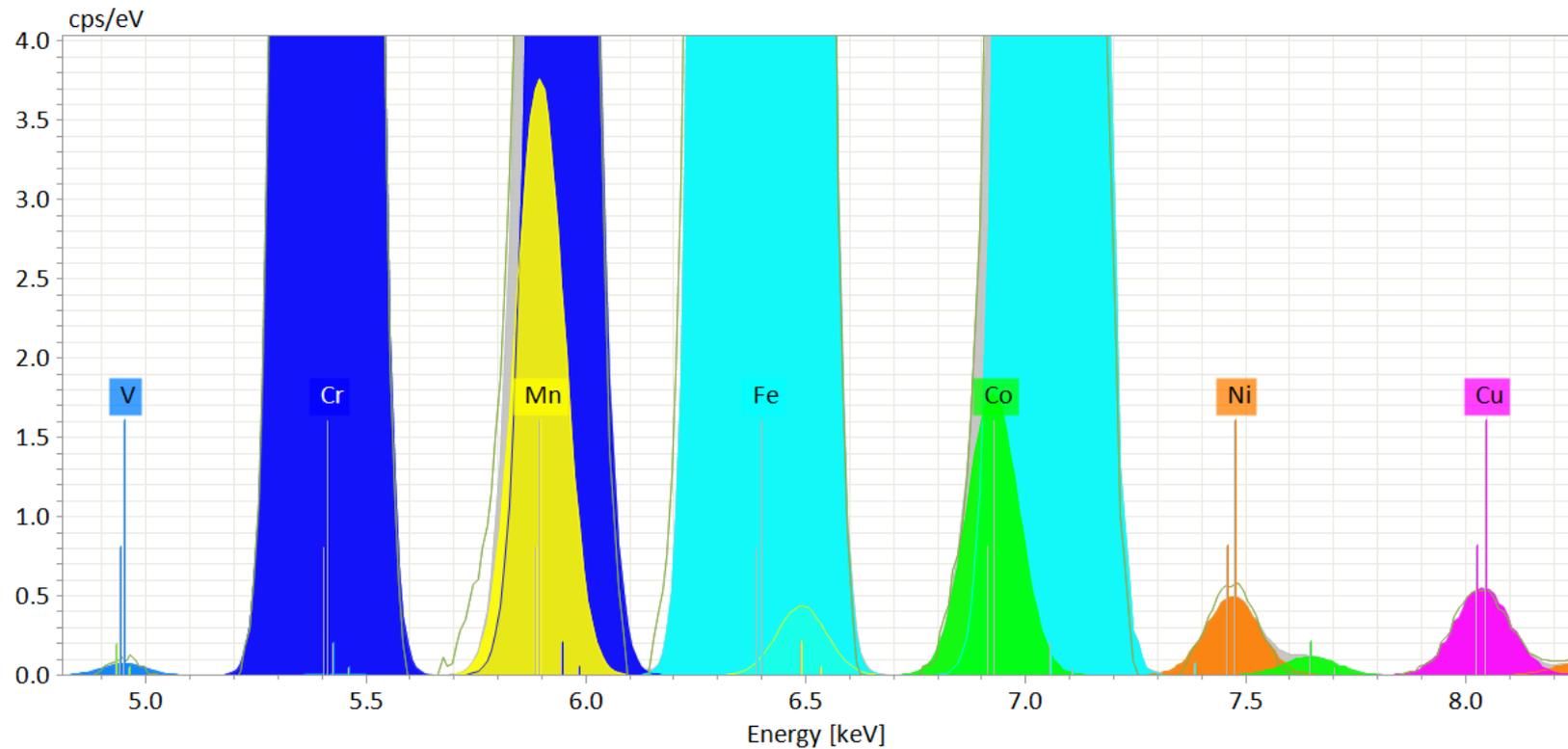


# Analysis of Steels and Alloys

## Deconvolution



### ➤ Deconvolution, Peak Overlaps, and Background

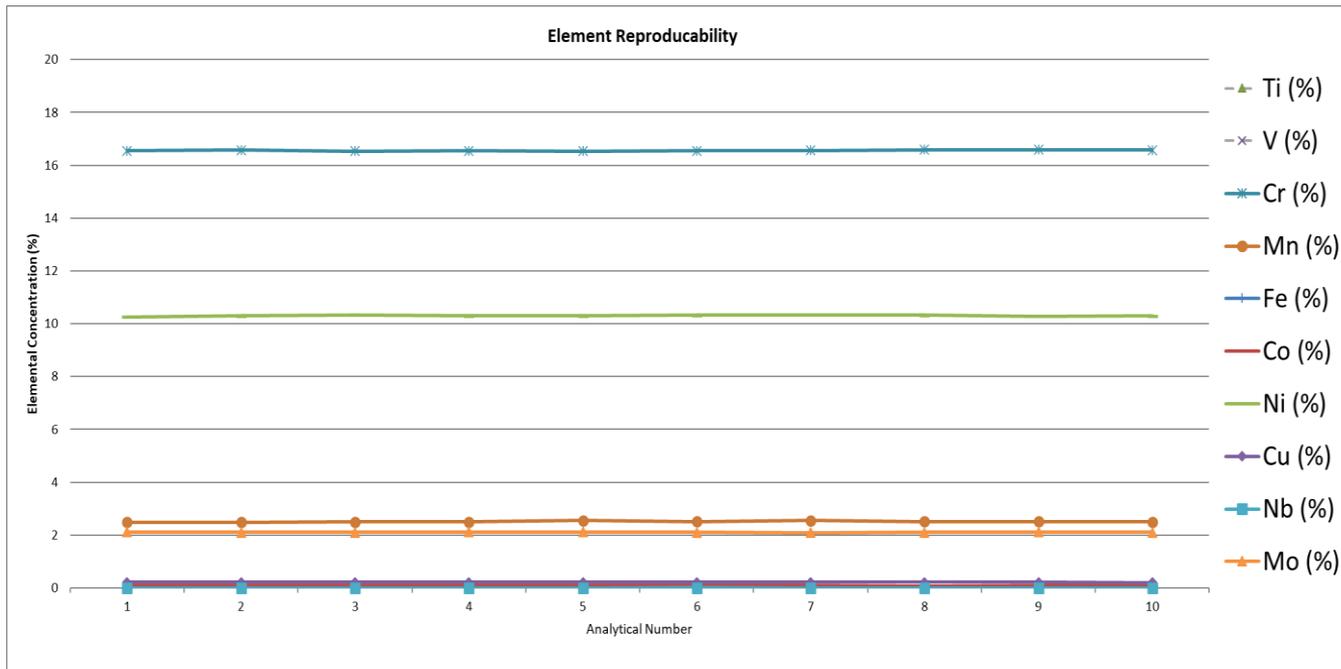


# Analysis of Steels and Alloys

## Repeat Analysis: Same Spot



- Reproducible analytical results when analysing the same spot
- Small Standard Deviation



Element	Accepted Values	Average
Ti	0.01	0.01
V	0.03	0.01
Cr	16.56	16.59
Mn	1.78	1.85
Fe	68.19	69.00
Co	0.14	0.09
Ni	10.38	10.18
Cu	0.17	0.21
Nb	0.00	0.00
Mo	2.11	2.10

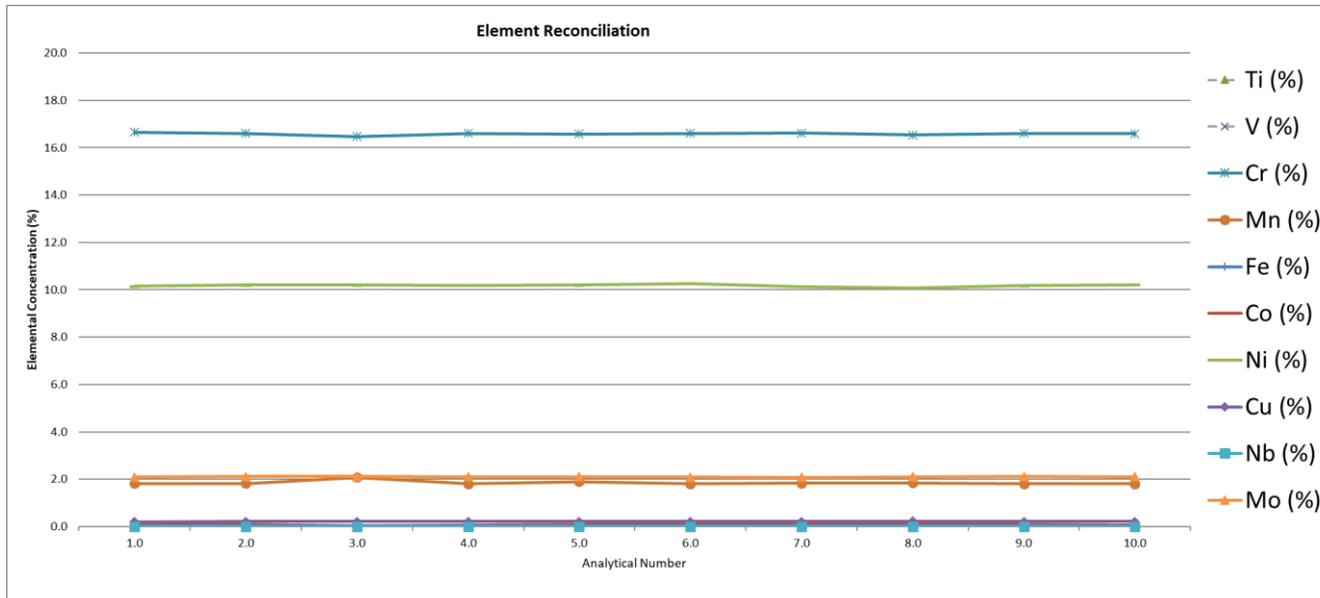
- Sample 25: AISI 316-5D

# Analysis of Steels and Alloys

## Repeat Analysis: Same Sample



- Reproducible analytical results when analysing the same sample in different positions
- Small Standard Deviation, indicates homogeneous sample



Element	Accepted Values	Average
Ti	0.01	0.01
V	<b>0.03</b>	<b>0.01</b>
Cr	16.56	16.56
Mn	<b>1.78</b>	<b>2.51</b>
Fe	68.19	68.68
Co	<b>0.14</b>	<b>0.08</b>
Ni	10.38	10.31
Cu	<b>0.17</b>	<b>0.21</b>
Nb	0.00	0.00
Mo	<b>2.11</b>	<b>2.11</b>

- Sample 25: AISI 316-5D

# Analysis of Steels and Alloys

## Repeat Analysis: Same Sample



- Reproducible analytical results when analysing the same spot
- Small Standard Deviation

	Ti (%)	V (%)	Cr (%)	Mn (%)	Fe (%)	Co (%)	Ni (%)	Cu (%)	Nb (%)	Mo (%)
1.spx	0.01	0.01	16.55	2.48	68.59	0.09	10.26	0.21	0.00	2.12
2.spx	0.01	0.02	16.57	2.48	68.70	0.09	10.30	0.21	0.00	2.11
3.spx	0.01	0.01	16.53	2.50	68.72	0.08	10.34	0.22	0.00	2.11
4.spx	0.01	0.02	16.55	2.50	68.67	0.08	10.30	0.21	0.00	2.12
5.spx	0.01	0.01	16.53	2.55	68.64	0.08	10.30	0.22	0.00	2.12
6.spx	0.01	0.01	16.55	2.52	68.67	0.08	10.32	0.22	0.00	2.11
7.spx	0.01	0.02	16.56	2.55	68.68	0.09	10.34	0.21	0.00	2.09
8.spx	0.01	0.01	16.59	2.51	68.70	0.08	10.32	0.22	0.00	2.11
9.spx	0.01	0.02	16.59	2.51	68.73	0.09	10.28	0.21	0.00	2.12
10.spx	0.01	0.01	16.57	2.49	68.70	0.08	10.29	0.21	0.00	2.11

# Analysis of Steels and Alloys

## Considerations



- As with all microanalysis, it is important to understand the analytical goal prior to analysis. This often requires an understanding of the sample.
- Steels are often homogenous on a micrometer scale.
- Thus the difference in analytical volume between electron and X-ray excitation will not influence the quantification and interpretation of results.

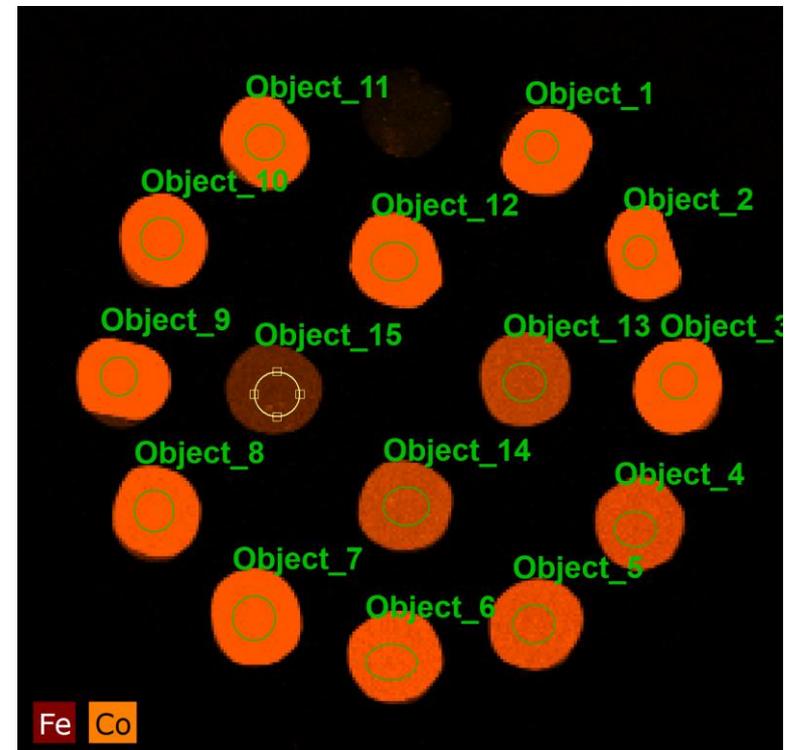
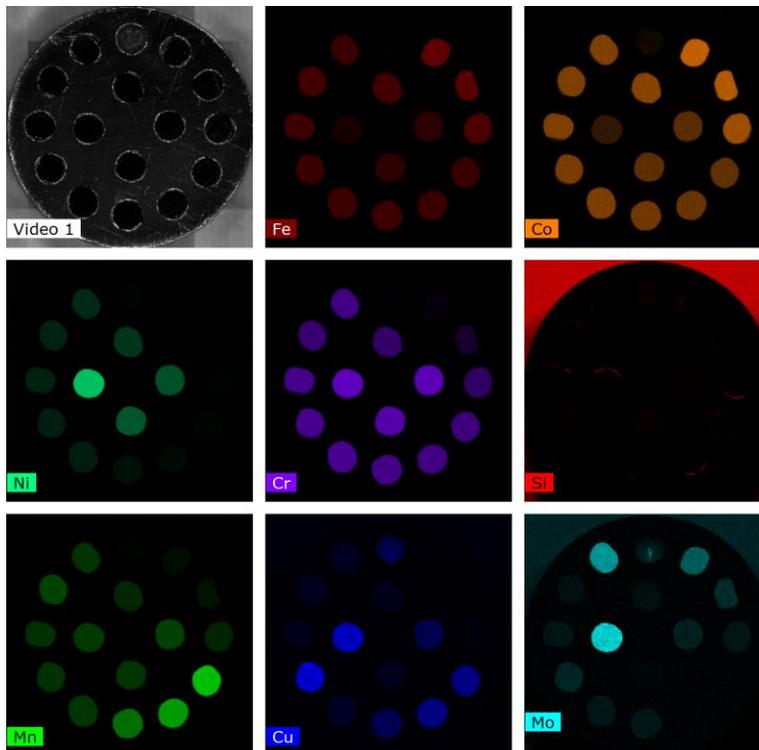
However, sometimes, steels are segregated and may contain small impurities. If so, then there could be a difference between the two techniques due to the excitation volume. That is the higher resolution of electron excitation and the greater penetration depth of the X-ray excitation may generate a combined result that is inconsistent for interpretation.

To resolve this, an area map could be done.

# Large Area Maps: Micro-XRF and Electron excitation



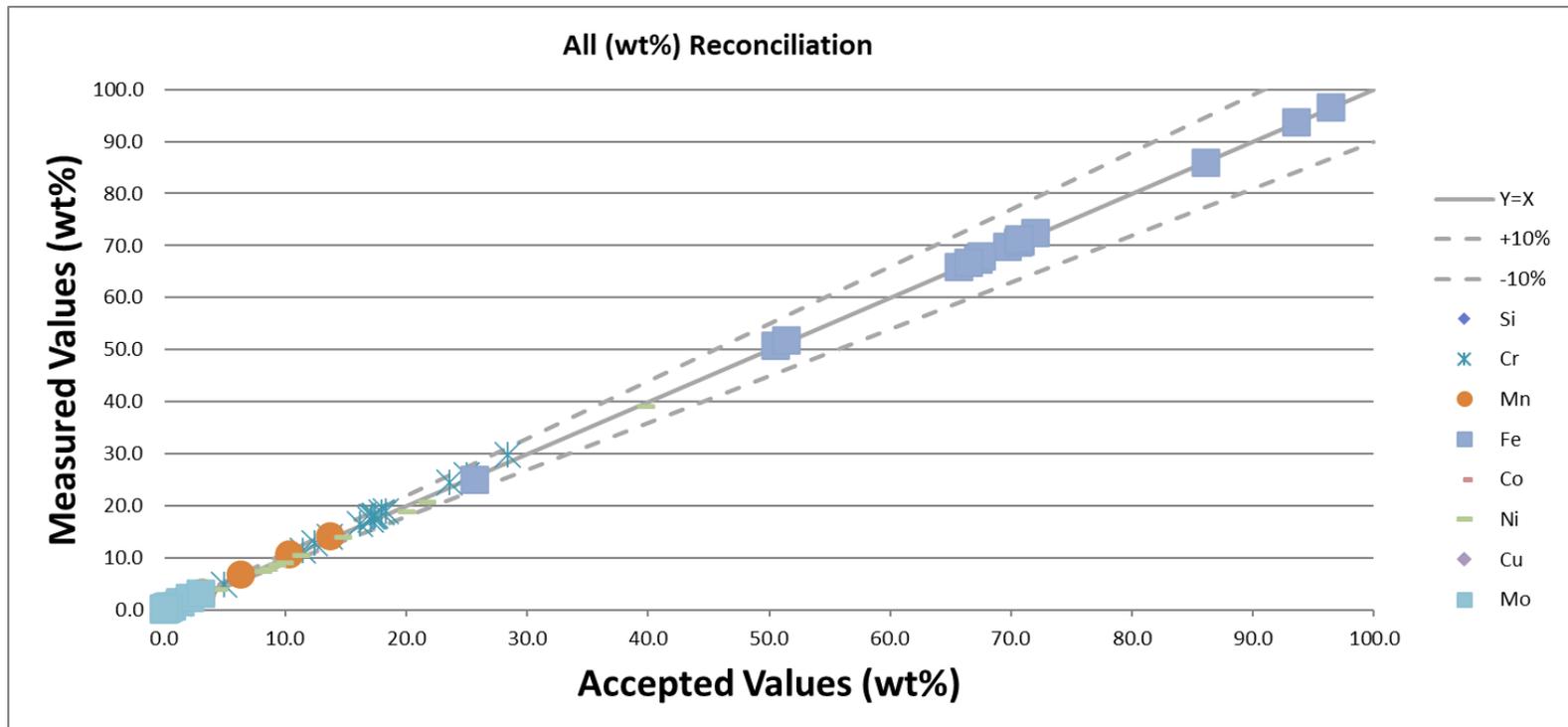
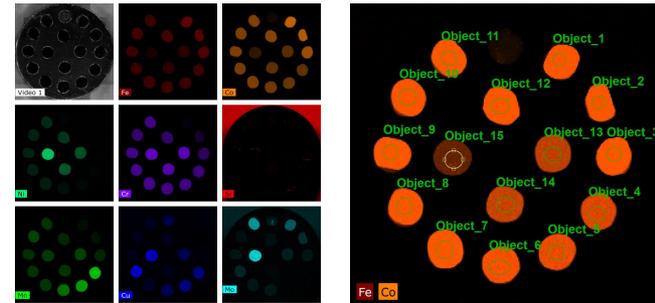
- Both micro-XRF and electron excitation are capable of both point analysis and mapping.
- Difference in Beam Size and Area of excitation



# Large Area Maps: Micro-XRF and Electron excitation



Quantification of selected areas  
from within a mapped area



# Summary and Conclusions: Analysis of Steels and Alloys



- X-ray and Electron Excitation can work in combination to provide improved quantitative results, using the benefits of each. Specifically:
  - Electron excitation is preferable for light elements, e.g. C to Si
  - X-ray excitation is preferable for heavy elements and trace concentrations
  
- Samples can be analysed and quantified either as spot (point) analyses or from hypermaps.
  
- Benefits of each analytical method can be utilised. For example:
  - Micro-XRF: Sample Preparation is minimal for micro-XRF
    - No carbon-coating, No polishing
  - Electron Beam: High resolution for detecting small inclusions

**For more information, please contact us:**

**Bruker Nano GmbH**

**[info.bna@bruker.com](mailto:info.bna@bruker.com)**

**and**

**If you want to learn more about practical micro-XRF analysis including sample, measurement setup, and evaluation, our latest video series is available via the Bruker website and youtube**

## Product Videos



Part I - Introduction to micro-XRF and the Rapid Stage on a SEM



Part II - Loading a sample and performing a measurement



Part III - Analyzing a measured dataset

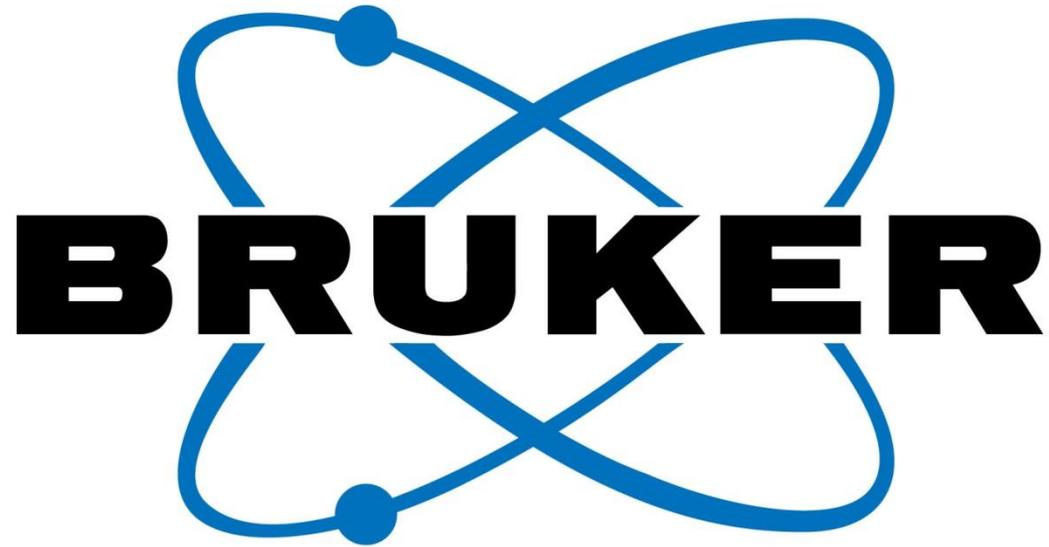
## Rapid Stage



The Rapid Stage can be mounted on top of the SEM stage for fast mapping over large sample areas.

## Are There Any Questions?

Please type in the questions you might have  
in the Q&A box and press *Send*.



Innovation with Integrity